

Introduction to matter – radiation interaction

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Content

- Macroscopic
 - Dielectric response of matter
 - Index of refraction, dielectric function
 - Attenuation and dispersion
 - Model dielectric functions
- Microscopic
 - Phenomenology, from IR to x-rays
 - Cross section and attenuation coefficient
 - Scattering processes and the atomic form factor
 - Photoelectric absorption
- Relation between the atomic form factor and index of refraction
- Interaction between radiation and hydrogen – like atoms, semi-classical
 - Photoelectric absorption cross - section
 - Scattering cross – section
- X-ray scattering from a single electron, an atom and a solid



References

- This presentation
- J.D. Jackson, *Classical Electrodynamics*, Wiley, 3rd Edition, 7.1, 7.2, 7.5 (parts)
- J. Als – Nielsen and D. McMorrow, *Introduction to Modern X-ray Physics*, Wiley, New York, 2001, Chap. 1
- D. Attwood and A. Sakdinawat, *X-rays and extreme ultraviolet radiation*, 2nd edition, Cambridge University Press (2017)



Dispersion and attenuation

- The dielectric function determines dispersion and attenuation of an EM wave propagating in matter
- In vacuum the dispersion relation for EM waves is

$$\omega = ck$$

c is the speed of light *in vacuo*.



Dispersion and attenuation

- In matter the dispersion relation is modified by the presence of the index of refraction $n(\omega)$:

$$\omega = \frac{c}{n(\omega)}k$$

- The index of refraction is

$$n(\omega) = \sqrt{\varepsilon(\omega)\mu(\omega)}$$

- Neglecting magnetic effects, $\mu(\omega) = 1$

$$n(\omega) = \sqrt{\varepsilon(\omega)}$$

- $n(\omega)$ and $\varepsilon(\omega)$ are macroscopic quantities which describe the interaction between the wave and the medium



Dispersion and attenuation

- Consider a plane wave propagating along x

$$E = E_0 e^{i(kx - \omega t)}; \quad k(\omega) = \frac{n(\omega) \omega}{c}$$

- $n(\omega)$ has a real and an imaginary part:

$$n(\omega) = n_1(\omega) + i n_2(\omega)$$

- Therefore, also the wave number has a real and imaginary part

$$k(\omega) = \frac{[n_1(\omega) + i n_2(\omega)] \omega}{c} = k_1(\omega) + i k_2(\omega)$$

$$k_1(\omega) = \frac{n_1(\omega) \omega}{c}, \quad k_2(\omega) = \frac{n_2(\omega) \omega}{c}$$



Dispersion and attenuation

- $k(\omega) = \frac{[n_1(\omega) + in_2(\omega)] \omega}{c} = k_1(\omega) + ik_2(\omega)$
 $k_1(\omega) = \frac{n_1(\omega) \omega}{c}, k_2(\omega) = \frac{n_2(\omega) \omega}{c}$
- The effect on the propagation in matter is

$$e^{ikx} = \underbrace{e^{ik_1(\omega)x}}_{\text{Propagation term}} \underbrace{e^{-k_2(\omega)x}}_{\text{Attenuation term}}$$

Propagation term

Attenuation term

- The space – time dependence of the wave is thus

$$E = E_0 e^{i[k_1(\omega)x - \omega t]} e^{-k_2(\omega)x}$$



Dispersion and attenuation

- $E = E_0 e^{i[k_1(\omega)x - \omega t]} e^{-k_2(\omega)x}$
- $k_1(\omega) = \frac{\omega n_1(\omega)}{c}$ is the modified wave vector
 - the phase velocity of the wave is $v = \frac{c}{n_1(\omega)}$
- $k_2(\omega) = \frac{\omega n_2(\omega)}{c}$ determines the attenuation of the wave as it propagates in matter
- $n_1(\omega)$: dispersion (modifies the speed of propagation)
 - If the wave crosses the interface between two media it will change direction (refraction)
- $n_2(\omega)$: attenuation



The linear attenuation coefficient

- $k_2(\omega) = \frac{\omega n_2(\omega)}{c}$: attenuation of the amplitude
- Since $I \propto |E|^2$ the attenuation coefficient of the intensity is

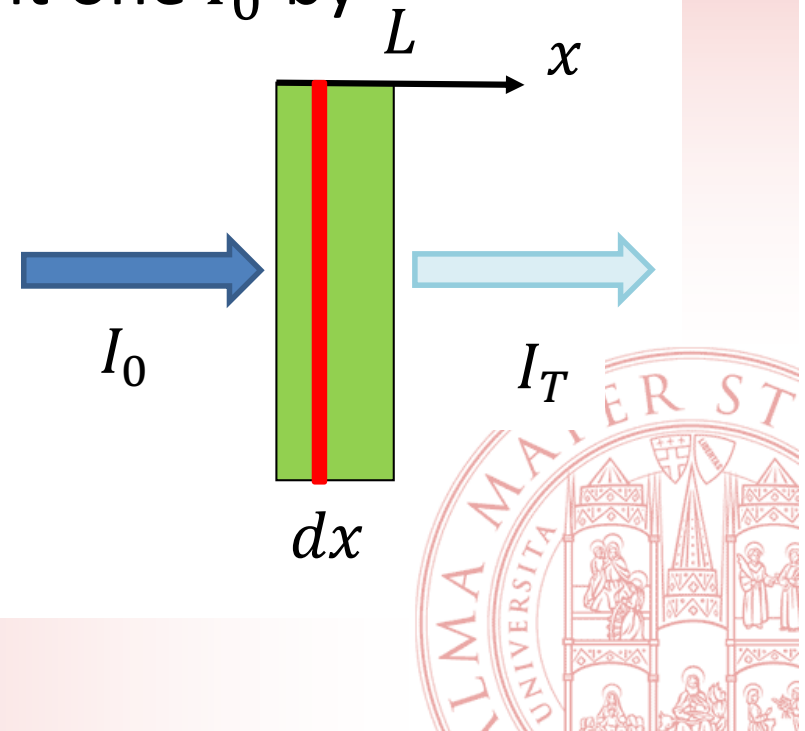
$$\mu = \frac{2\omega n_2(\omega)}{c}$$

- If the total thickness traversed is L the transmitted intensity I_T is related to the incident one I_0 by

$$I_T = I_0 e^{-\mu L}$$

- For an infinitesimal thickness

$$\frac{dI}{I} = -\mu dx$$



Model dielectric function

- Consider matter as an ensemble of damped harmonic oscillators
- The oscillators are the electrons, each of which is bound to a particular atom
- Simplify to the case of a set of identical atoms



A medium as an ensemble of polarizable atoms

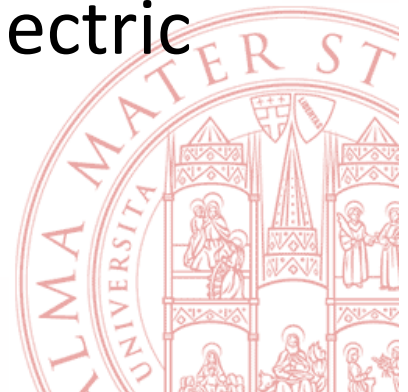
- If the (number) density of identical atoms is ρ

$$\varepsilon(\omega) = 1 + \frac{\rho e^2}{\varepsilon_0 m} \sum_{j=1}^M \frac{f_j}{(\omega_j^2 - \omega^2 - i\gamma_j \omega)}$$

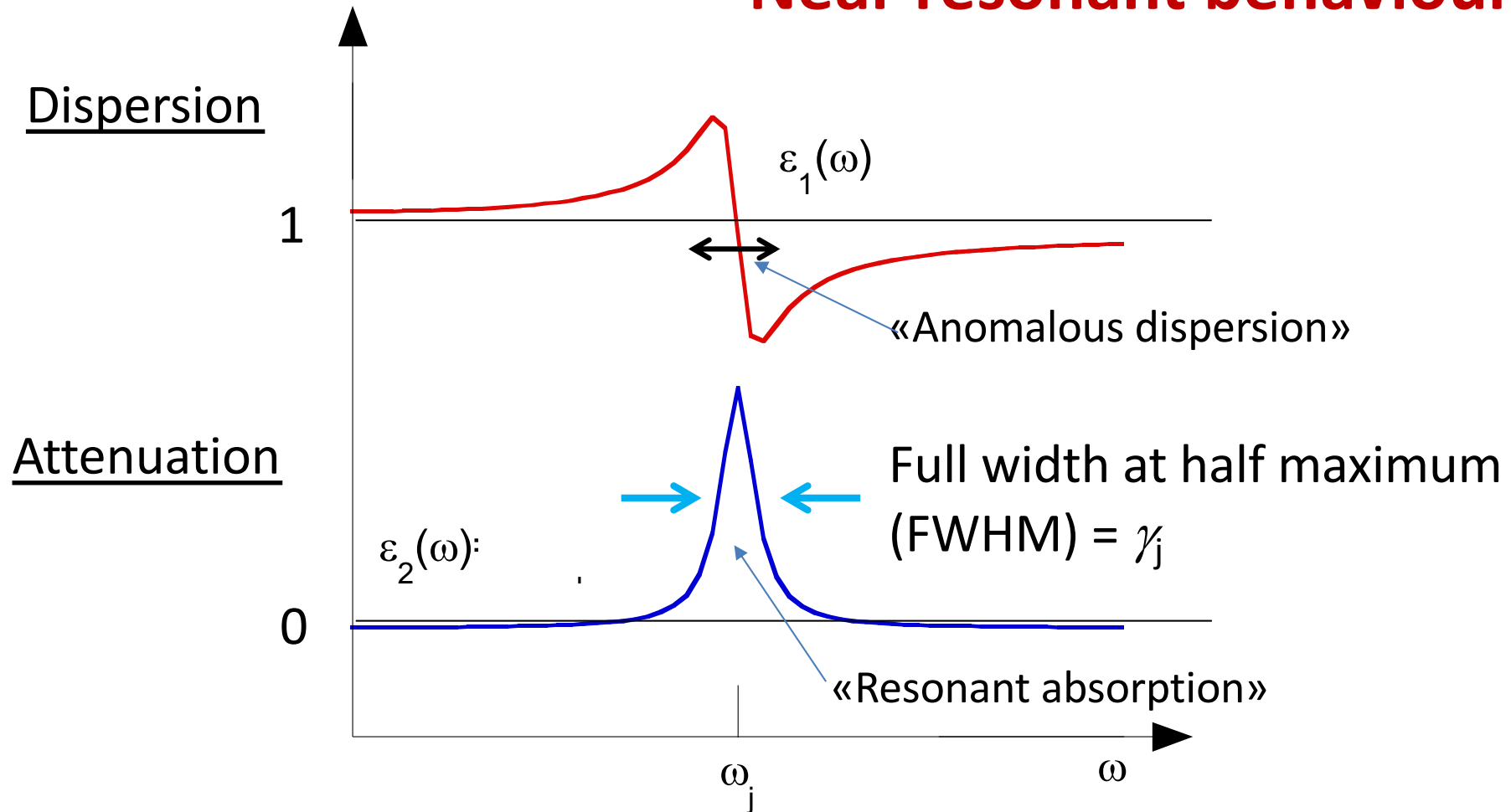
- $\varepsilon_1(\omega) = 1 + \frac{\rho e^2}{\varepsilon_0 m} \sum_{j=1}^M \frac{f_j(\omega_j^2 - \omega^2)}{[(\omega_j^2 - \omega^2)^2 + (\gamma_j \omega)^2]}$

- $\varepsilon_2(\omega) = \frac{\rho e^2}{\varepsilon_0 m} \sum_{j=1}^M \frac{f_j \gamma_j \omega}{[(\omega_j^2 - \omega^2)^2 + (\gamma_j \omega)^2]}$

- Kramers – Heisenberg or electric dipole dielectric function



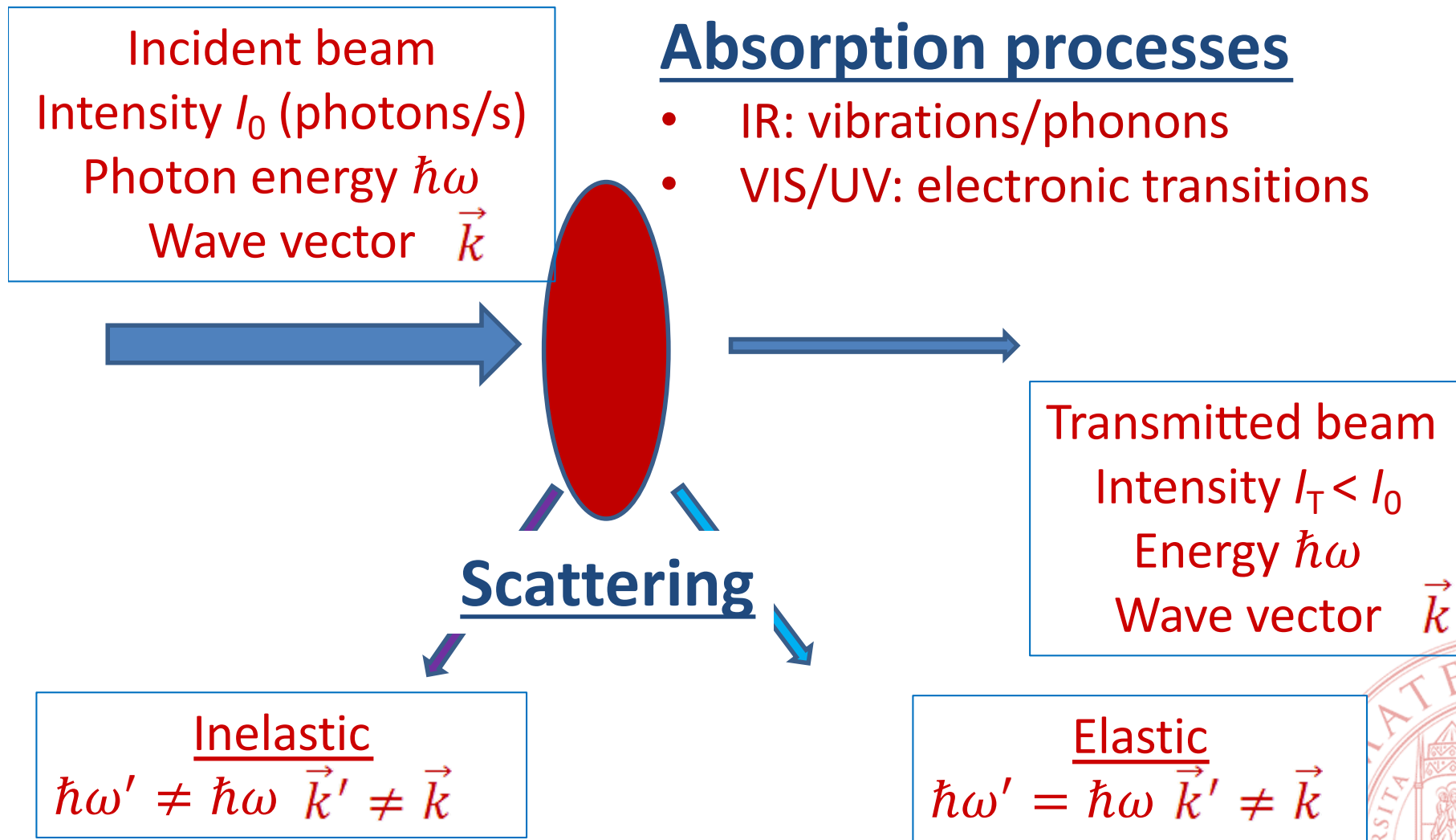
Near resonant behaviour



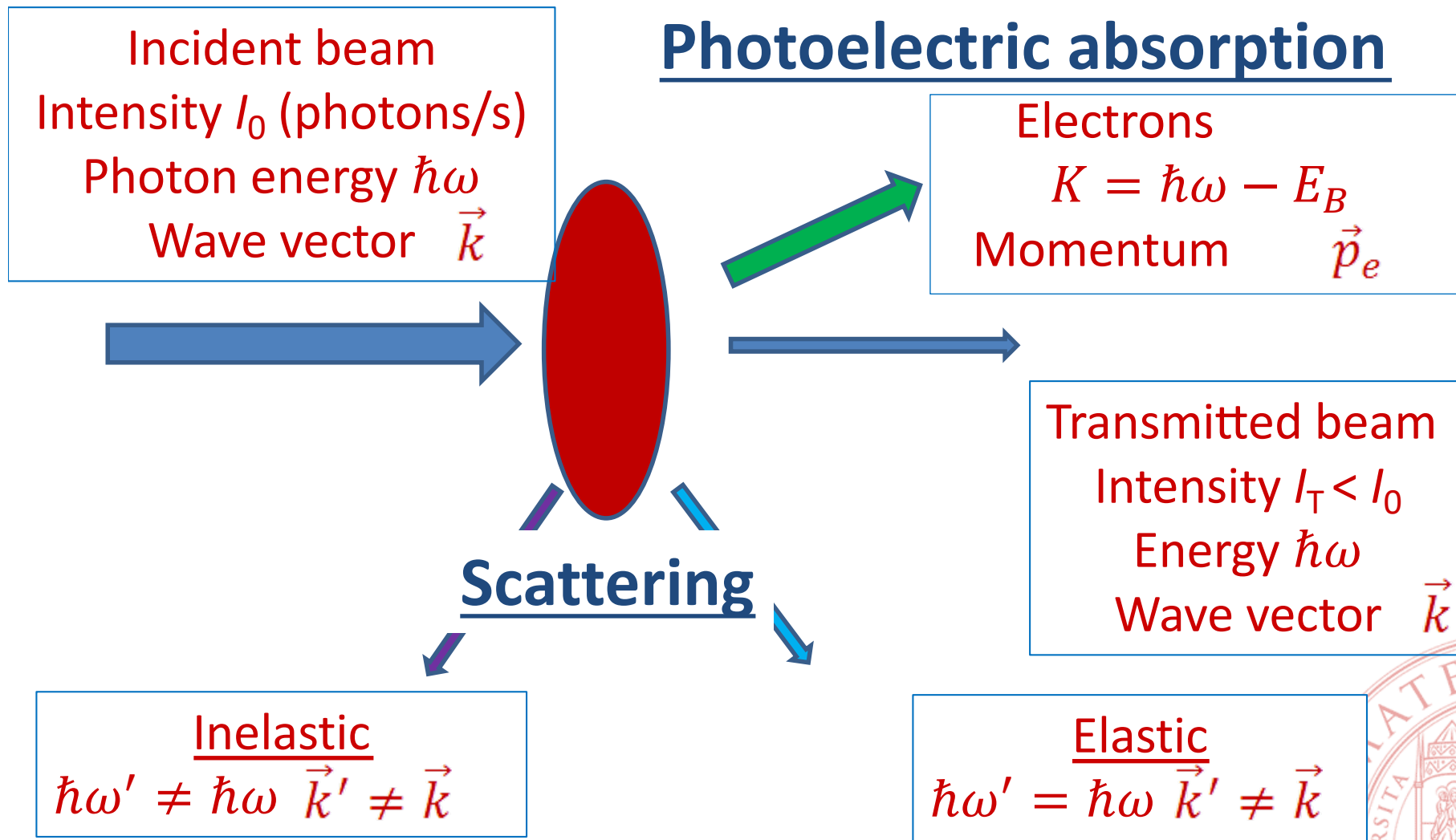
- A simple model which reproduces well the response of polarizable media in many frequency ranges



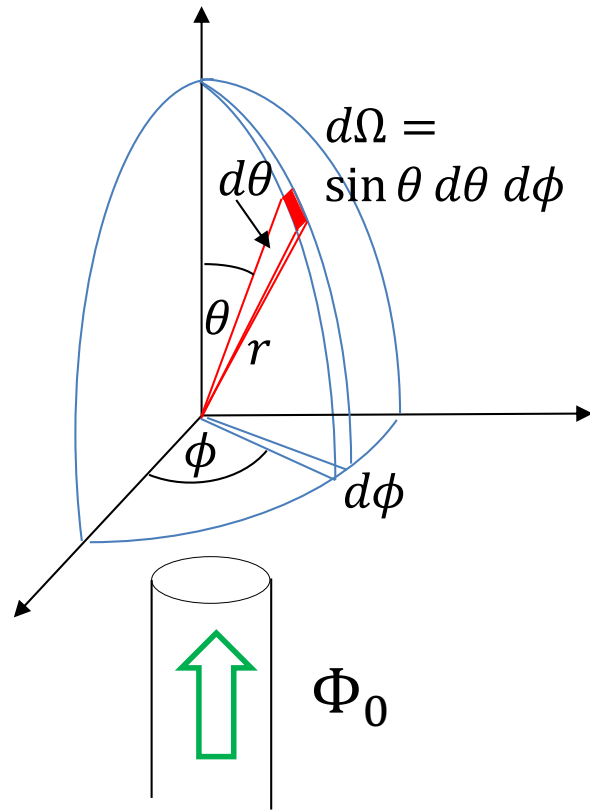
Interaction: atomic approach (IR to vis/UV)



Interaction: atomic approach (UV to X-rays)



Cross section



Impinging beam of
monochromatic
photons, flux Φ_0
 $\Phi_0 = \text{photons}/(\text{s cm}^2)$

- A single atom in the origin
- Interaction produces dN particles per unit time in the solid angle $d\Omega$

$$dN = \Phi_0 d\sigma = \Phi_0 \left(\frac{d\sigma}{d\Omega} \right) d\Omega$$

$$\sigma = \int_{4\pi} \left(\frac{d\sigma}{d\Omega} \right) d\Omega$$

$$[\sigma] = \text{cm}^2$$

$$1 \text{ barn} = 10^{-24} \text{ cm}^2$$

$$\text{Ge, } Z = 32, \hbar\omega = 10 \text{ keV}$$

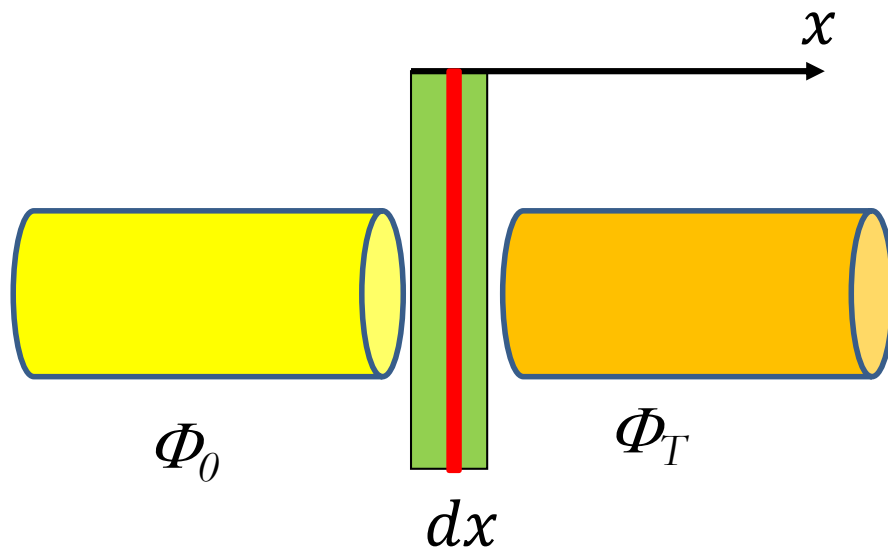
$$\sigma_{\text{photo}} = 4 \times 10^3 \text{ barn}$$

$$\sigma_{\text{el}} = 2 \times 10^2 \text{ barn}$$

$$\sigma_{\text{inel}} = 1 \times 10^1 \text{ barn}$$



Cross section & linear attenuation coefficient



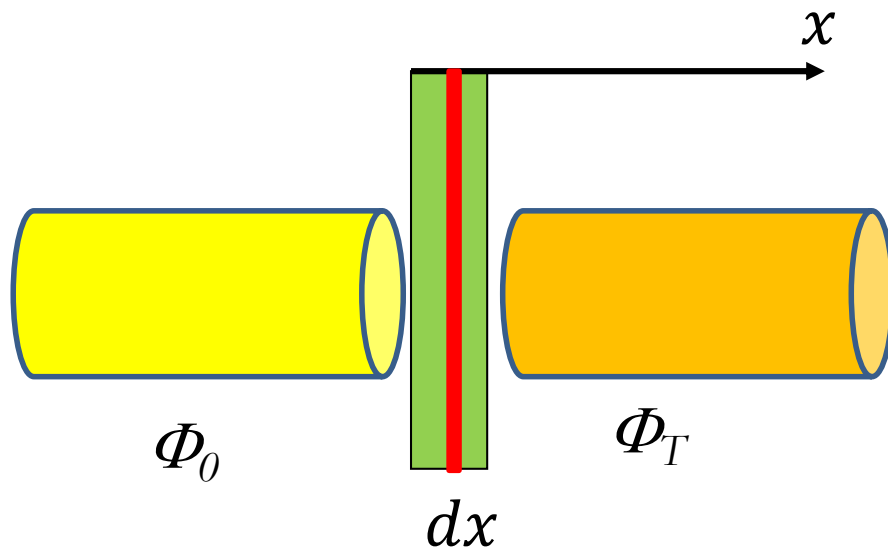
A homogeneous sample composed by identical atoms with density ρ (atoms/cm³)

- Single scattering approximation: the number of particles created by the interaction is $\propto$ number of atoms involved
- The number of particles created by a slab of thickness dx at position x in the full solid angle is

$$\begin{aligned} dN &= \Phi \sigma dn \\ &= \Phi \sigma \rho A dx \\ &= I \sigma \rho dx \end{aligned}$$



Cross section & linear attenuation coefficient



A homogeneous sample
composed by identical atoms
with density ρ

- $dN = I \sigma \rho dx$
- Single particle approximation:
an impinging photon can
create only one particle.
Therefore

$$dN = -dI$$

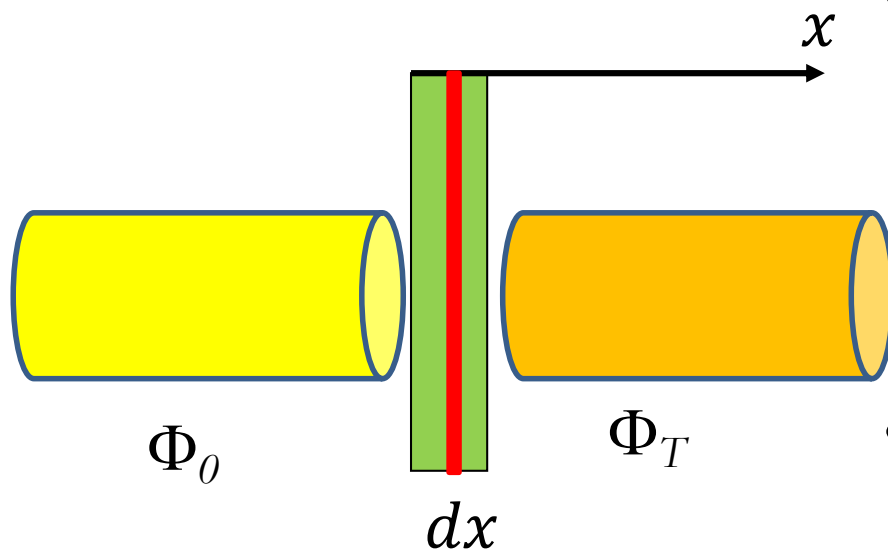
$$dI = -I \sigma \rho dx$$

- This is the same relation
which defines the linear
attenuation coefficient, thus

$$\mu = \sigma \rho$$



Cross section & linear attenuation coefficient



- If the sample is composed of different atoms with densities ρ_i and cross sections σ_i then

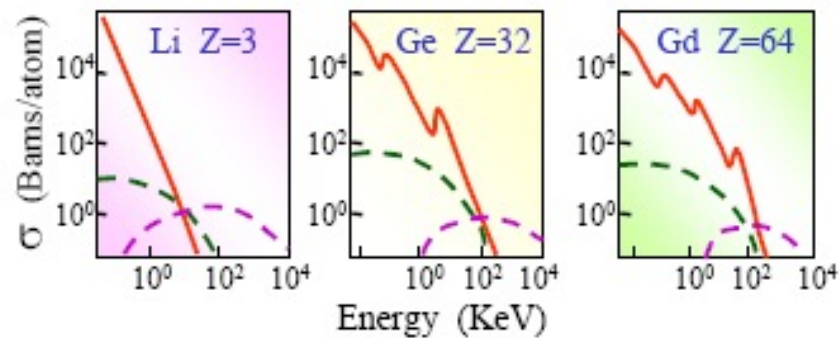
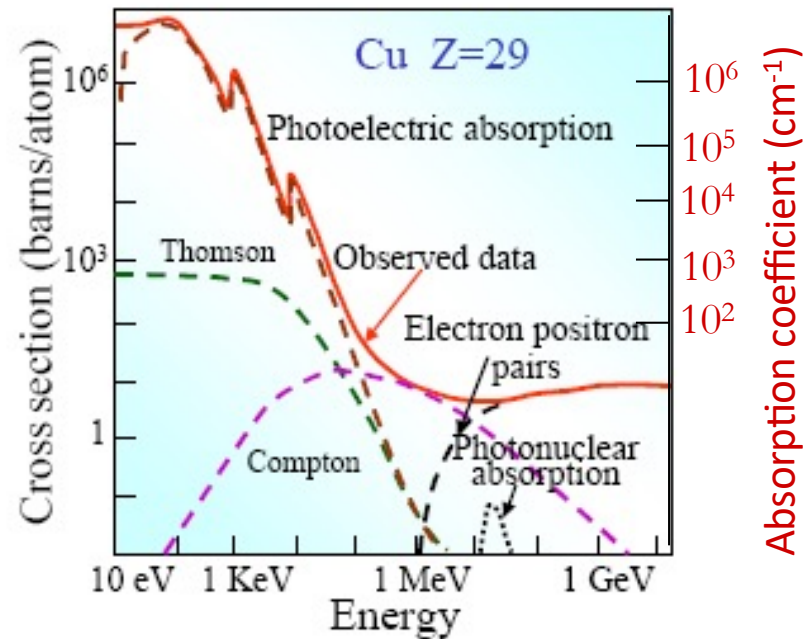
$$\mu = \sum_i \rho_i \sigma_i$$

- Cross sections are also expressed as mass attenuation coefficients, expressed as cm^2/g , so that

$$\begin{aligned} \mu(cm^{-1}) \\ = \rho(g/cm^3) \sigma(cm^2/g) \end{aligned}$$



Cross section of various processes



NB order of magnitude

e.g. Cu @ 10 keV

$$\mu^{-1} \sim 10^{-5} \text{ cm} = 10 \mu\text{m}$$



Elastic scattering from 1 free electron (Thomson)

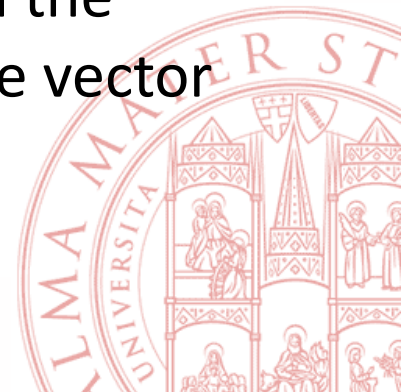
- Thomson scattering = coherent scattering
- The scattered electric field is (linear polarization case)

$$E(\vec{r}, t) = -E_0 r_0 \left(\frac{e^{i(kr - \omega t)}}{r} \right) \sin \alpha$$

$$r_0 = \frac{e^2}{4\pi\epsilon_0 mc^2} \cong 2.82 \times 10^{-15} \text{ m}$$

“Thomson scattering length”
or “classical electron radius”

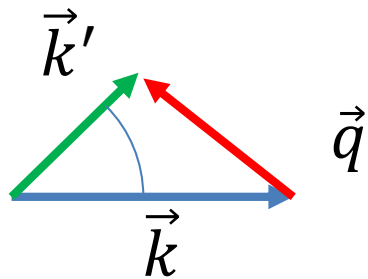
Angle between the polarization
vector and the
scattered wave vector



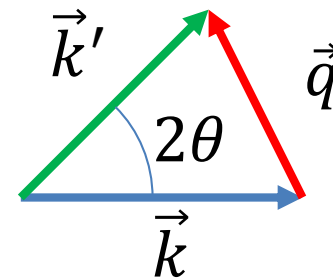
The exchanged wave vector

- $\vec{q} = \vec{k}' - \vec{k}$
- For elastic scattering $|\vec{k}'| = |\vec{k}| = k$ and

$$q = \frac{4\pi}{\lambda} \sin \theta$$



General case



Elastic scattering



Elastic scattering from 1 free electron (Thomson)

- The differential cross section is

$$\frac{d\sigma}{d\Omega} = r_0^2 (\hat{\varepsilon} \cdot \hat{\varepsilon}')^2$$

Polarization vectors of the
incident and scattered waves



Elastic scattering from 1 free electron (Thomson)

- The angle integrated (total) cross section is

$$\sigma = \frac{8\pi}{3} r_0^2$$

- NB: it is independent of energy



Elastic scattering from one atom

- For one atom composed of a nucleus and Z free electrons

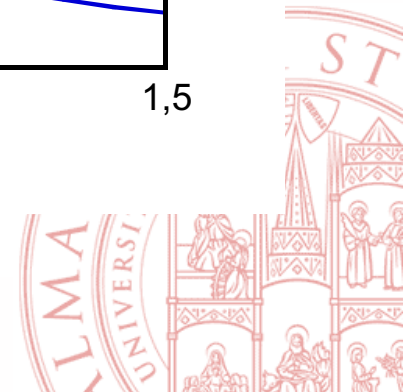
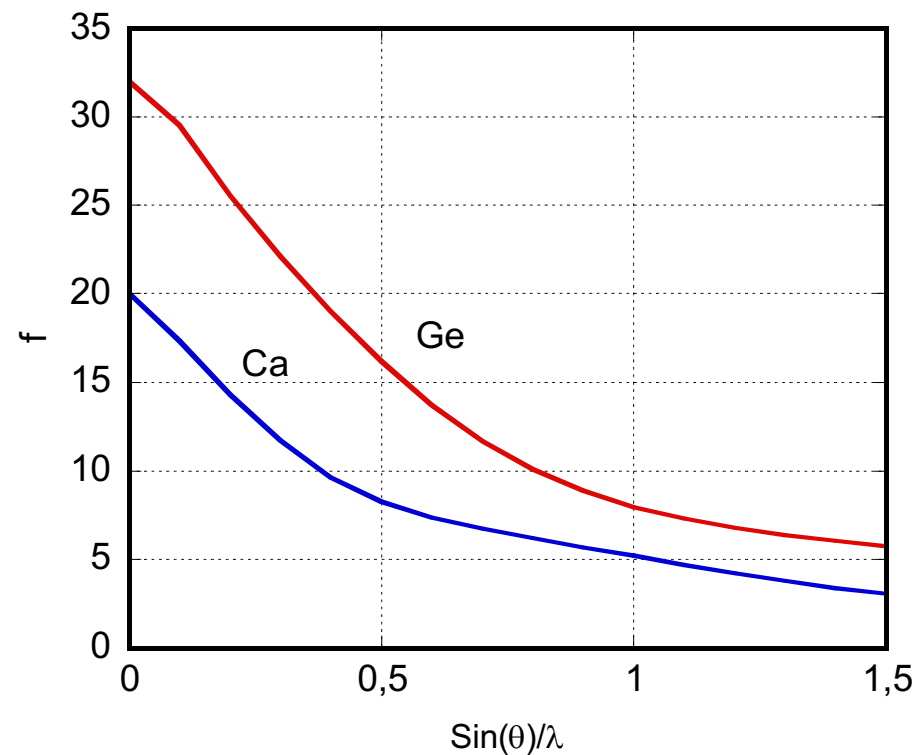
$$E(\vec{r}, t) = -E_0 r_0 \left(\frac{e^{i(kr - \omega t)}}{r} \right) f(Z, \theta) \sin \theta$$

- $f(Z, \theta)$ is the «atomic form factor» or «scattering amplitude»; no physical dimensions
- Scattering is due only to the electrons, nuclei are fixed



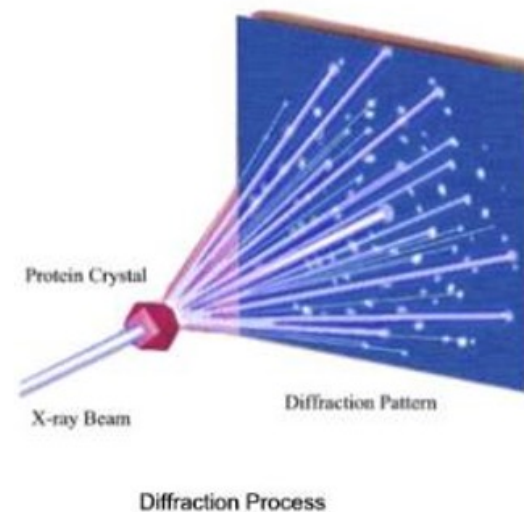
Elastic scattering from one atom

- The differential cross section $\frac{d\sigma}{d\Omega} = \frac{d\sigma_{electron}}{d\Omega} |f(Z, \theta)|^2$
- f depends quasi linearly on Z
- $f(\theta = 0) = Z$



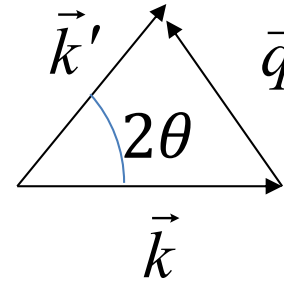
X-ray diffraction

- Elastic X-ray scattering is at the base of X-ray scattering (XRD), a class of methods which are the premiere experimental methods to determine the atomic structure of condensed matter
 - Measure the intensity of x-ray beams scattered by a sample as a function of their deviation



The exchanged wave vector («scattering vector»)

$$\vec{q} = \vec{k}' - \vec{k}$$



- The variation of (linear) momentum is $\hbar\vec{q}$
- In XRD literature the scattering angle is indicated as 2θ
- For elastic scattering

$$|\vec{k}'| = |\vec{k}| = k \quad q = 2k \sin \theta = \frac{4\pi}{\lambda} \sin \theta$$

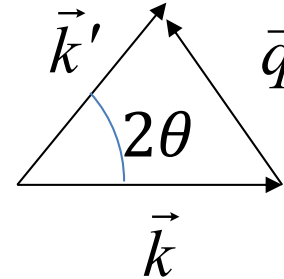
- The space defined by wave vectors is also known as «reciprocal space»



Two ways to change the scattering vector

$$q = \frac{4\pi}{\lambda} \sin \theta$$

$$\vec{q} = \vec{k}' - \vec{k}$$

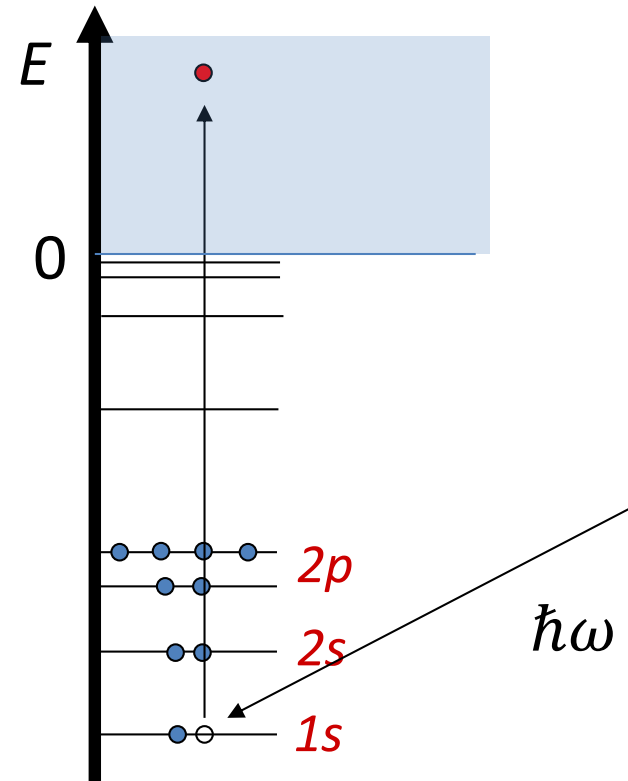


- If λ is constant (more common choice):
measure as a function of θ (angular dispersion)
- If it is more convenient to keep θ constant:
measure as a function of λ (wavelength or energy dispersion)



Photoelectric absorption

- A photon is absorbed and gives its energy to an electron.
- The electron makes a transition to
 - A bound state (excitation)
or
 - An unbound state (ionization): a photoelectron is created

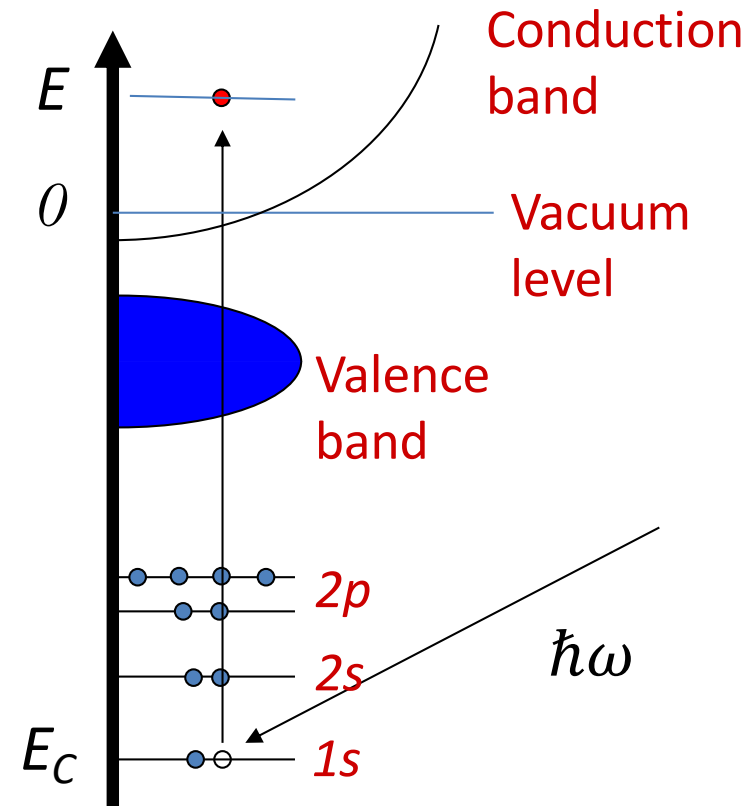


Energetics of photoelectric absorption
in an atom (ionization)



Photoelectric absorption in solids

- In solids valence electrons form bands
 - Insulators and semiconductors: valence and conduction bands
 - metals: conduction band
- The «vacuum level» is the least energy an electron can have to leave the solid (with K.E. = 0)
 - Often taken as reference level
- At sufficiently high energies the photoelectron can be considered «free»: it has only kinetic energy



Conservation of energy

- In the one electron approximation for a transition from a core level:

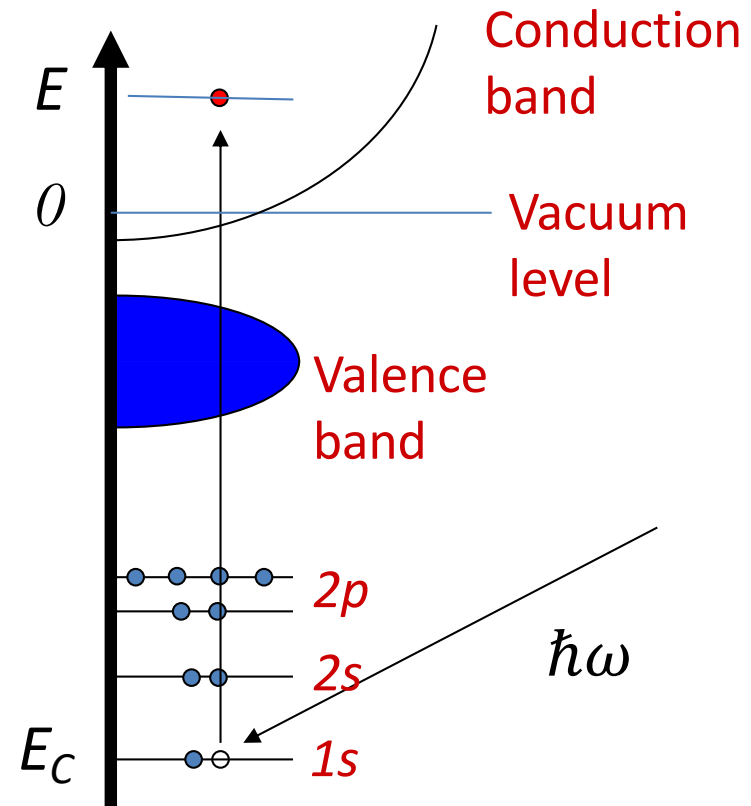
Initial state energy
= $\hbar\omega + E_C$ ($E_C < 0$)

Final state energy = K

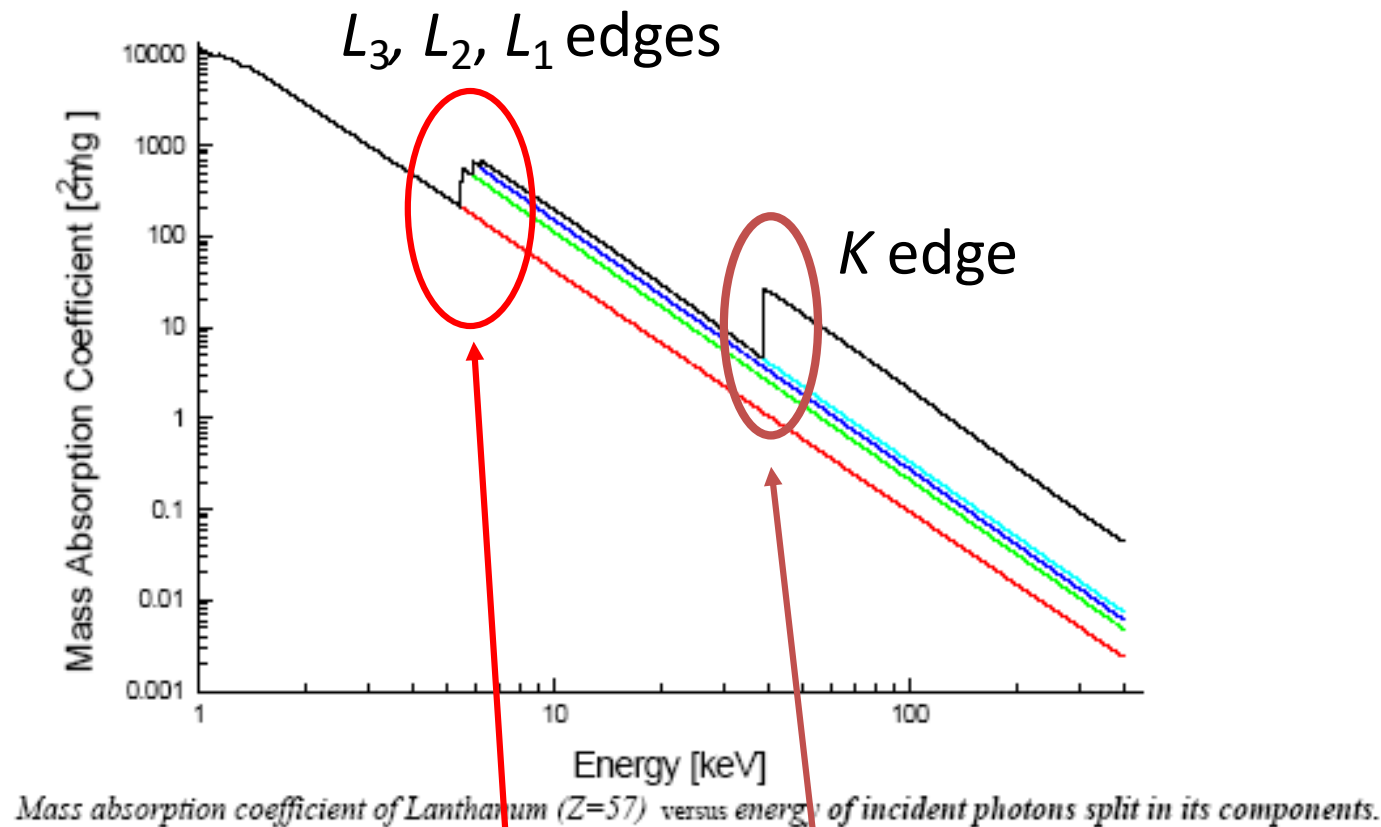
$$K = \hbar\omega + E_C$$

$E_C = -E_B$ (binding energy)

$$K = \hbar\omega - E_B$$



Absorption coefficient: energy dependence

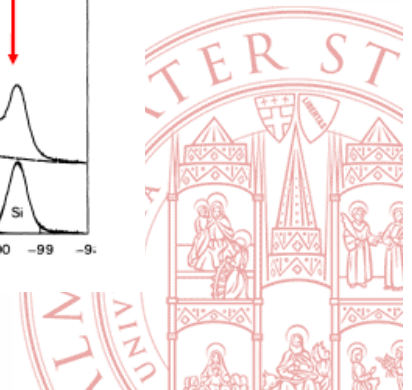
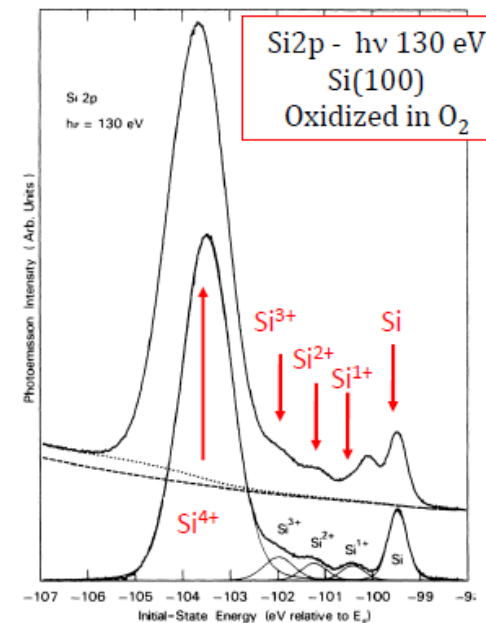
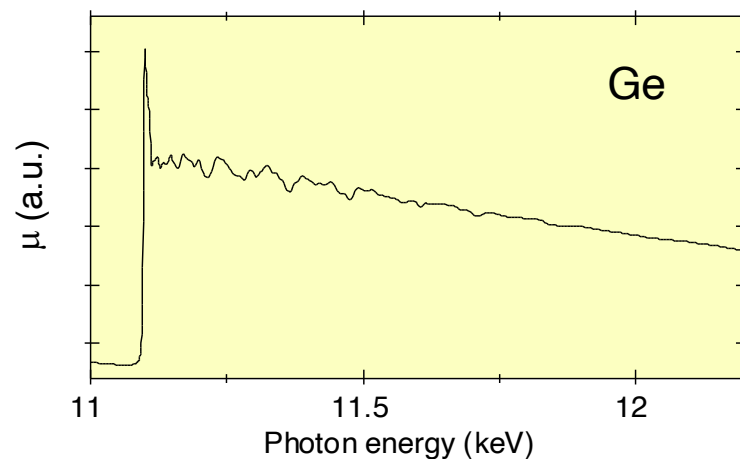


Absorption edges



Photoemission and X-ray absorption spectroscopy

- Photoelectric absorption is at the base of
 - Photoemission spectroscopy (PES, UPS, XPS), the premiere experimental method to determine electronic structure
 - measure the spectrum of electrons emitted from a sample
 - X-ray absorption spectroscopy (XAS, XAFS), the premiere experimental method to determine local atomic and electronic structure with chemical sensitivity
 - measure the photon energy dependence of the fine structure of the attenuation coefficient



Relationship between two approaches

- Two approaches to describe the interaction between x-rays and matter
 - «Macroscopic»: by means of the dielectric function which describes the overall response
 - «Microscopic»: by means of interactions between photons and atoms
- What is the relation between these approaches?
- It is possible to derive a simple relation between the index of refraction and the atomic form factor



Relation between n and f

- For a sample composed of identical atoms with form factor f and density ρ

$$n(\omega) - 1 = - \frac{2\pi r_0 \rho f(\omega, q = 0)}{k^2}$$

- $n(\omega)$ has a real and imaginary part: also $f(\omega)$!
 - Real part: dispersion
 - Imaginary part: attenuation



«Anomalous" corrections to the form factor

- It is common to separate the dependence on $\vec{q}$ and ω : $f(\vec{q}, \omega) = f^0(\vec{q}) + f'(\omega) - if''(\omega)$
 - $f^0(\vec{q})$: atomic scattering far from resonance frequencies / absorption edges
 - $f'(\omega)$: correction to dispersive part, important near resonance frequencies
 - $f''(\omega)$: correction to attenuation part, important near resonance frequencies



The total cross section

- The total cross section determines the attenuation, $\sigma_T(\omega) = \frac{\mu(\omega)}{\rho} = \frac{2\omega n_2(\omega)}{c\rho}$
- Since $n_2(\omega) = \frac{2\pi r_0 \rho f''(\omega, q=0)}{k^2}$

$$\sigma_T(\omega) = \frac{4\pi r_0}{k} f''(\omega, q = 0)$$

one form of the «optical theorem», which links the total cross section to the imaginary part of the forward scattering amplitude



Interaction between EM radiation and hydrogen-like atoms: semiclassical theory



References

- B.H. Bransden & C.J. Joachain, “Physics of atoms and molecules”, 2nd edition, Pearson Education – Prentice Hall (2003)
Chapter 4 (except 4.4), in parts



Introduction

- Semi-classical theory of the interaction between radiation and hydrogen – like atoms.
- Semi-classical since
 - Radiation is treated as wave
 - Atom is treated with quantum mechanics
- This approach is adequate since it can describe scattering and stimulated absorption and emission
 - It cannot describe spontaneous emission
- Full quantum treatment requires quantization of EM field: more formal
- All phenomena occurring in hydrogen – like atoms are present in many electron ones



Interaction between a wave and an atom

- The interaction is treated with time dependent perturbation theory
- The unperturbed atom's Hamiltonian is H_0
- The perturbation is an EM wave and the time dependent interaction Hamiltonian is $H'(t)$
- The EM wave has a harmonic dependence on time, thus it is expressed by an Hermitian operator of the type

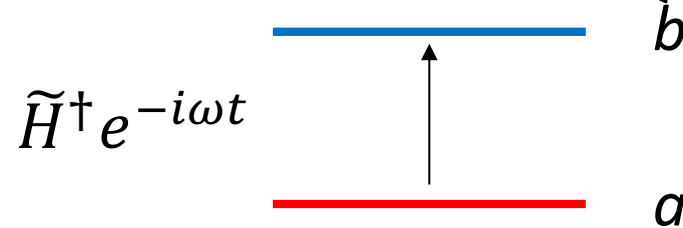
$$H'(t) = \tilde{H}e^{i\omega t} + \tilde{H}^\dagger e^{-i\omega t}$$

in which $\tilde{H}$ is an operator which does not depend on time

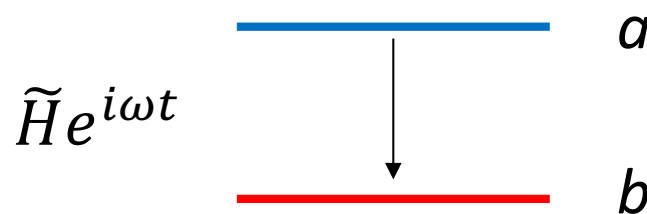


Time dependent perturbation theory

- The unperturbed atom has eigenstates labelled « a » and « b » with energies E_a^0 and E_b^0
 - Often called the «initial» and «final» states
- It can be demonstrated that the transition probability is maximized for two «resonant» conditions deriving from different terms in $H'(t)$



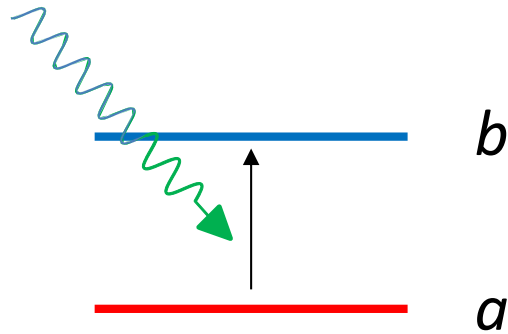
$$\hbar\omega = E_b^0 - E_a^0$$



$$\hbar\omega = E_a^0 - E_b^0$$



Time dependent perturbation theory

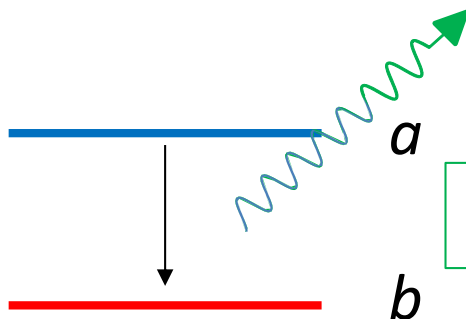


$$\tilde{H}^\dagger e^{-i\omega t}$$

$$\hbar\omega = E_b^0 - E_a^0$$

Stimulated absorption

- A photon of energy $\hbar\omega$ is absorbed by the atom
- The atom makes a transition from a to b



$$\tilde{H} e^{i\omega t}$$

$$\hbar\omega = E_a^0 - E_b^0$$

Stimulated emission

- A photon of energy $\hbar\omega$ is emitted from the atom
- The atom makes a transition from a to b



Fermi's golden rule: transition to discrete states

- For the case of absorption it can be demonstrated that, to first order in the perturbation, the transition probability per unit time for transitions between discrete levels a and b is

$$W_{ba} = \frac{2\pi}{\hbar} |\tilde{H}^\dagger_{ba}|^2 \delta(E_b^0 - E_a^0 - \hbar\omega)$$

- $\tilde{H}^\dagger_{ba} = \langle b | \tilde{H}^\dagger | a \rangle$ is the matrix element of the perturbation
- The Dirac δ function is an expression of the conservation of energy
 - Apparently unphysical: the probability is always 0 except at resonance in which case it diverges. This will be resolved by introducing the concept of lifetime of the eigenstates



Fermi's golden rule: transition to continuum states

- For absorption with final states b in the continuum it can be shown that

$$W_{ba} = \frac{2\pi}{\hbar} |\tilde{H}^\dagger_{ba}|^2 \rho(E_b^0)$$

with the condition that $E_b^0 = E_a^0 + \hbar\omega$

- $\rho(E)$ is the density of states, such that the number of states between E and $E + dE$ is

$$dN = \rho(E) dE$$



Dipole approximation

- In deriving the cross section for photoelectric absorption a key element is the dipole approximation
- Call d the spatial extent of the initial state wave function. At most, it is the atomic size $\sim 1 \text{ \AA}$.
- The dipole approximation is
$$2\pi d \ll \lambda$$
- Visible range: dipole approximation always valid
- For valence initial states the dipole approximation is valid up to the UV.
- For core level initial states of not too light atoms the dipole approximation continues to be valid even in the X-ray range.



The cross section in the dipole approx.

$$\sigma = 4\pi^2 \hbar\omega \alpha |\langle\psi_b|\hat{\epsilon} \cdot \vec{r}|\psi_a\rangle|^2 \delta(E_b^{(0)} - E_a^{(0)} - \hbar\omega)$$

- Clearly, dimensions = L^2
- The order of magnitude is determined by the dipole matrix element, an effective “area” roughly of the order of a_0^2 , depending on the overlap of initial and final wavefunctions
- The Dirac δ function is an expression of the conservation of energy
- The apparent unphysical divergence will be solved introducing the concept of lifetime of states



Selection rules

- Using the properties of the spherical harmonics it can be shown that the selection rule on ℓ is

$$\Delta\ell = \pm 1$$

- The selection rule on m depends on the state of polarization of the radiation

➤ For linealy polarized radiation

$$\Delta m = 0$$

➤ For circularly polarized radiation

$$\Delta m = \pm 1$$



Selection rules

$$\Delta\ell = \pm 1$$

Conservation of angular momentum
(modulus)

$$\Delta m = \pm 1$$

Conservation of angular momentum
(quantization axis component)

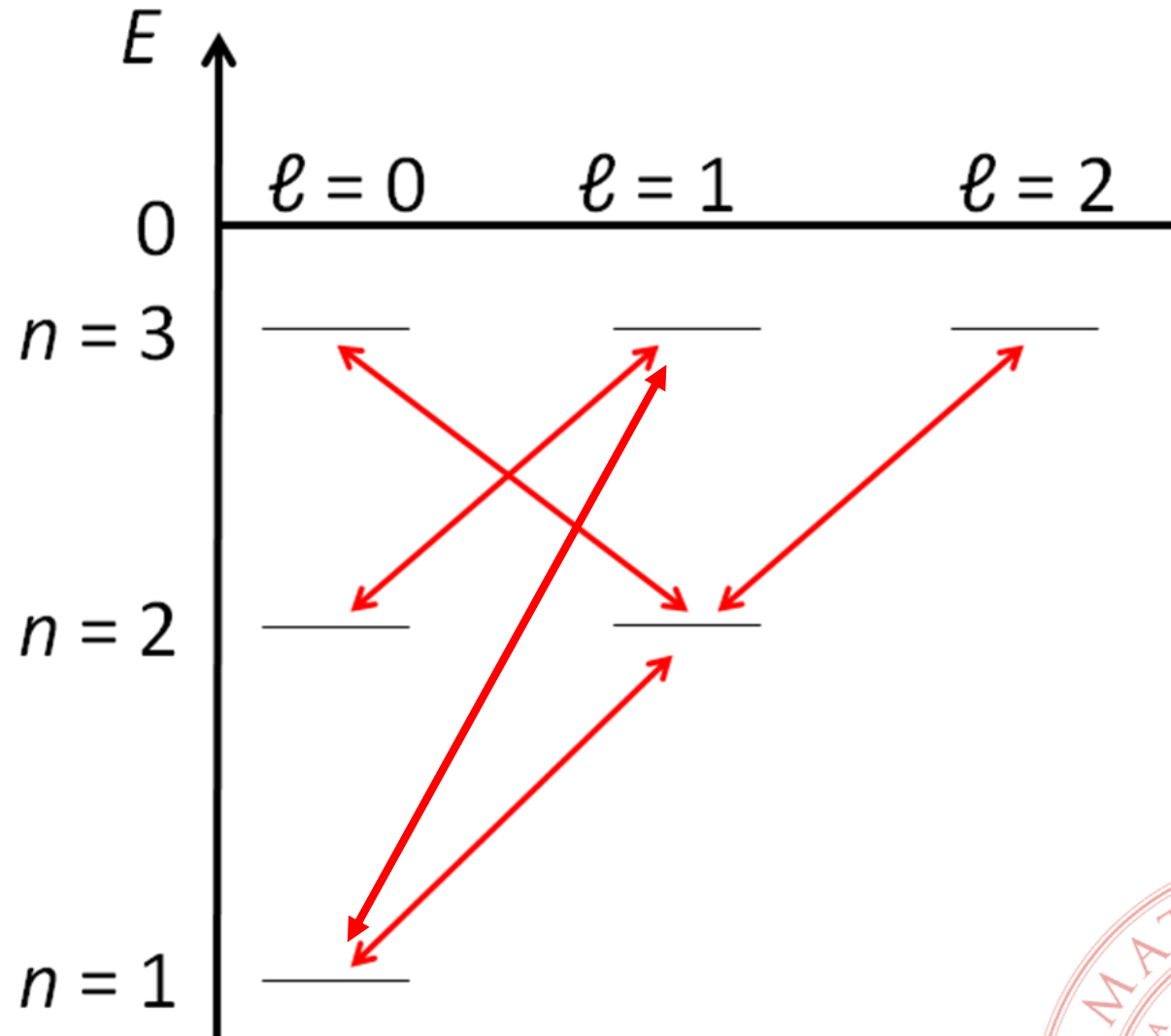


Selection rules

$$\Delta \ell = \pm 1$$

$$\Delta m_\ell = 0, \text{ lin}$$

$$= \pm 1, \text{ circ}$$



Lifetime and lineshape

- We have implicitly assumed that all atomic eigenstates have infinite lifetime. Apart from the fundamental state (1s) this is not true.
- All states have a finite lifetime due to
 - Spontaneous emission, also present for isolated atoms
 - Collisions between atoms, which induce electron transitions, present in gases at non negligible pressure
- If N_0 atoms are in a given state at $t = 0$, their number decays exponentially as

$$N(t) = N_0 e^{-\frac{t}{\tau}}$$

- For the H atom, the lifetimes τ of electronic states are

Level	2p	3s	3p	3d	4s	4p	4d	4f
Lifetime (ns)	1.6	160	5.4	15.6	230	12.4	36.5	73



Lifetime and lineshape

- A finite lifetime implies a spectral broadening
 - Transitions do not occur at a single photon energy

$$\hbar\omega_{ba} = E_b^0 - E_a^0$$

- Transitions occur in a band centered around $\hbar\omega_{ba}$ with a broadening Γ which can be estimated from the Heisenberg uncertainty principle
- From the energy – time Heisenberg uncertainty principle, interpret τ as uncertainty in time, thus

$$\Gamma \geq \frac{\hbar}{\tau}$$



Lifetime and lineshape

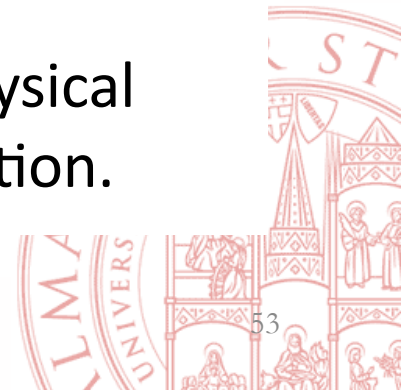
- It can be proved that this spectral broadening results in a Lorentzian lineshape as a function of energy
- For a transition between states with lifetimes τ_a and τ_b the Lorentzian half width at half maximum (HWHM) is

$$\Gamma = \hbar \left(\frac{1}{\tau_a} + \frac{1}{\tau_b} \right)$$

- The energy dependence of the cross section, the lineshape, is proportional to

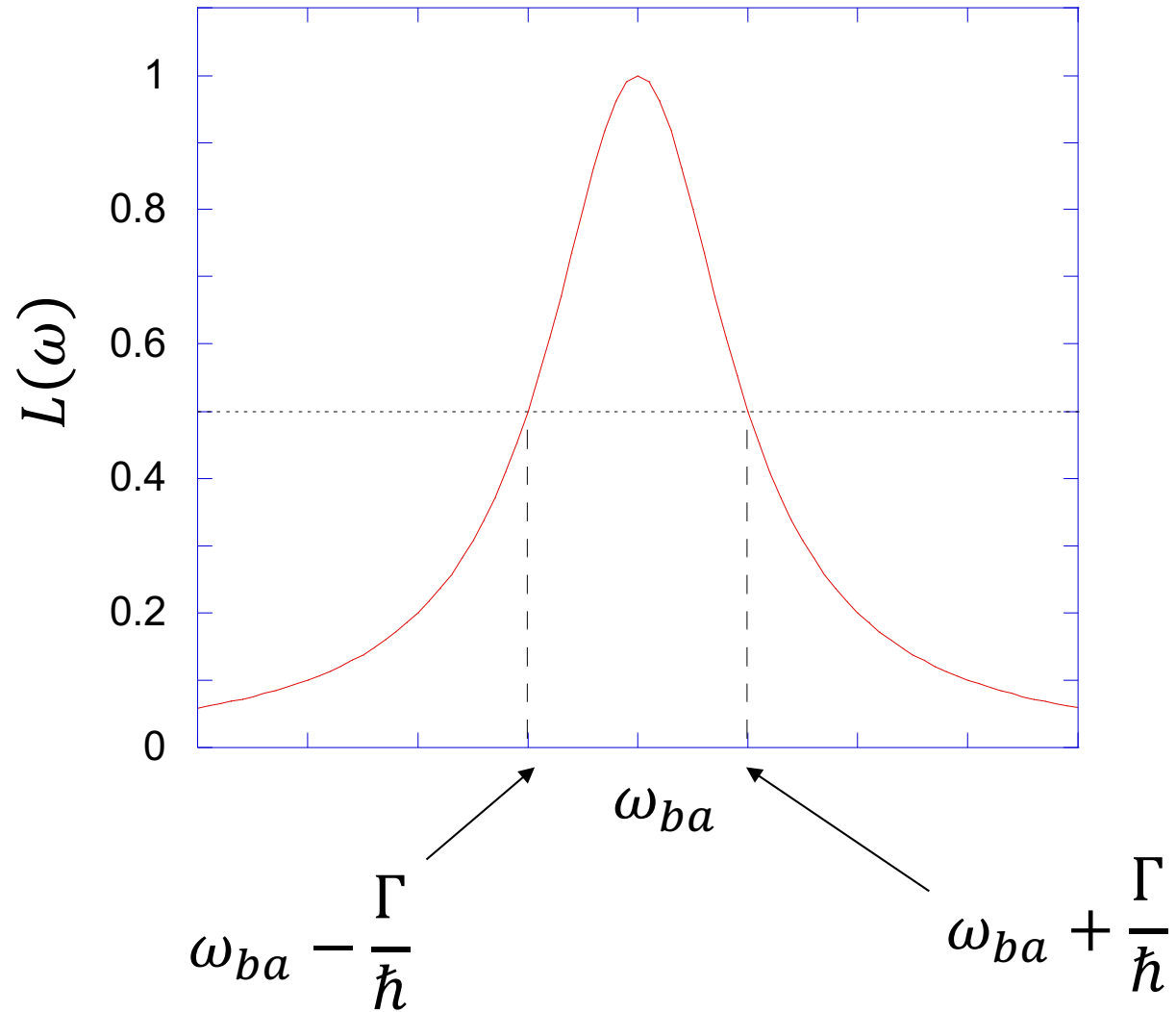
$$L(\omega) = \frac{\Gamma^2}{\hbar^2 (\omega_{ba} - \omega)^2 + \Gamma^2}$$

- This spectral broadening resolves the apparently unphysical result that the cross section is proportional to a δ function.

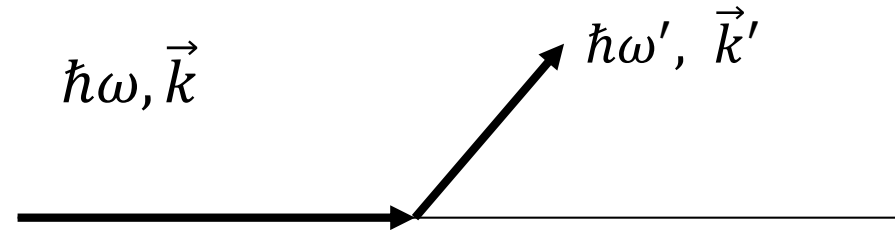


Lifetime and lineshape

$$L(\omega) = \frac{\Gamma^2}{\hbar^2(\omega_{ba} - \omega)^2 + \Gamma^2}$$



Scattering of radiation



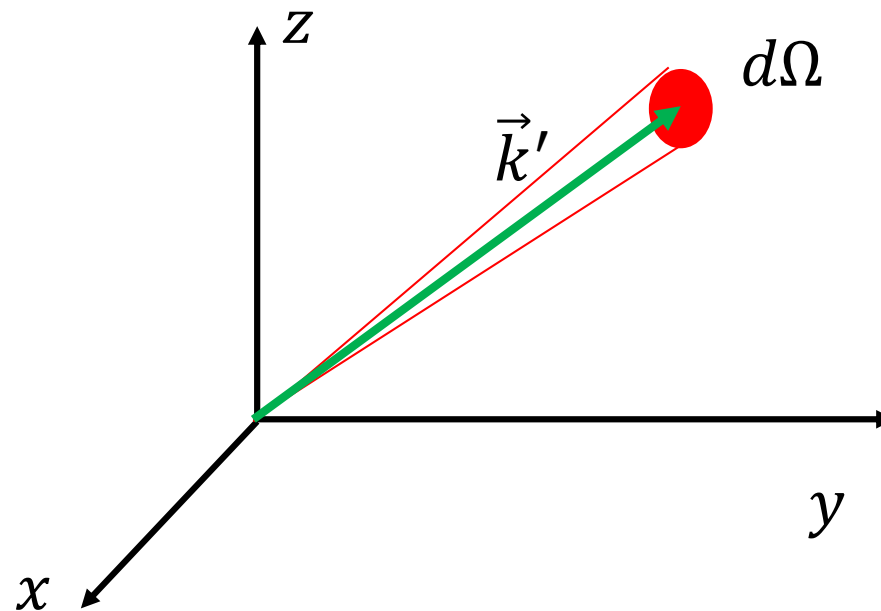
- From the particle point of view scattering is a 2 photon process: a photon is absorbed (destroyed) and another is emitted (created).
- The scattered photon in general has a different energy and different wave vector (modulus and/or direction)

- $\omega' = \omega$: elastic scattering
 - In general: Raleigh
 - For a free electron: Thomson

- $\omega' \neq \omega$: inelastic scattering
 - In general: Raman
 - For a free electron: Compton ($\omega' < \omega$)

Scattering geometry

- We will discuss the cross section for scattering in which the scattered photon has direction defined by the wave vector $\vec{k}'$ within an infinitesimal solid angle $d\Omega$



X-ray scattering cross section

- It can be proved that the differential cross section is

$$\begin{aligned}\frac{d\sigma}{d\Omega} &= \frac{e^4}{16\pi^2 \varepsilon_0^2 m^2 c^4} \left(\frac{\omega'}{\omega} \right) (\hat{\varepsilon} \cdot \hat{\varepsilon}')^2 |\langle b | e^{-i\vec{q} \cdot \vec{r}} | a \rangle|^2 \\ &= r_0^2 \left(\frac{\omega'}{\omega} \right) (\hat{\varepsilon} \cdot \hat{\varepsilon}')^2 |\langle b | e^{-i\vec{q} \cdot \vec{r}} | a \rangle|^2\end{aligned}$$

$r_0 = \frac{e^2}{4\pi\varepsilon_0 mc^2} \cong 2.82 \times 10^{-15} \text{ m}$, the «classical electron radius» or «Thomson scattering length»



Scattering of radiation: general case

- It can be demonstrated that in the dipole approximation the differential cross section is the Kramers – Heisenberg formula

$$\frac{d\sigma}{d\Omega} = r_0^2 \omega \omega'^3 \left| m \sum_n \left[\frac{(\hat{\varepsilon}' \cdot \vec{r}_{bn})(\hat{\varepsilon} \cdot \vec{r}_{na})}{(E_n^0 - E_a^0 - \hbar\omega)} + \frac{(\hat{\varepsilon} \cdot \vec{r}_{bn})(\hat{\varepsilon}' \cdot \vec{r}_{na})}{(E_n^0 - E_a^0 + \hbar\omega')} \right] \right|^2$$

with the condition that

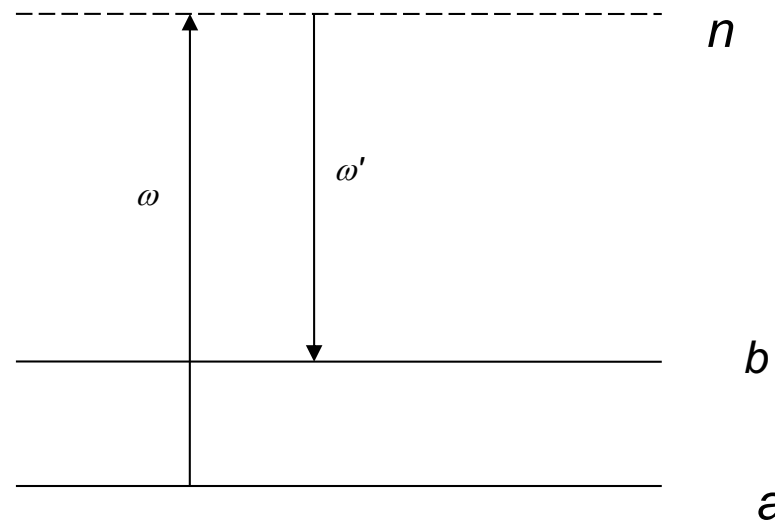
$$E_a^0 + \hbar\omega = E_b^0 + \hbar\omega'$$

and the sum is over all atomic states n .



Scattering of radiation: general case

- $\frac{d\sigma}{d\Omega} = r_0^2 \omega \omega'^3 \left| m \sum_n \left[\frac{(\hat{\varepsilon}' \cdot \vec{r}_{bn})(\hat{\varepsilon} \cdot \vec{r}_{na})}{(E_n^0 - E_a^0 - \hbar\omega)} + \frac{(\hat{\varepsilon} \cdot \vec{r}_{bn})(\hat{\varepsilon}' \cdot \vec{r}_{na})}{(E_n^0 - E_a^0 + \hbar\omega')} \right] \right|^2$
- A «picture» of this equation
 - Scattering is due to the sum of «virtual» transitions to intermediate states.
 - Conservation of energy is valid only globally, not for transitions to intermediate «virtual» states



“Anomalous” scattering

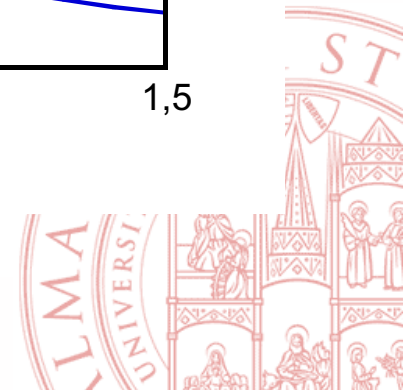
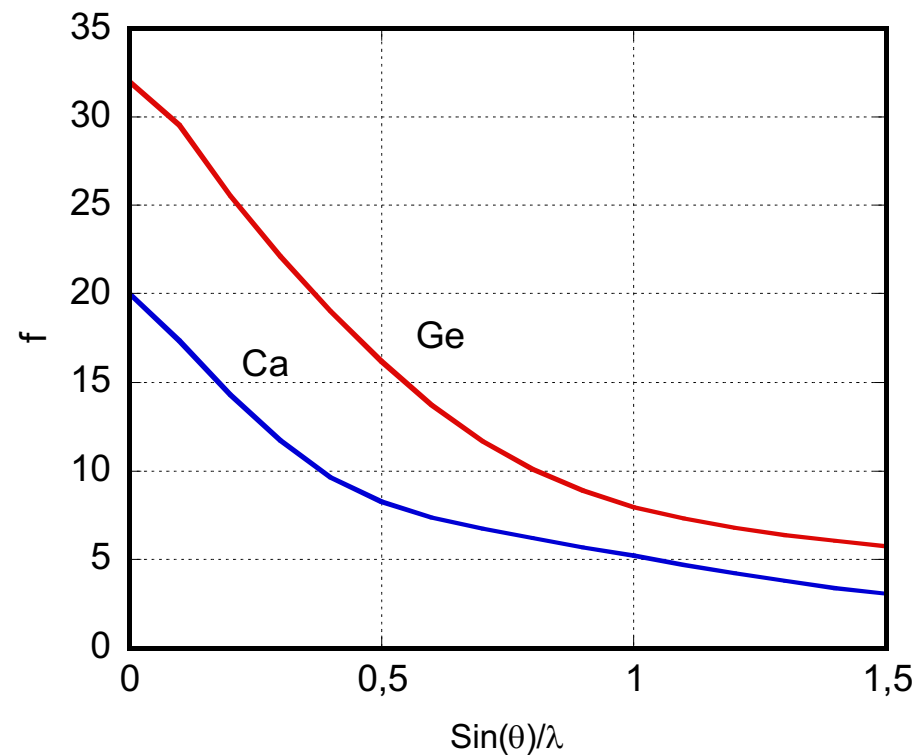


Elastic scattering from one atom

- Recall the differential cross section

$$\frac{d\sigma}{d\Omega} = \frac{d\sigma_{electron}}{d\Omega} |f(Z, \theta)|^2$$

- f depends quasi linearly on Z
- $f(\theta = 0) = Z$



The atomic form factor

- It can be shown that the form factor is the Fourier transform of the electronic density

$$f(\vec{q}) = \int_{V_{at}} d^3r \rho_{e-at}(\vec{r}) e^{-i\vec{q} \cdot \vec{r}}$$



«Anomalous» corrections to f

- More realistically, electrons are not free but bound to specific atoms. This introduces «anomalous» corrections to the atomic form factor, with a real and imaginary component

$$f(q, \omega) = f^0(q) + f'(\hbar\omega) + if''(\hbar\omega)$$

- These corrections are significant when the X-ray photon energy, $\hbar\omega$, is close to an absorption edge (resonance condition)

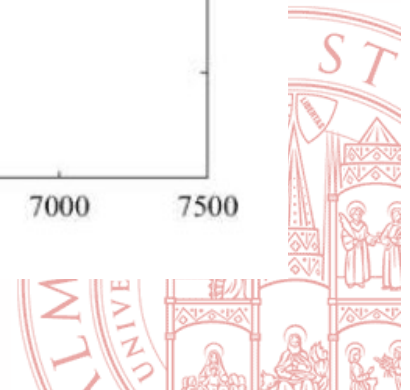
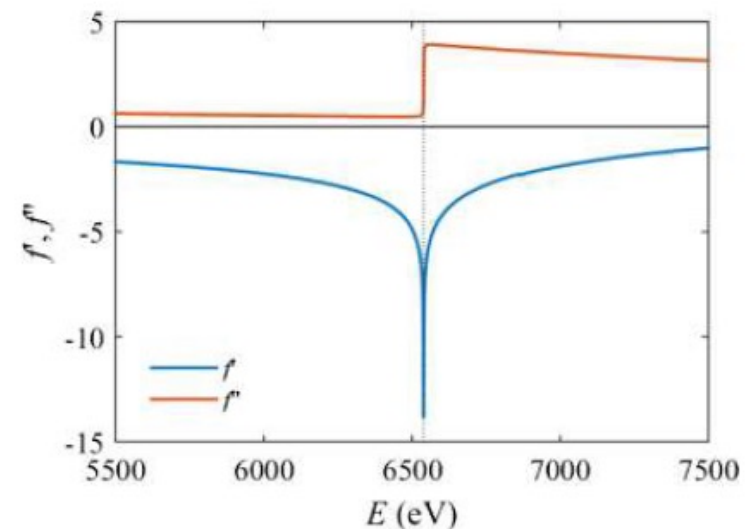


«Anomalous" corrections to the form factor

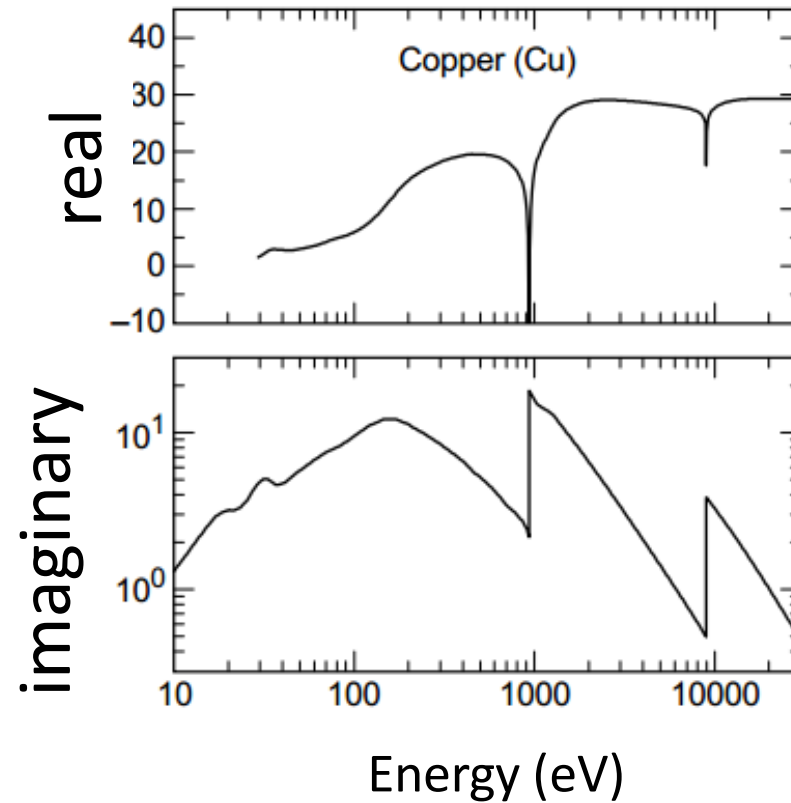
- It is common to separate the dependence on $\vec{q}$ and ω :
$$f(\vec{q}, \omega) = f^0(\vec{q}) + f'(\omega) + if''(\omega)$$
 - $f^0(\vec{q})$: atomic scattering far from resonance frequencies / absorption edges
 - $f'(\omega)$: correction to dispersive part, important near resonance frequencies
 - $f''(\omega)$: correction to attenuation part, important near resonance frequencies

- Corrections to the form factor for Mn, as a function of the photon energy $\hbar\omega$

Giannini et al, Prog. Mat Science, 112, 100667 (2020), © Elsevier



Form factor of Cu



Optical theorem:

$$\sigma_T(\omega) = \frac{4\pi r_0}{k} f''(\omega, q = 0)$$



Origin of “anomalous” corrections

- In the atomic approach: differential cross section for elastic scattering in the dipole approximation

$$\frac{d\sigma}{d\Omega} = r_0^2 m^2 \omega^4 \left| \sum_n \left[\frac{(\hat{\epsilon}' \cdot \vec{r}_{an})(\hat{\epsilon} \cdot \vec{r}_{na})}{(E_n^0 - E_a^0 - \hbar\omega)} + \frac{(\hat{\epsilon} \cdot \vec{r}_{an})(\hat{\epsilon}' \cdot \vec{r}_{na})}{(E_n^0 - E_a^0 + \hbar\omega')} \right] \right|^2$$

- If $\hbar\omega \cong E_{n^*}^0 - E_a^0$ one particular term will dominate the sum
- If Γ is the lifetime of state n^* it can be shown that

$$\frac{d\sigma}{d\Omega} = r_0^2 \omega^4 m^2 \frac{(\hat{\epsilon}' \cdot \vec{r}_{an^*})^2 (\hat{\epsilon} \cdot \vec{r}_{n^*a})^2}{(E_n^0 - E_a^0 - \hbar\omega)^2 + \frac{1}{4}\Gamma^2}$$



Origin of “anomalous” corrections

- In the macroscopic approach: relation between the index of refraction and atomic form factor

$$n(\omega) - 1 = - \frac{2\pi r_0 \rho f(\omega, q=0)}{k^2}$$

- Since n has real and imaginary part so must f

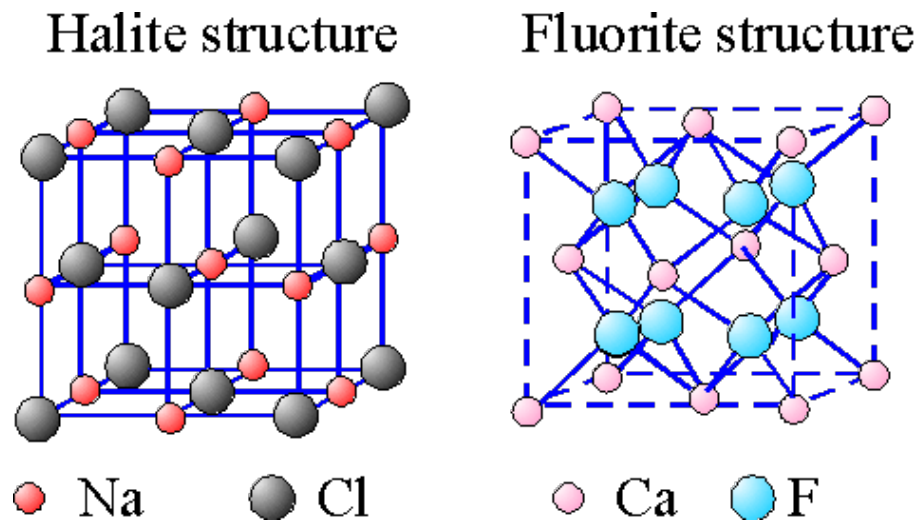


Diffraction from crystals: kinematic theory



Kinematic diffraction theory

- Now consider scattering of x-rays from a crystal characterized by long range order with translational symmetry

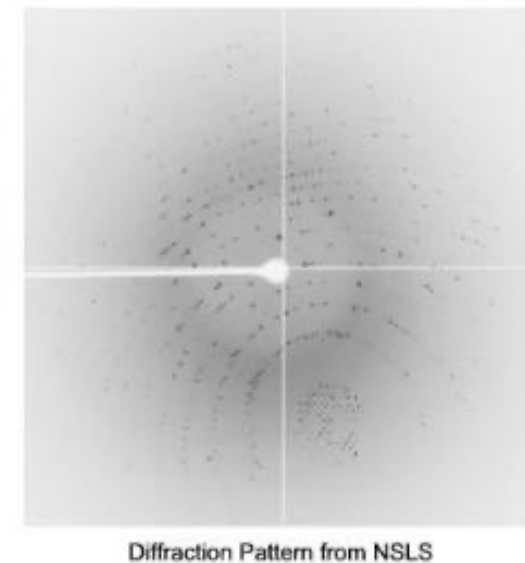
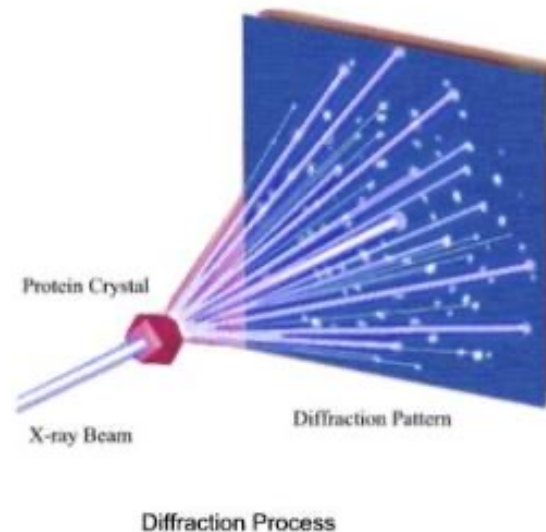


Laue conditions

- There will be a maximum of intensity when the three «Laue conditions» are satisfied

$$\vec{q} \cdot \vec{a}_1 = h2\pi \quad \vec{q} \cdot \vec{a}_2 = k2\pi \quad \vec{q} \cdot \vec{a}_3 = \ell2\pi; \quad h, k, \ell = 0, 1, 2, \dots$$

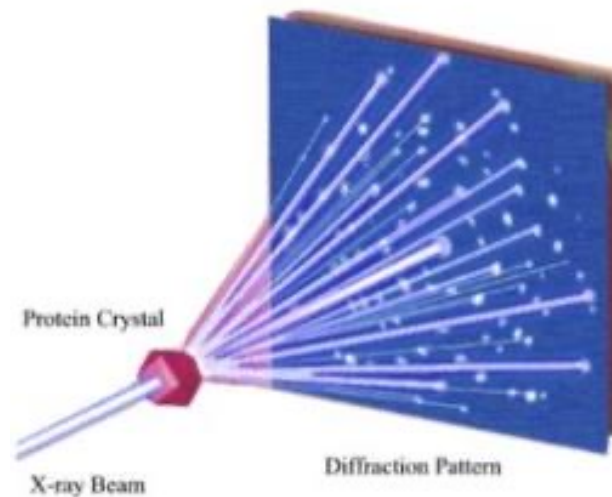
- These are
diffraction peaks
or reflections



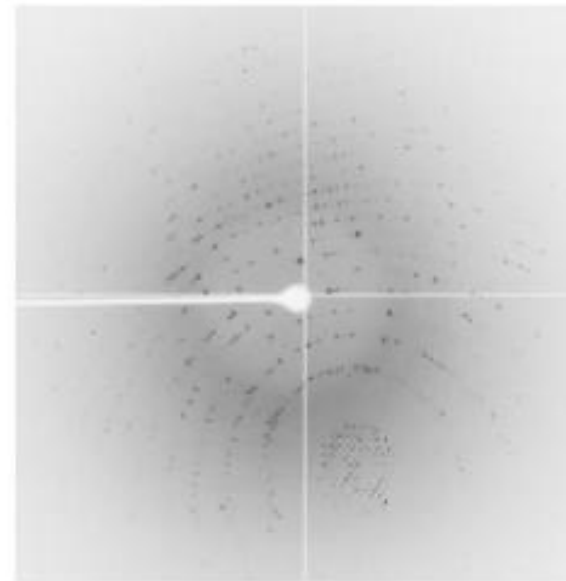
Laue conditions

- Each peak is identified by three integers ($h\ k\ \ell$)

$$\vec{q} \cdot \vec{a}_1 = h2\pi \quad \vec{q} \cdot \vec{a}_2 = k2\pi \quad \vec{q} \cdot \vec{a}_3 = l2\pi \quad h, k, \ell = 0, 1, 2, \dots$$



Diffraction Process



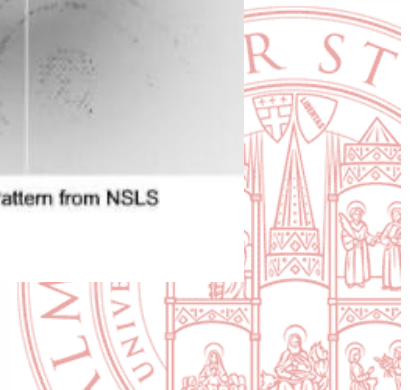
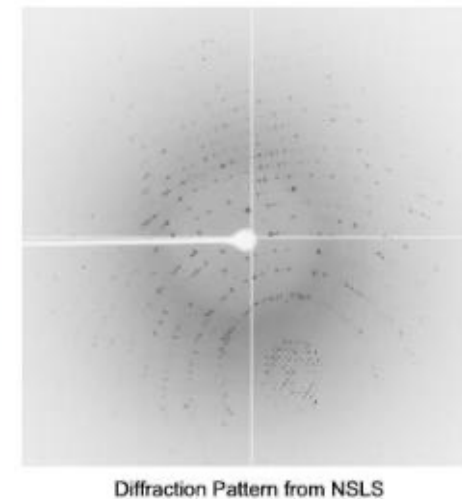
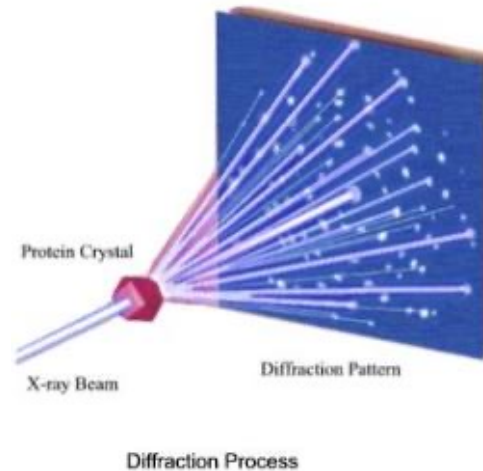
Diffraction Pattern from NSLS



Laue conditions

$$\vec{q} \cdot \vec{a}_1 = h2\pi \quad \vec{q} \cdot \vec{a}_2 = k2\pi \quad \vec{q} \cdot \vec{a}_3 = \ell 2\pi; \quad h, k, \ell = 0, 1, 2, \dots$$

- The position of peaks depends on the three primitive lattice vectors
- By measuring the position of the reflections it is possible to determine the type of lattice, the lattice parameters and angles



Laue conditions

$$\vec{q} \cdot \vec{a}_1 = h2\pi \quad \vec{q} \cdot \vec{a}_2 = k2\pi \quad \vec{q} \cdot \vec{a}_3 = \ell 2\pi; \quad h, k, \ell = 0, 1, 2, \dots$$

- Also reciprocal lattice vectors satisfy the same conditions, so we may also write

$$\vec{q} = \vec{G}_{h,k,\ell} = h\vec{b}_1 + k\vec{b}_2 + \ell\vec{b}_3$$

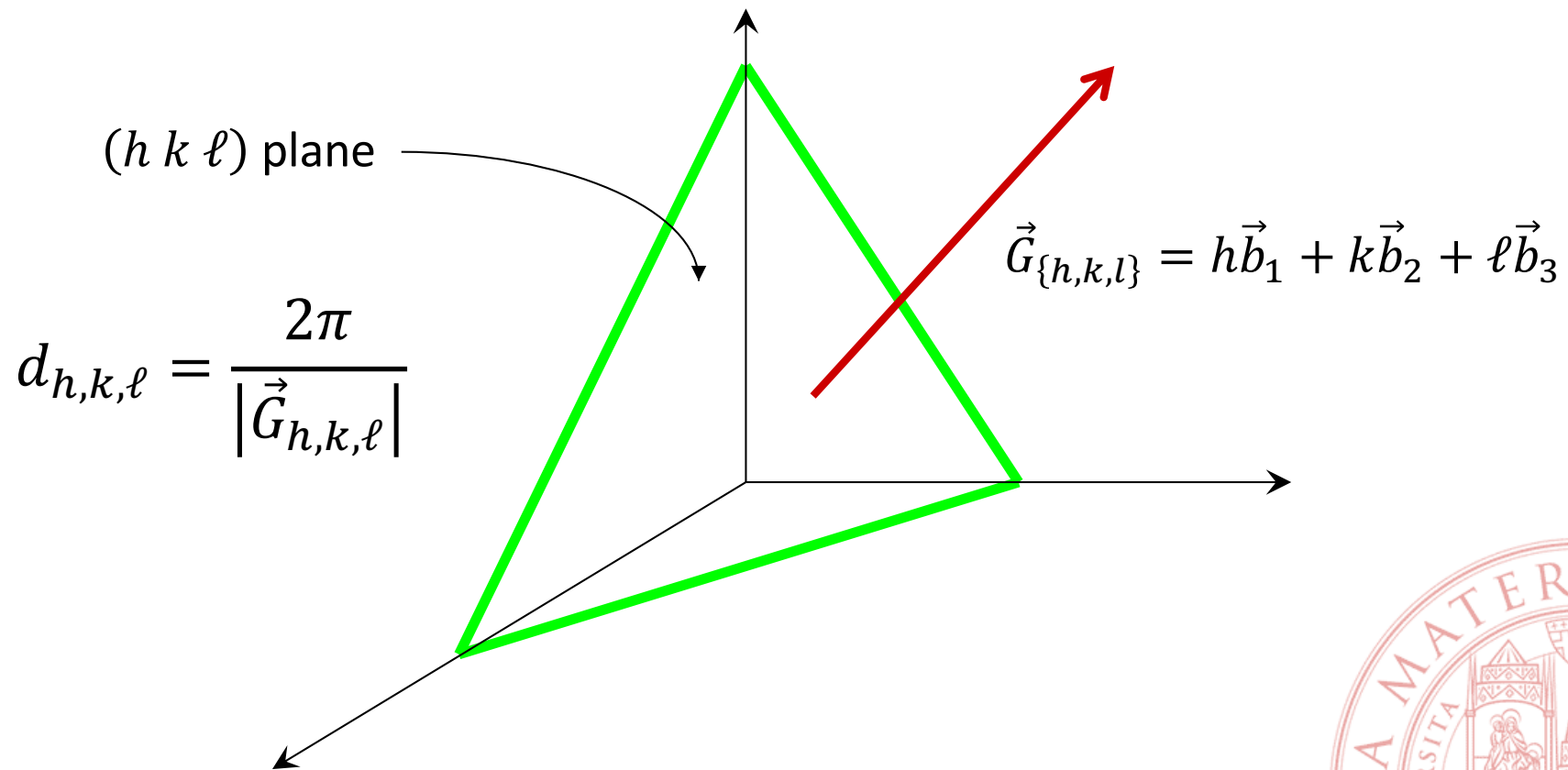
$$\vec{b}_i = 2\pi \frac{\vec{a}_j \times \vec{a}_k}{\vec{a}_i \cdot (\vec{a}_j \times \vec{a}_k)}$$

- A reciprocal lattice vector corresponds to each diffraction peak

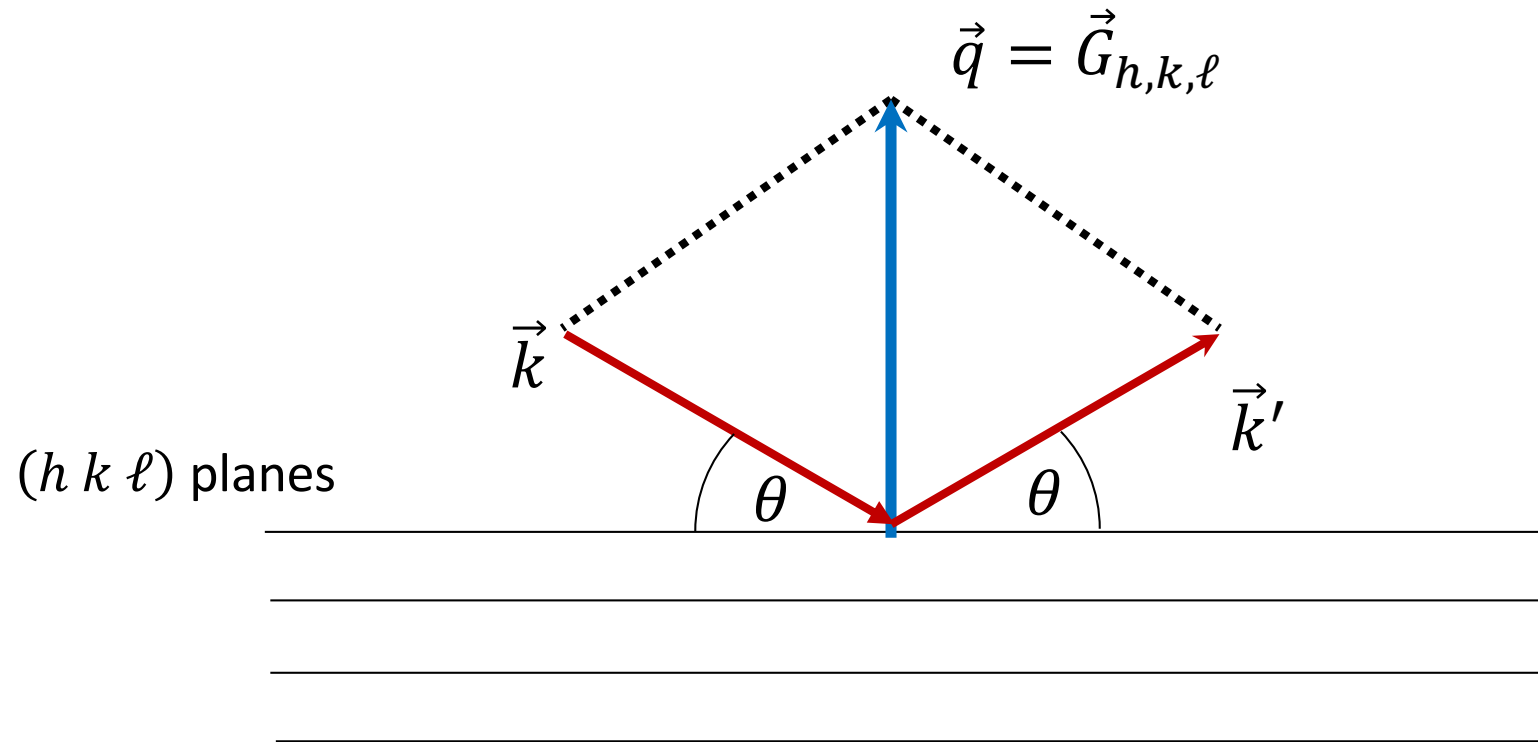


Lattice planes

- The reciprocal lattice vector $(h \ k \ \ell)$ is perpendicular to the lattice plane the Miller indices of which are $(h \ k \ \ell)$
- The lattice plane distance is inversely proportional to the modulus of the reciprocal lattice vector



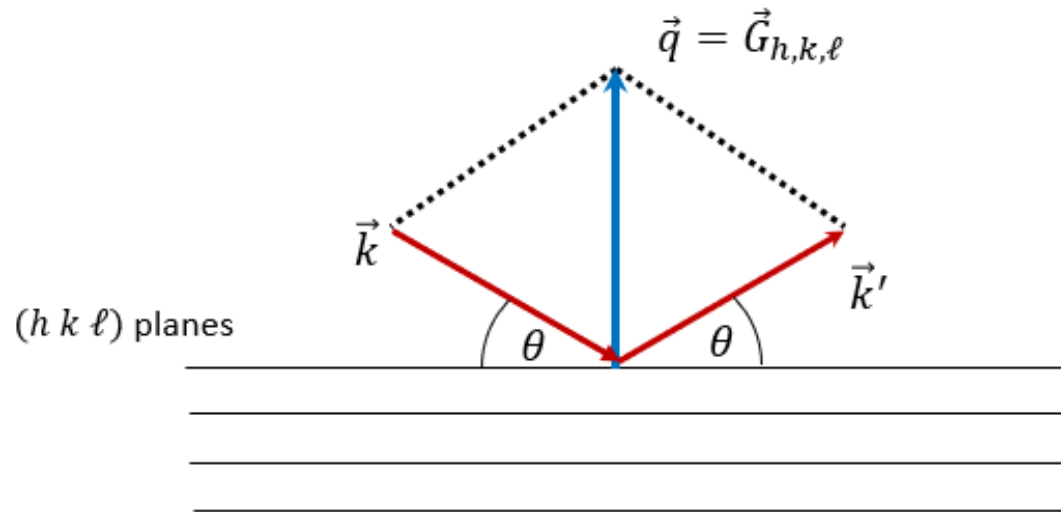
“Reflections” from lattice planes



- Analogy with reflection from a surface



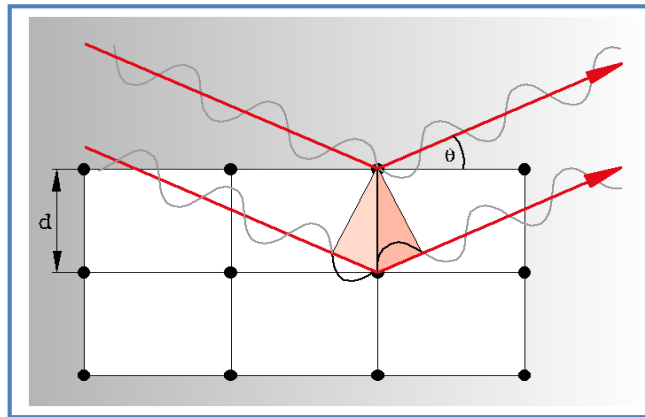
Braggs' Law



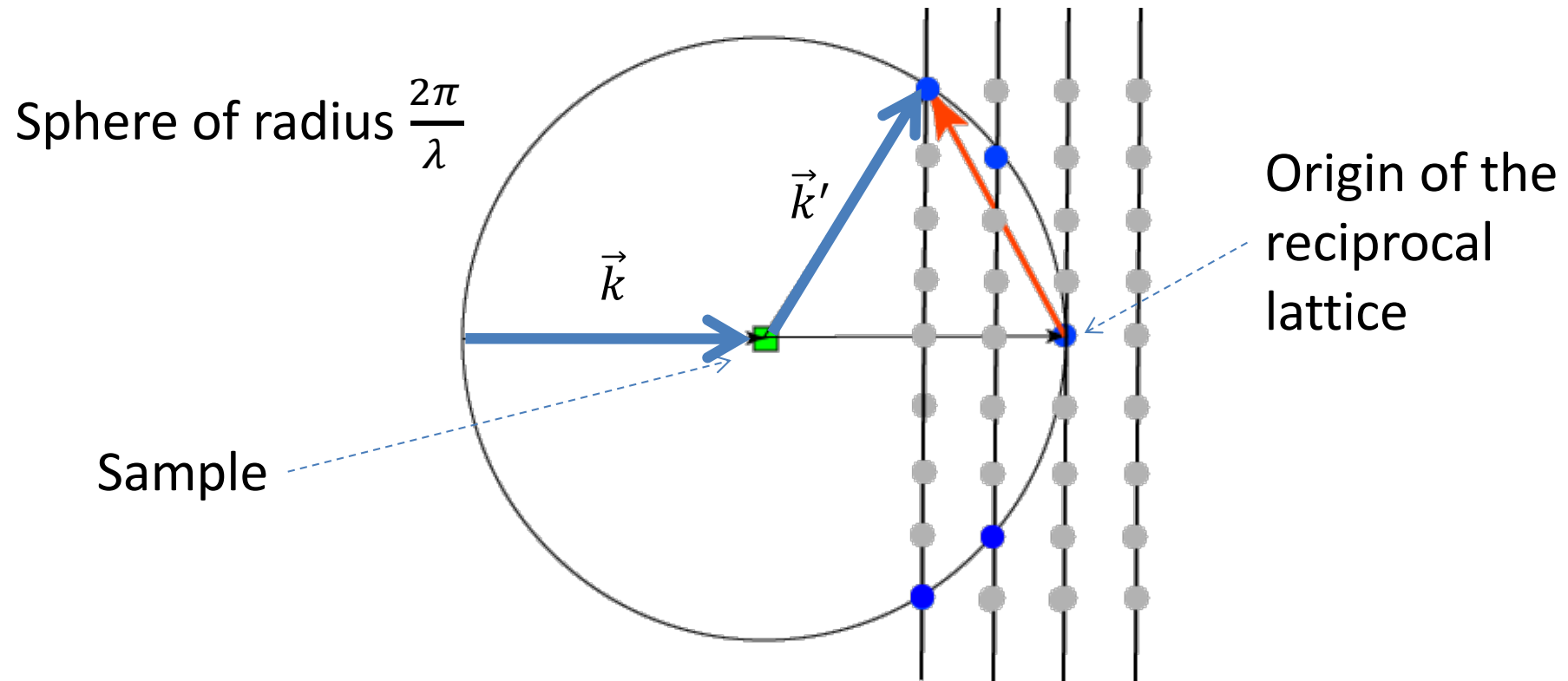
$$|\vec{q}| = |\vec{G}_{h,k,\ell}|$$

$$\frac{4\pi}{\lambda} \sin \theta = \frac{2\pi}{d_{h,k,\ell}}$$

$$\lambda = 2d_{h,k,\ell} \sin \theta$$

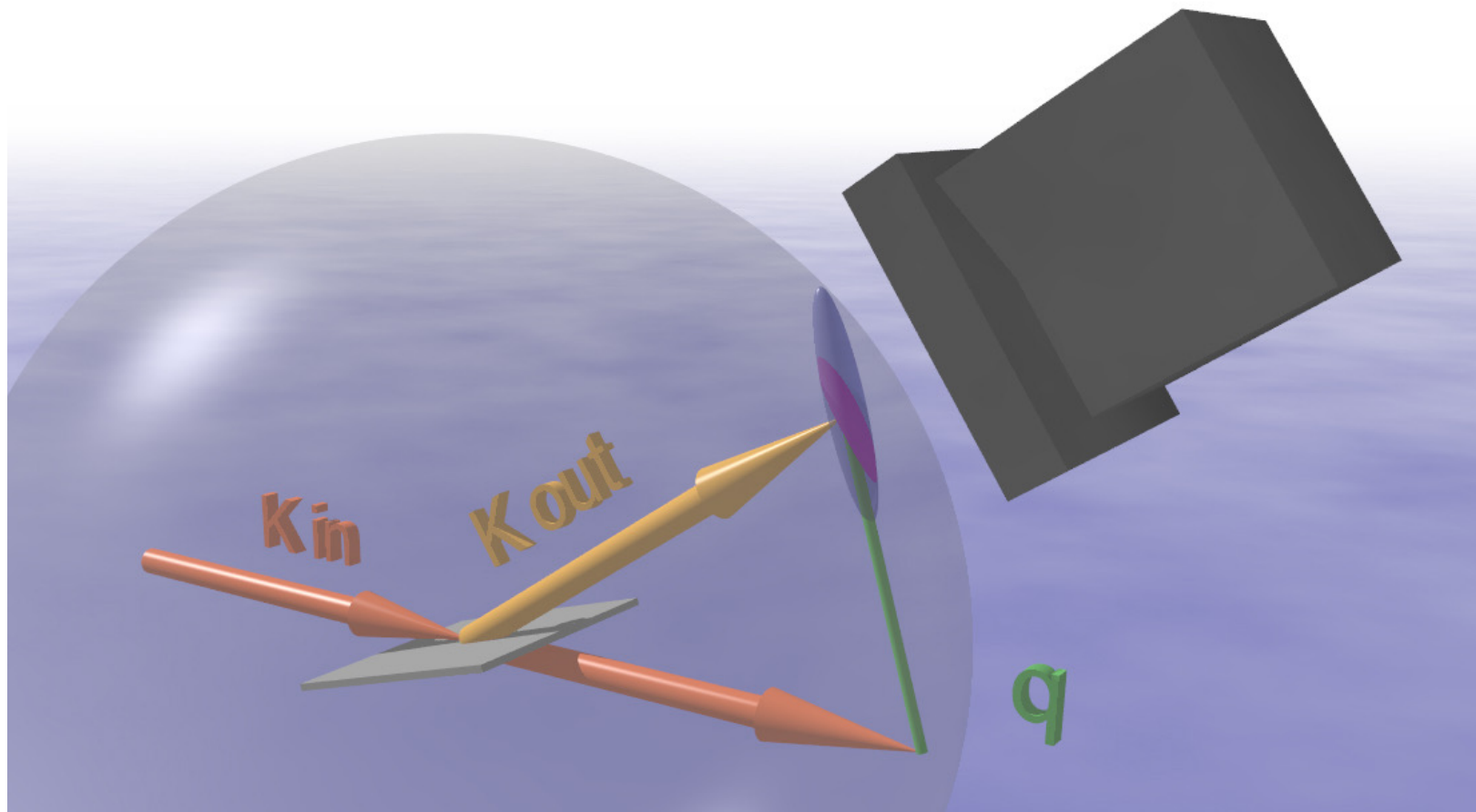


Ewald's sphere



- When the sphere intersects a reciprocal lattice point a reflection will be observed

Ewald's sphere



Intensity of reflections

- When Laue conditions are satisfied the intensity is proportional to

$$I(\vec{q}) \propto |F(\vec{q})|^2, \quad F(\vec{q}) = \sum_{j=1}^Q f_j(q) e^{-i\vec{q} \cdot \vec{r}_j}$$

- Since $\vec{q} = \vec{G}_{hk\ell}$
the notation used is

$$I_{hk\ell} \propto |F_{hk\ell}|^2, \quad F_{hk\ell} = \sum_{j=1}^Q f_j(G_{hk\ell}) e^{-i\vec{G}_{hk\ell} \cdot \vec{r}_j}$$

$$\vec{G}_{hk\ell} = h\vec{b}_1 + k\vec{b}_2 + \ell\vec{b}_3$$



Intensity of reflections

- It is possible that $F_{hkl} = 0$, in which case the diffraction peak/ reflection is absent
- Laue conditions are only necessary, they are not sufficient to observe a reflection
- When $F_{hkl} = 0$ for a set of $(h \ k \ \ell)$ the term “systematic extinctions” is used
- Simple rules can be derived which identify the type of lattice. E.g. $F_{hkl} \neq 0$ if:
 - FCC e ZB: $h \ k \ \ell$ same parity
 - BCC: $h + k + \ell = \text{an odd number}$



Intensity of reflections

- F_{hkl} depends on the atomic positions within the unit cell: by measuring the intensity of a good number of reflections it is possible to determine them

