

An Introduction to X-ray Methods



Sakura Pascarelli

sakura.pascarelli@xfel.eu

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Società
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Sincrotrone



Elettra Sincrotrone Trieste



Outline

■ Radiation – matter interaction

■ Experimental Techniques

■ Based on Scattering

- ▶ Elastic Scattering
- ▶ Inelastic Scattering

■ Based on Photoelectric Absorption

- ▶ Core-hole Spectroscopies
 - X-ray Absorption Spectroscopy
 - X-ray Emission Spectroscopy, RIXS, X-ray Raman
 - X-ray Polarization Dependent Spectroscopies
 - X-ray Photoemission Spectroscopy

■ Imaging Methods

■ Pause : 11h40-11h50

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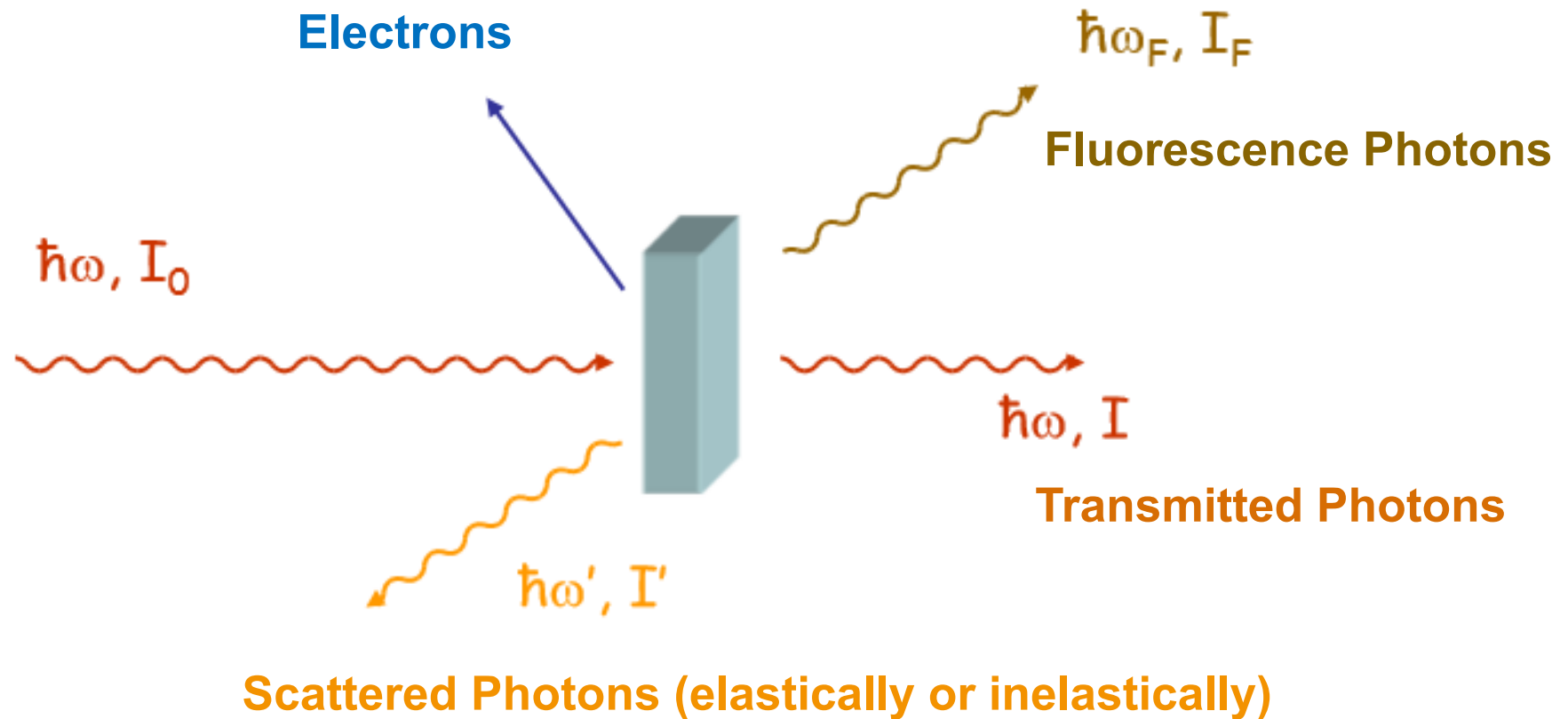
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X-rays-matter interaction

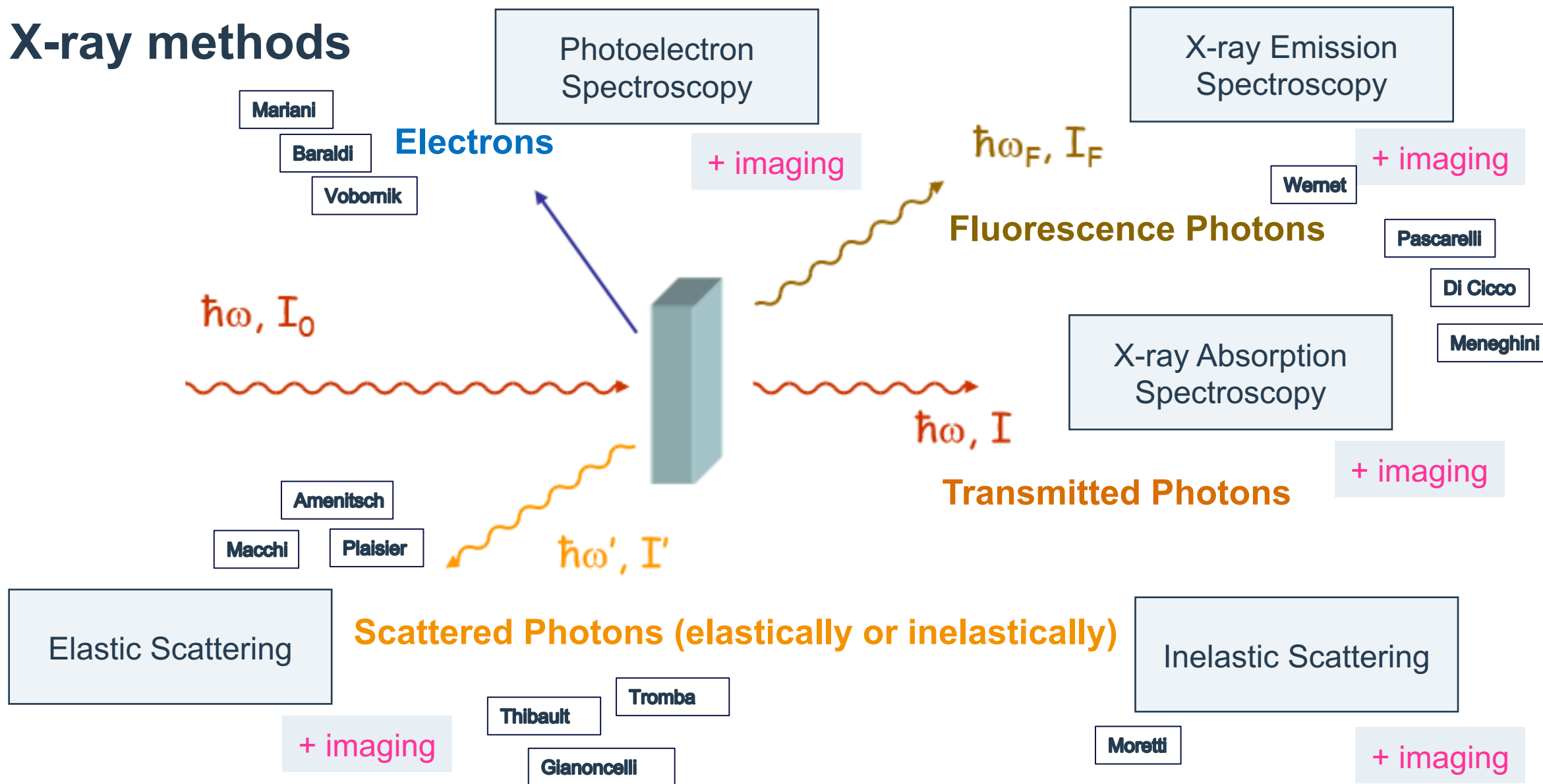


X-rays-matter interaction

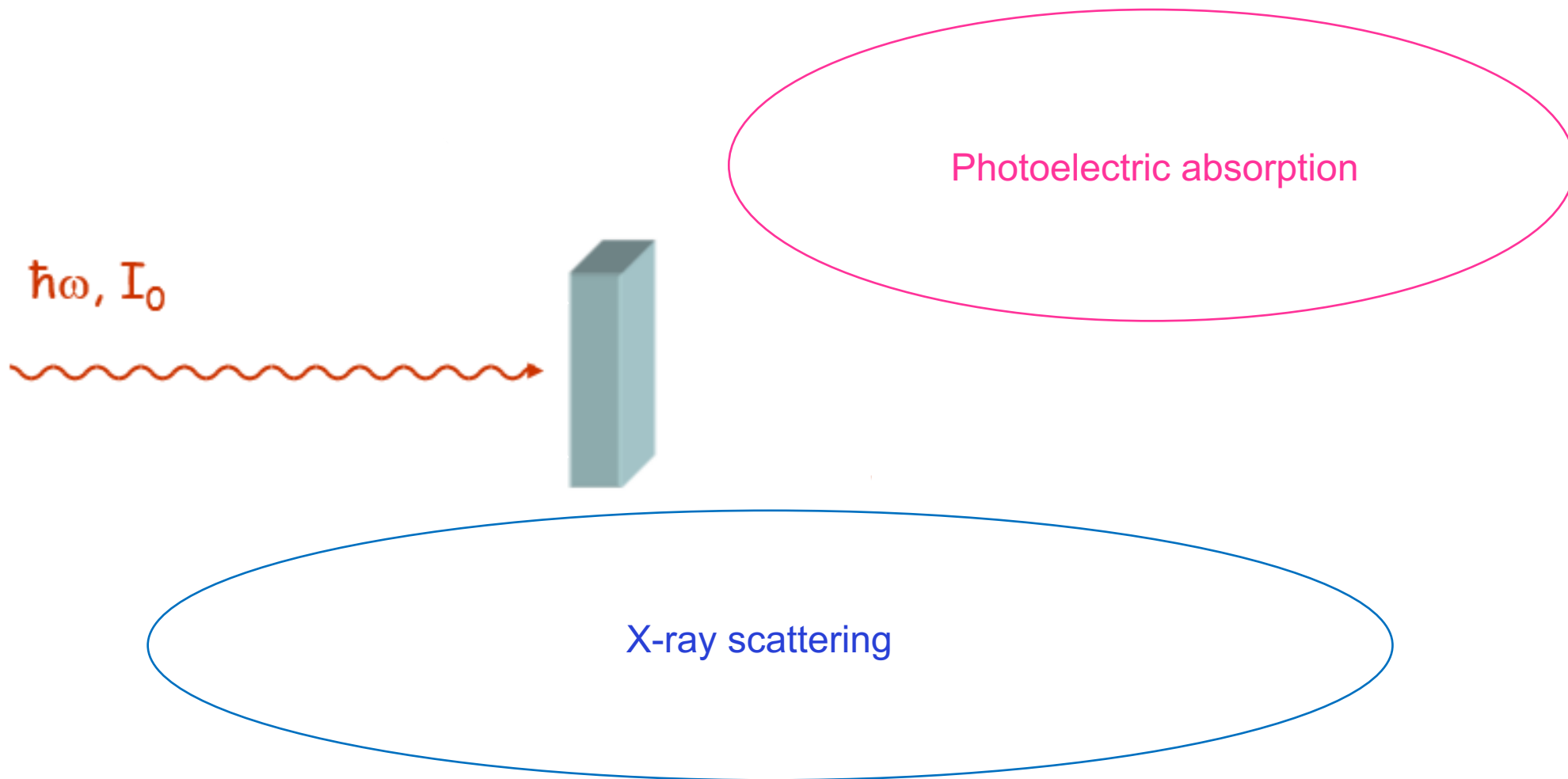


Boscherini 22 September

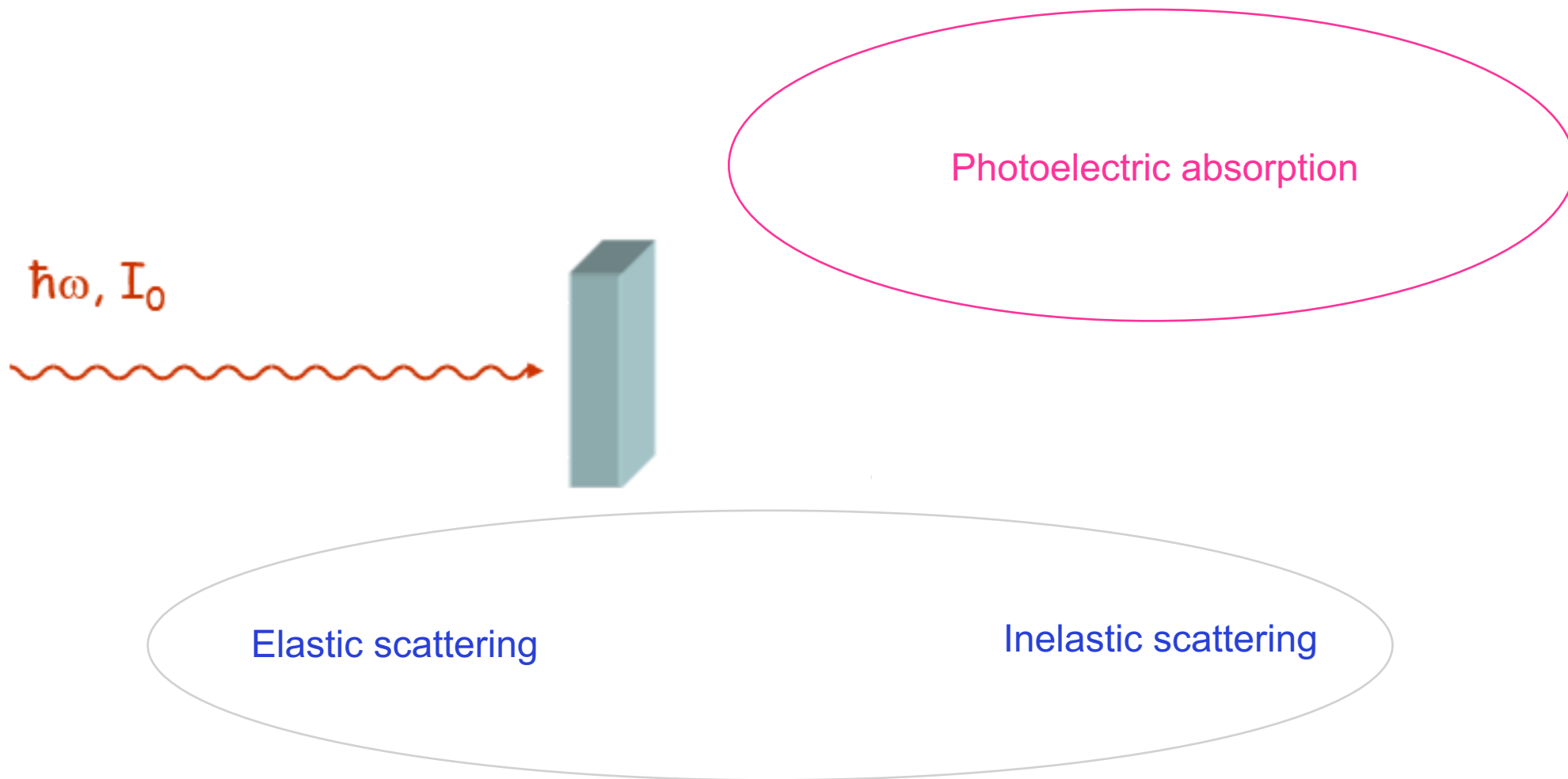
Main X-ray methods



Two fundamental X-ray matter interactions



Two fundamental X-ray matter interactions



Two families of experimental techniques



Elastic Scattering

Elastic scattering

Photoelectric absorption

Spectroscopy

Inelastic scattering

Two families of experimental techniques



Elastic Scattering

no exchange of energy

microscopic geometric structure

- diffraction: crystals (XRD, GIXRD, ...)
- scattering: amorphous, liquids (XRS, WAXS, SAXS)

Spectroscopy

exchange of energy

electronic structure

local structure

dynamics

- absorption (XAS, EXAFS, XANES, ..)
- emission (XES, HERFD, ..)
- photoelectron (PES, XPEEM, ARPES)
- inel scattering (IXS, RIXS, X-Raman)

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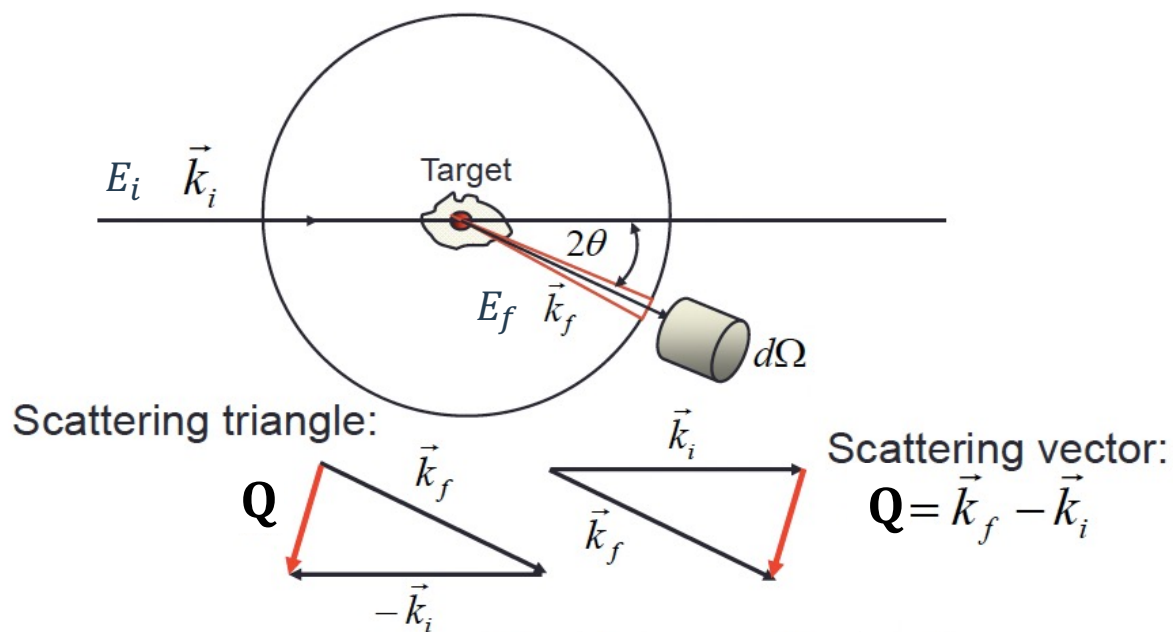
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X-ray Scattering: Momentum and Energy Transfer

- X-rays interact primarily with the **electron density**
- Scattering is described by conservation of **momentum** and **energy**
- X-ray scattering measures the system response as a function of **momentum transfer Q** and **energy transfer $\hbar\omega$** .



Incident wavevector: $\mathbf{k}_i = \frac{2\pi}{\lambda} \hat{\mathbf{k}}_i$

Scattered wavevector: $\mathbf{k}_f = \frac{2\pi}{\lambda} \hat{\mathbf{k}}_f$

Momentum transfer $\boxed{\mathbf{Q} = \mathbf{k}_f - \mathbf{k}_i}$

Energy transfer $\boxed{\hbar\omega = E_i - E_f}$

Elastic X-ray Scattering: basics

■ The photon changes direction but **not energy** $E_i = E_f$, thus $|\vec{k}_i| = |\vec{k}_f| = k = 2\pi/\lambda$, $|\vec{Q}| = 2k \sin \theta$

■ Only momentum is transferred

■ Elastic scattering probes the **static electron density**

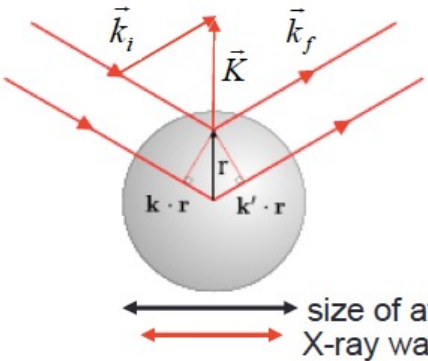
■ The quantity measured: $I(\vec{Q}) \propto |F(\vec{Q})|^2$ where

- $F(\vec{Q}) = \sum_j f_j(\vec{Q}) e^{i\vec{Q} \cdot \vec{r}_j}$ is the **crystal structure factor**

- $f_j(\vec{Q})$ is the **atomic form factor**



Reciprocal Space



Atomic form factor is the Fourier transform of the electron density distribution in an atom $\rho(r)$:

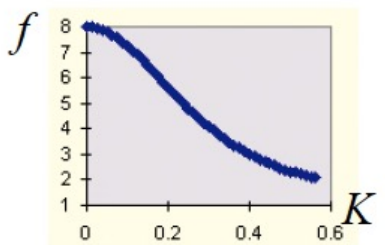
$$f(\vec{Q}) = \int \rho_e(\vec{r}) e^{i\vec{Q} \cdot \vec{r}} d\vec{r}$$

For spherical symmetric charge distribution follows:

$$f(Q) = \int \rho_e(r) e^{iQr \cos \varphi} 2\pi r^2 dr \sin \varphi d\varphi$$

$$= 4\pi \int_0^\infty \rho_e(r) r^2 \frac{\sin Qr}{Qr} dr$$

$$f(Q=0) = Z$$



Elastic X-ray Scattering from a perfect crystal: the Bragg condition

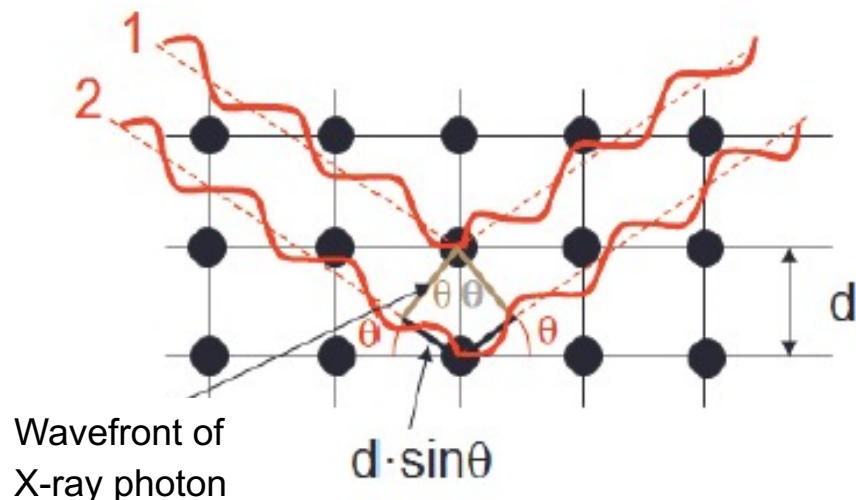
- Sharp peaks appear at θ that fulfill the Bragg condition
- The path difference between wavefront1 and 2: $\Delta r = 2d\sin\theta$
- The condition for constructive interference: $\Delta r = n\lambda$
- Different crystal planes d diffract to different angles θ

$$2d\sin\theta = n\lambda$$

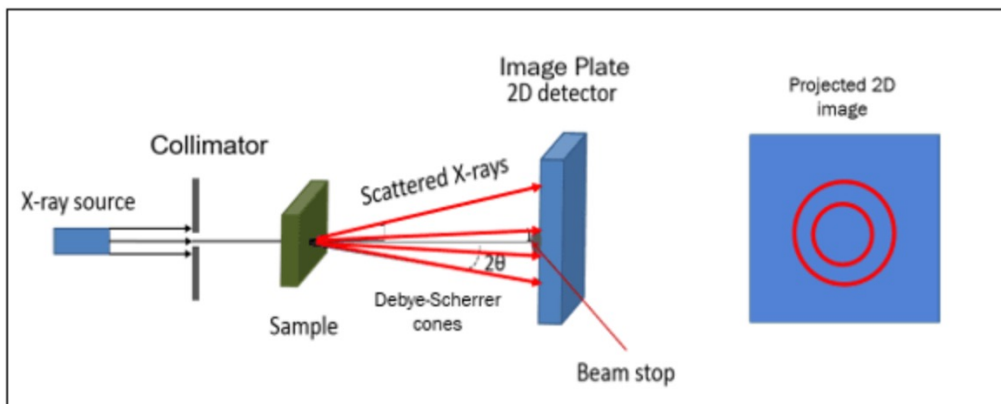
Bragg condition for interference maxima

$$Q = G_{hkl}$$

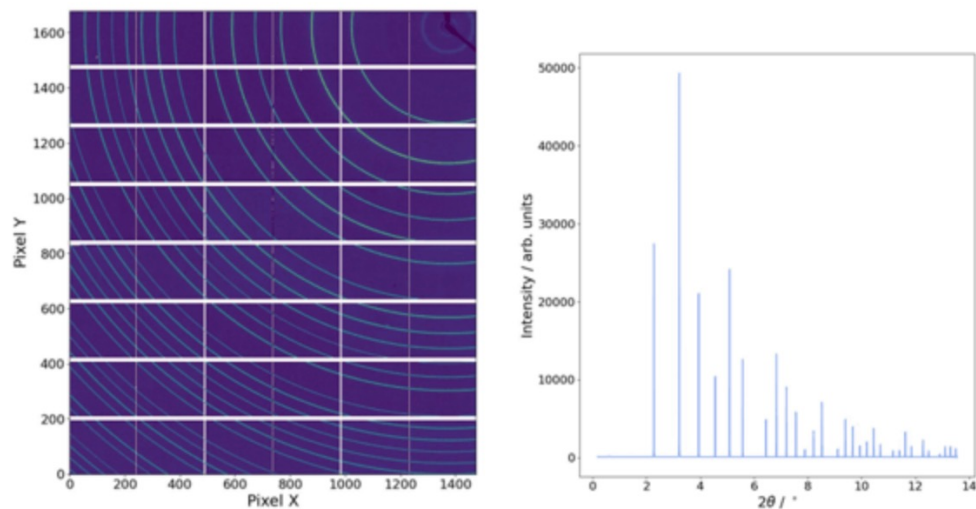
G_{hkl} = Reciprocal lattice vector



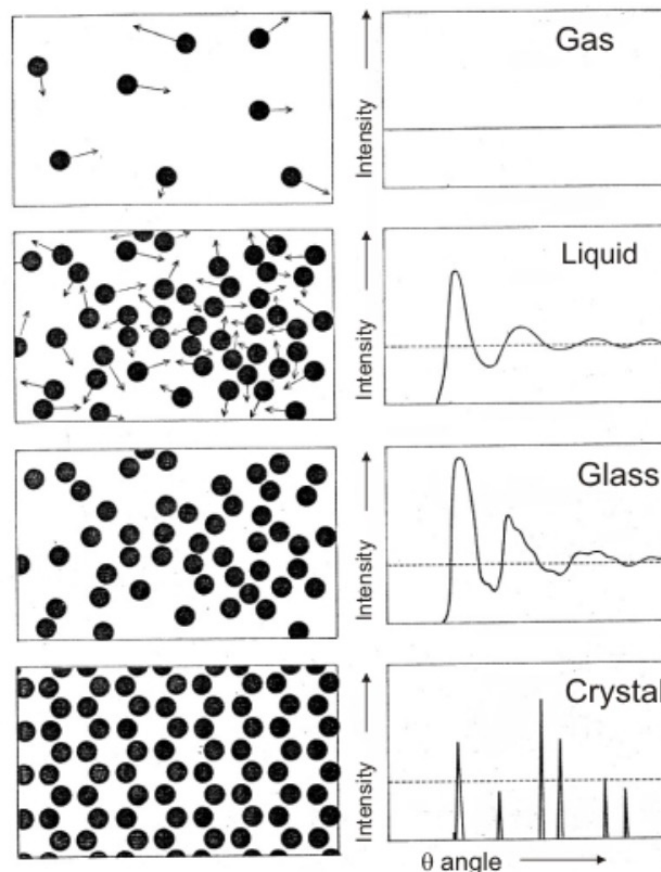
Elastic X-ray Scattering: measuring structure



(a)



lcmd.com



Reveals

- ❖ Crystal structure
- ❖ Lattice parameters
- ❖ Long-range order
- ❖ Defects
- ❖ Nanostructure

Typical techniques

- ❖ X-ray diffraction (XRD)
- ❖ Small-angle X-ray scattering (SAXS)
- ❖ Wide-angle X-ray scattering (WAXS)
- ❖ Coherent diffraction imaging (CDI)

Small Angle vs Wide Angle X-ray Scattering: Reciprocal Space and Length Scale

Relationship between scattering angle and length scale

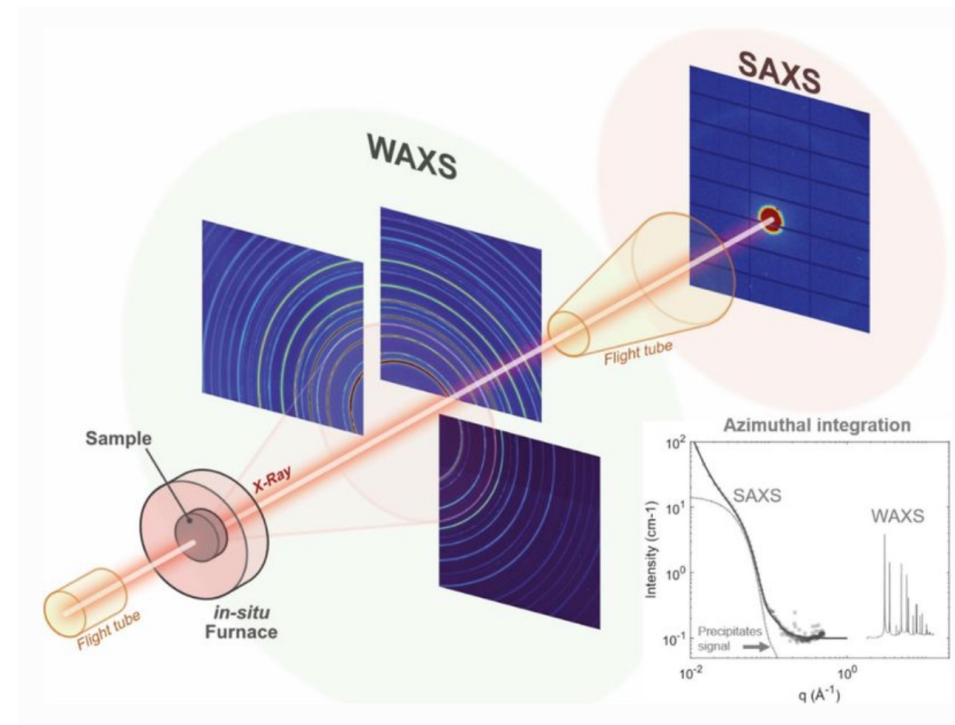
Momentum transfer: $q = \frac{4\pi}{\lambda} \sin \theta$

Characteristic length scale: $d \approx \frac{2\pi}{q}$

SAXS and WAXS probe different regions of reciprocal space, corresponding to different real-space length scales.

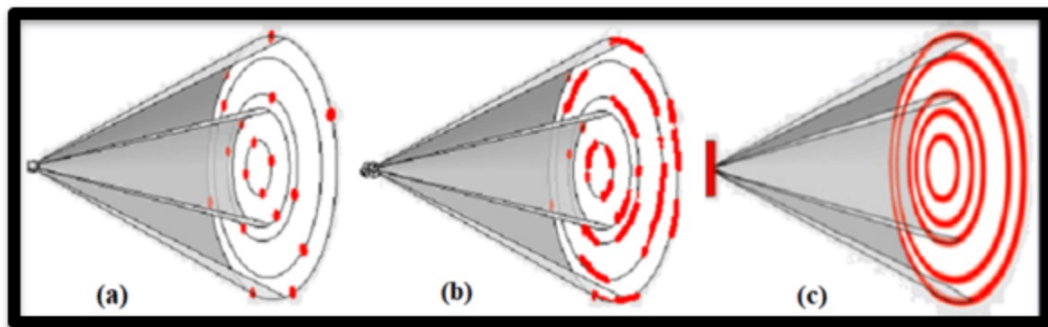
SAXS: long-range electron density fluctuations

WAXS: atomic correlations and crystal lattice order



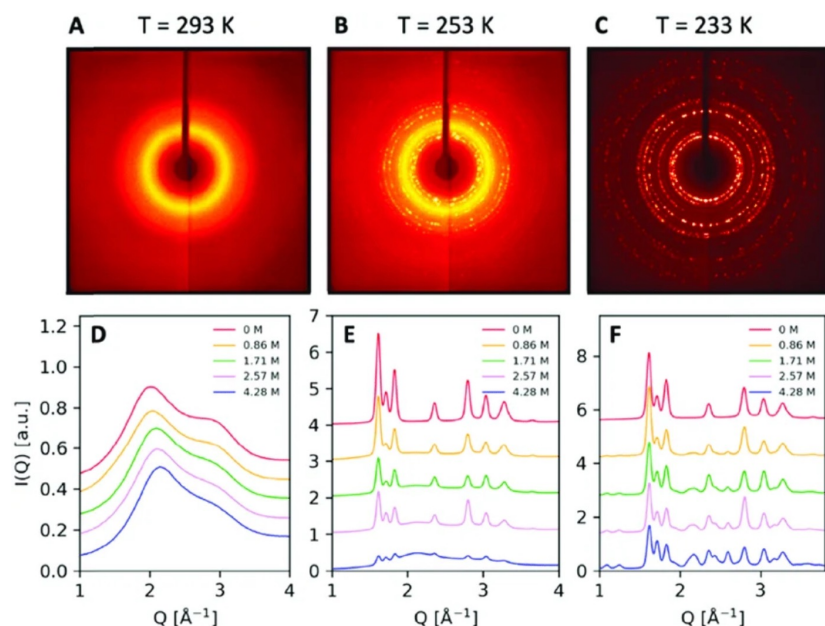
Schematic of the in situ simultaneous SAXS/WAXS measurement setup using a furnace sample environment at beamline P21.2, PETRA III.

Elastic X-ray Scattering: what do the patterns look like ?



Crystalline Materials (Long-Range Order)

- ❖ **Atomic Structure:** Highly ordered, repeating 3D lattice geometry.
- ❖ **Scattering Pattern:** Sharp, well-defined spots or sharp concentric rings (Bragg peaks).



Amorphous Materials & Liquids (Short-Range Order)

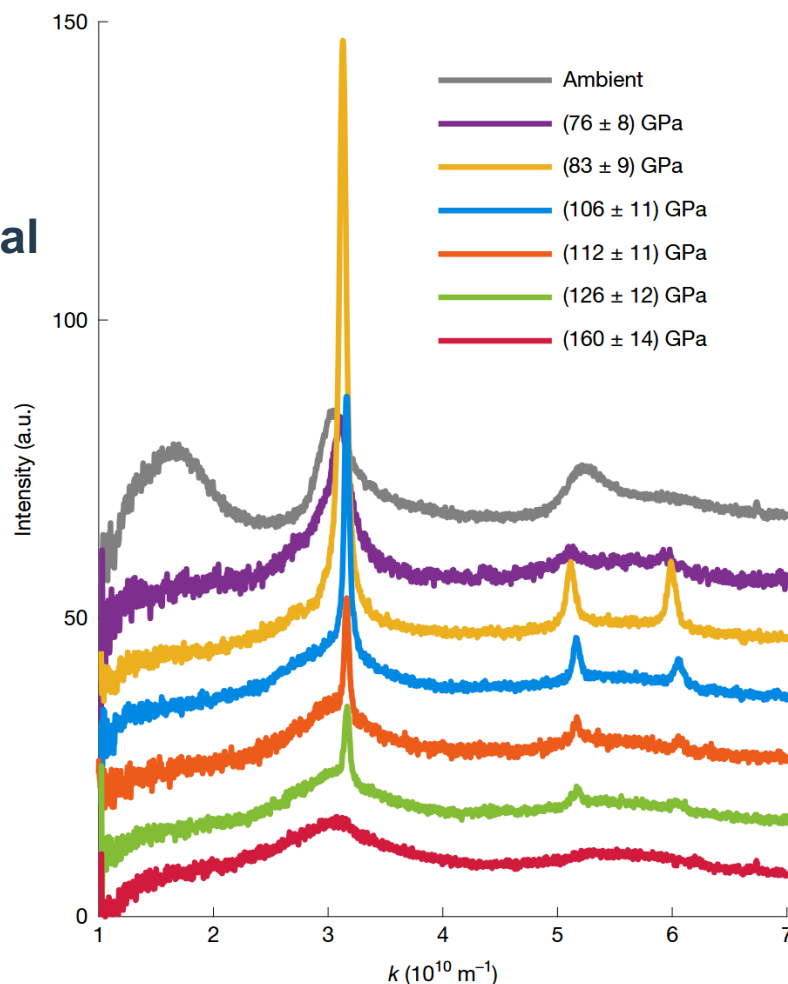
- ❖ **Atomic Structure:** Disordered, random arrangement with only nearest-neighbor consistency.
- ❖ **Scattering Pattern:** Broad, diffuse halos or a singular wide amorphous hump.

Local structure in liquid carbon

Towards better control of the synthesis of novel carbon-based materials

- First experimental $S(k)$ of liquid C
- Confirms tetrahedral coordination ($N \sim 4$)
- First experimental determination of local structure in low Z matter at extreme conditions**

D. Kraus et al., Nature 642, 351 (2025)



- glassy carbon: broad amorphous structures
- peaks appear: diamond formation
- complete disappearance of diamond, only melt visible

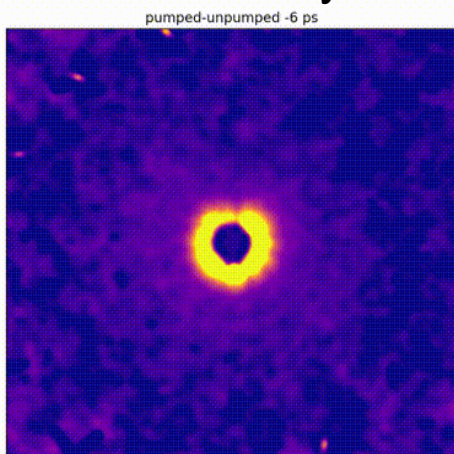
HED@EuXFEL X-ray scattering on ns-laser shocked targets

Magnetic fluctuations in ferromagnetic multilayers by ps-resolved-SAXS

Towards fast energy-efficient memory applications

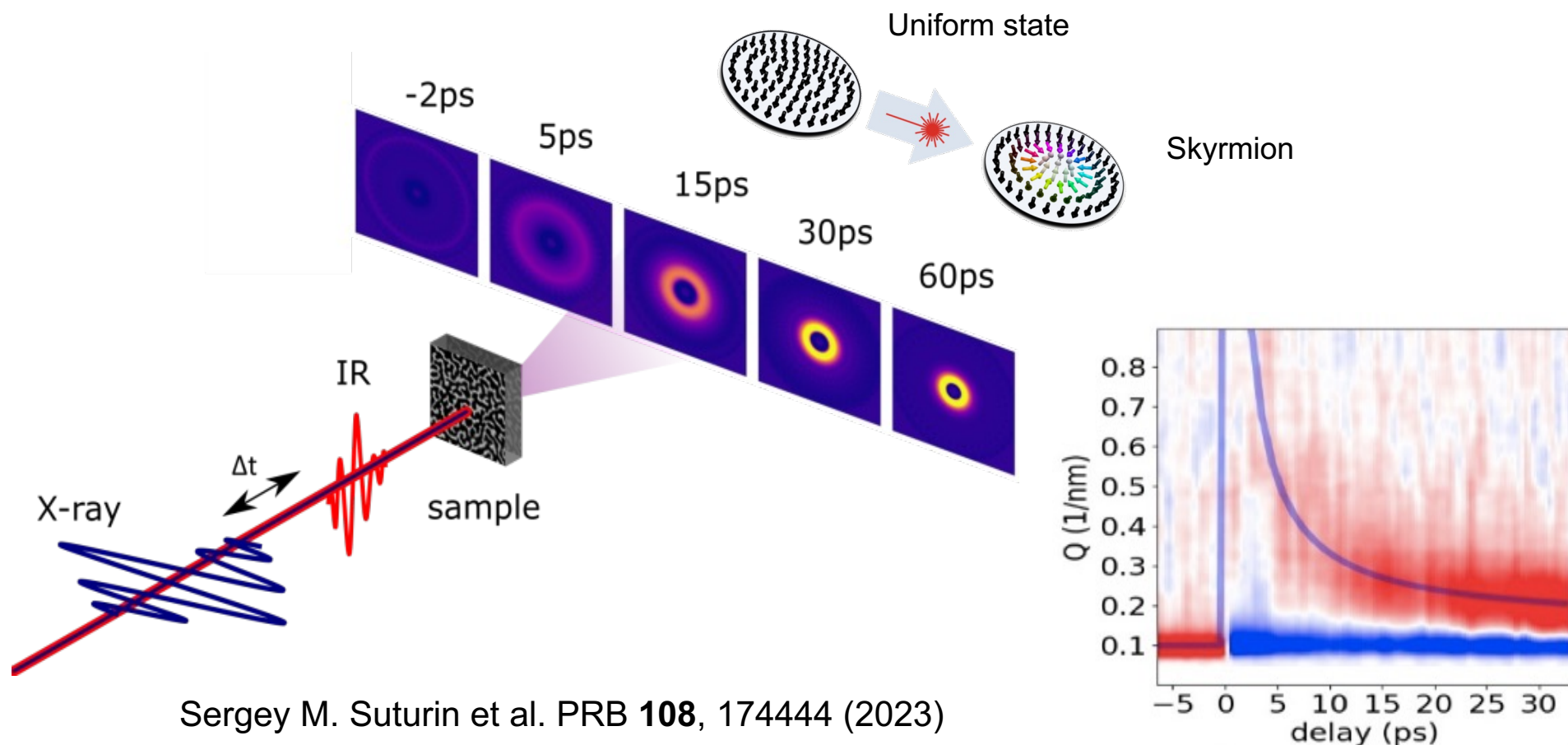
- Nanosize Co/Pt multilayers with perpendicular magnetization
- Probe **ultrafast nucleation and subsequent coalescence of magnetic fluctuations (skyrmions)**

SAXS vs delay time



■ SCS@XFEL ps-resolved resonant magnetic SAXS at Co L_3

■ European XFEL



Inelastic X-ray Scattering: basics

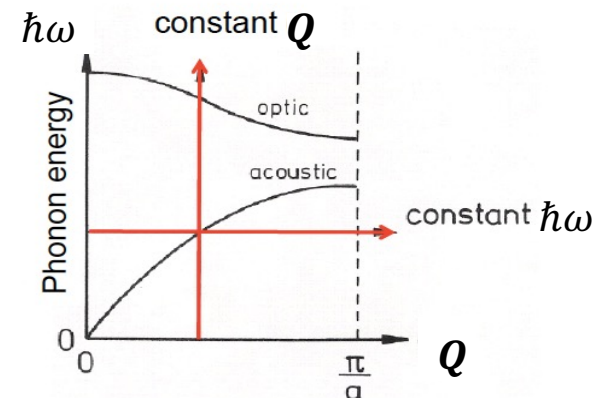
■ The photon exchanges energy with elementary excitations $E_i \neq E_f$.

■ Energy transfer : $\hbar\omega = E_i - E_f$

■ Momentum transfer : $\mathbf{Q} = \mathbf{k}_f - \mathbf{k}_i$

■ This energy exchange reveals collective excitations and dynamical correlations

■ The quantity measured is proportional to the **dynamic structure factor** $S(\mathbf{Q}, \omega)$ which contains information on the density correlations.



Elastic	Inelastic
$\Delta E = 0$	$\Delta E \neq 0$
Static structure	Dynamics
Electron density	Elementary excitations
Structure factor $F(\mathbf{Q})$	Dynamic structure factor $S(\mathbf{Q}, \omega)$

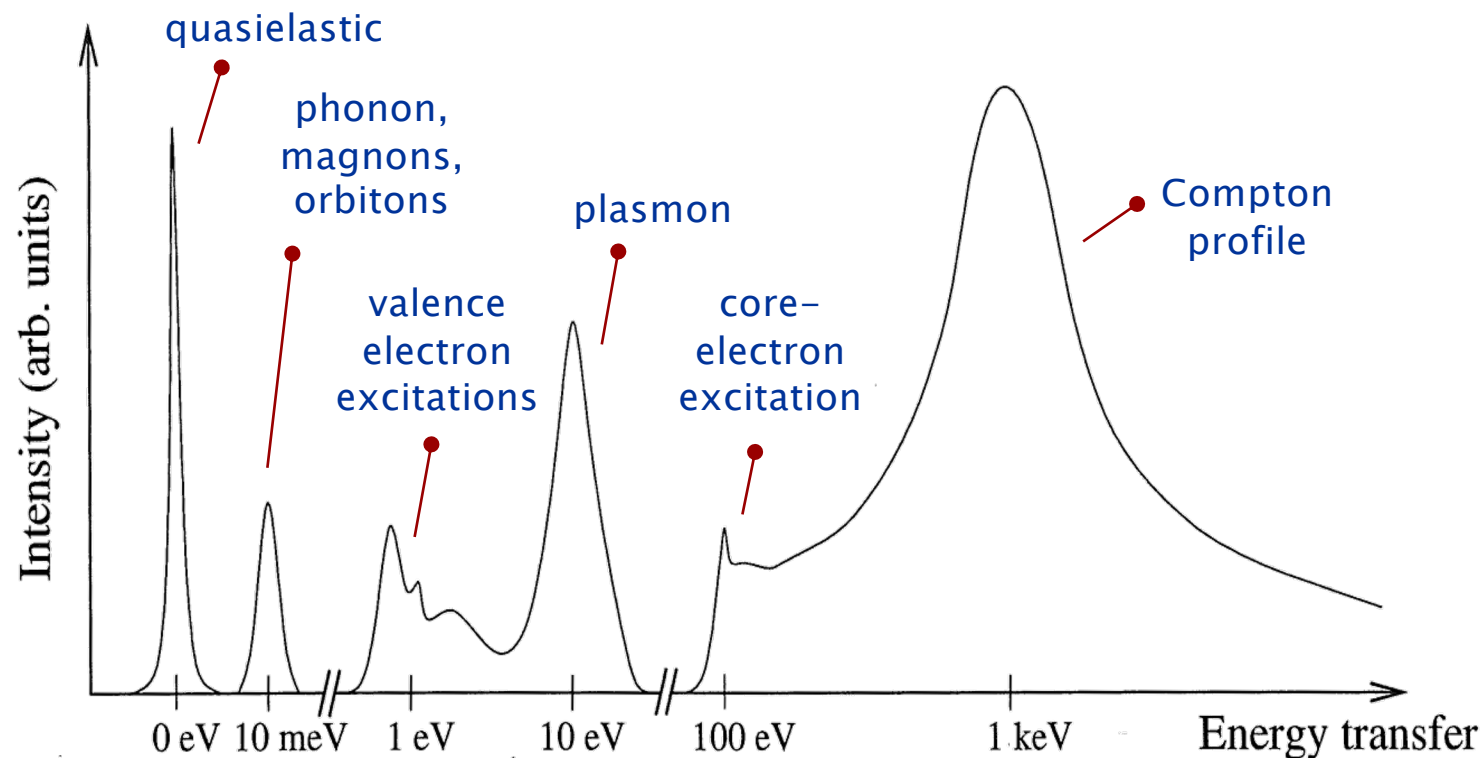
Inelastic X-ray Scattering: measuring excitations

Excitations observed

- Phonons
- Magnons
- Plasmons
- Crystal-field excitations
- Charge-transfer excitations
- Electronic band excitations

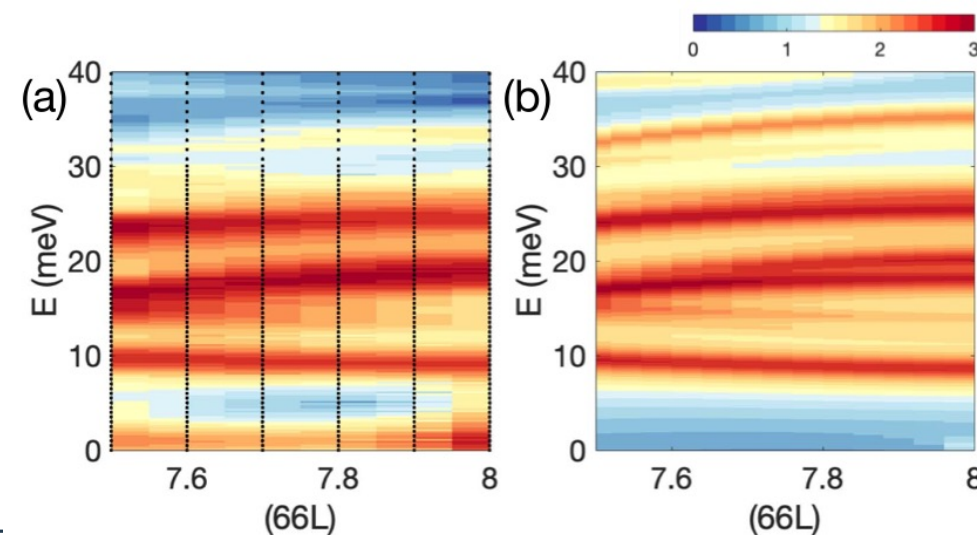
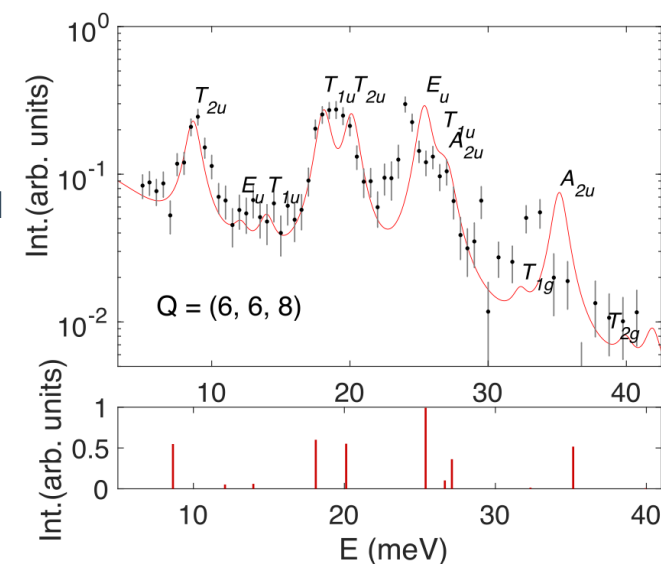
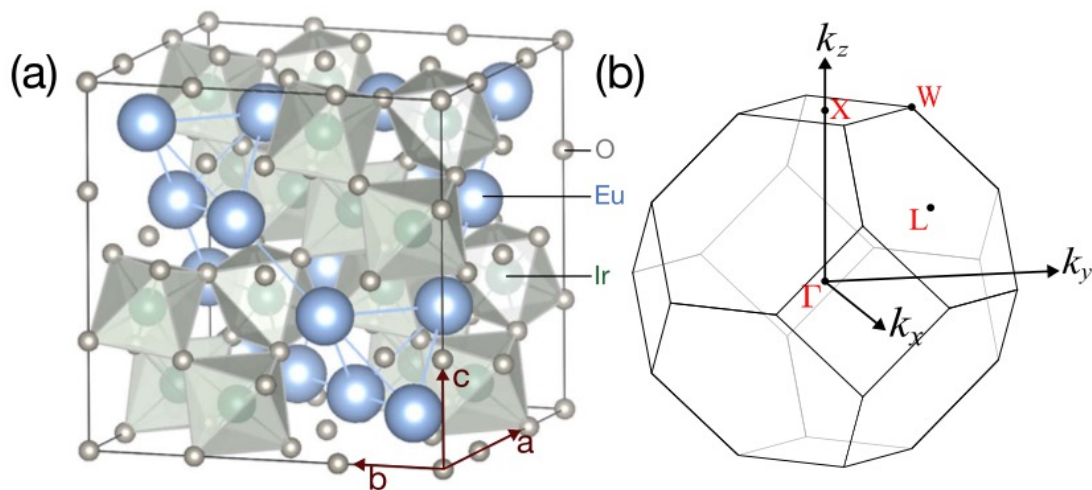
Experimental techniques

- Inelastic X-ray Scattering (IXS)
- Resonant Inelastic X-ray Scattering (RIXS)
- X-ray Raman Scattering (X-ray Raman)



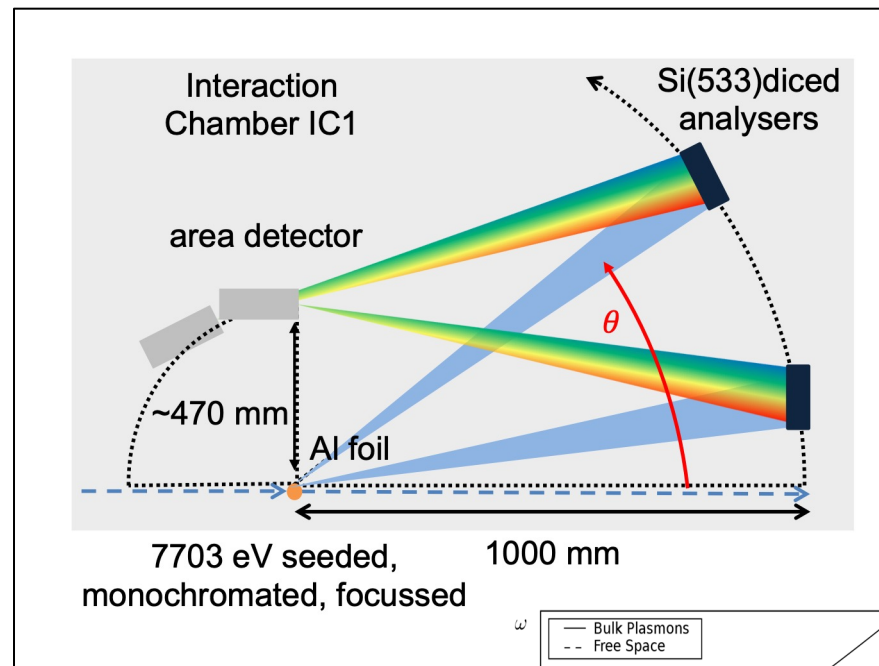
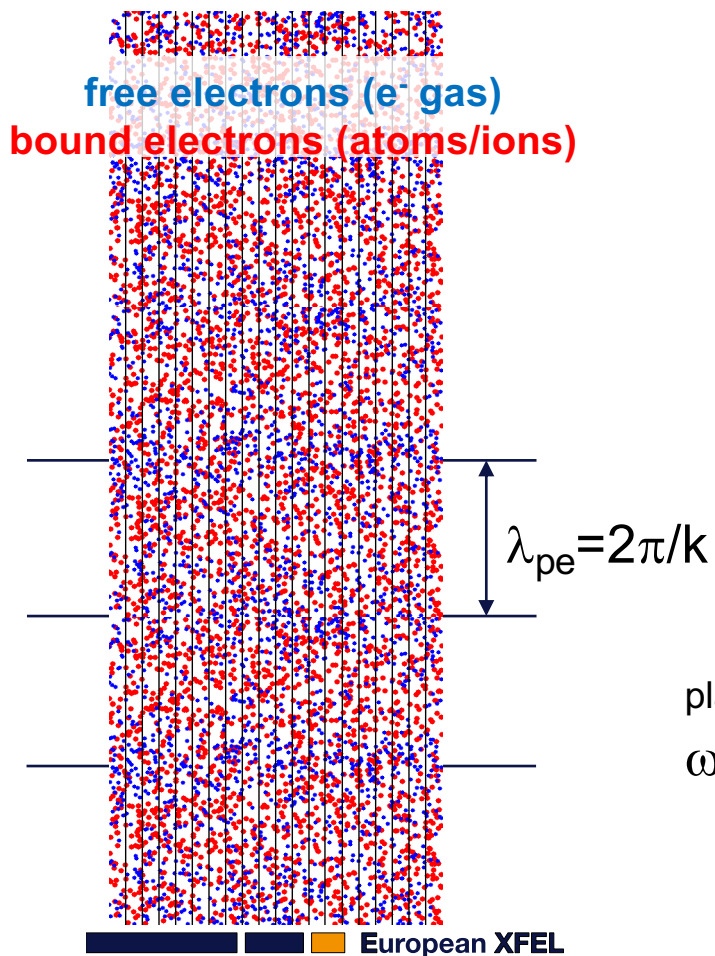
Phonon excitations in $\text{Eu}_2\text{Ir}_2\text{O}_7$ probed by inelastic x-ray scattering

- Phonon dynamics and its interplay with magnetic ordering is crucial for understanding the unique quantum phases in the pyrochlore iridates.
- 4d and 5d heavy transitional metal electronic systems → cutting-edge research domain in condensed matter physics.
- Unique amalgamation of extended 5d/4d orbitals, strong spin–orbit coupling and geometrically frustrated lattices, → rich platform to search for exotic correlated states.



Plasmon dispersion in shocked Aluminium

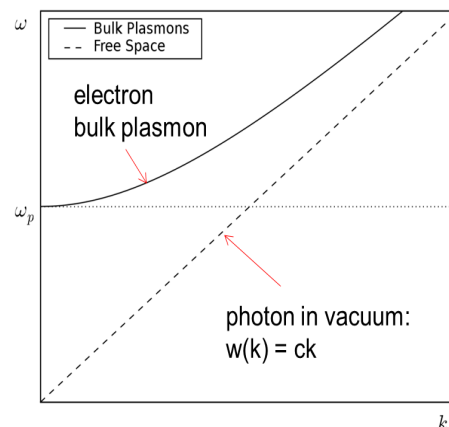
Achieving precision required to benchmark theoretical models



plasmon dispersion:

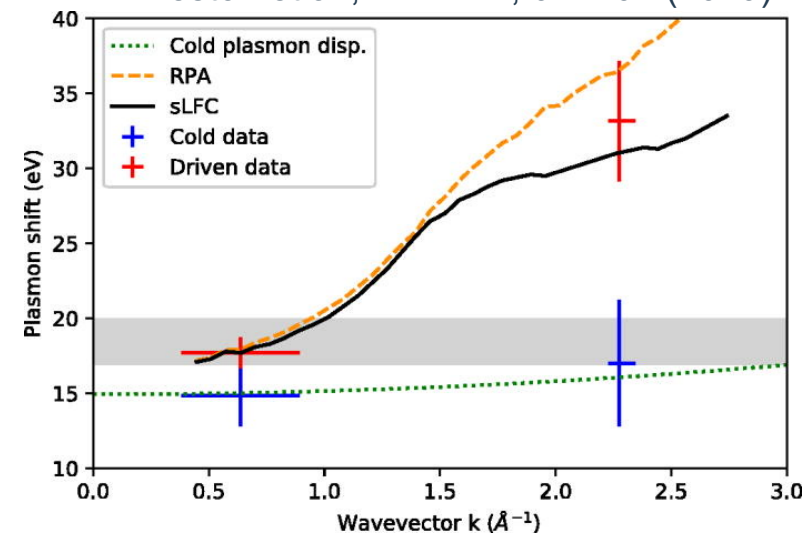
$$\omega_{\text{plasmon}}^2 = \omega_{pe}^2 + c^2 k^2$$

$$\omega_{pe} = \sqrt{\frac{e^2 n_e}{\epsilon_0 m_e}}$$

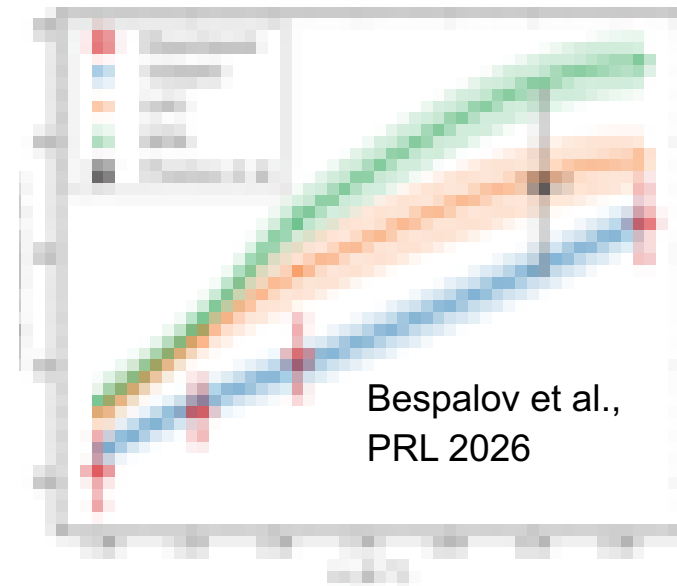


First measured (LCLS MEC, 2017)

Preston et al., *APL* 114, 014101 (2019)



Repetition (EuXFEL, HED-HIBEF, 2024)



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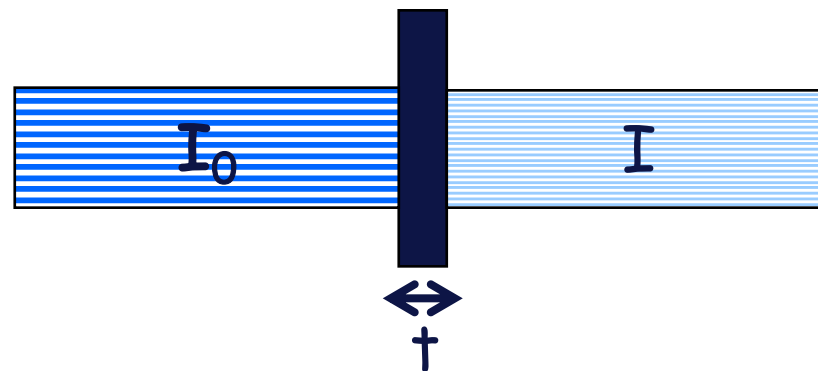
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The Absorption Coefficient μ



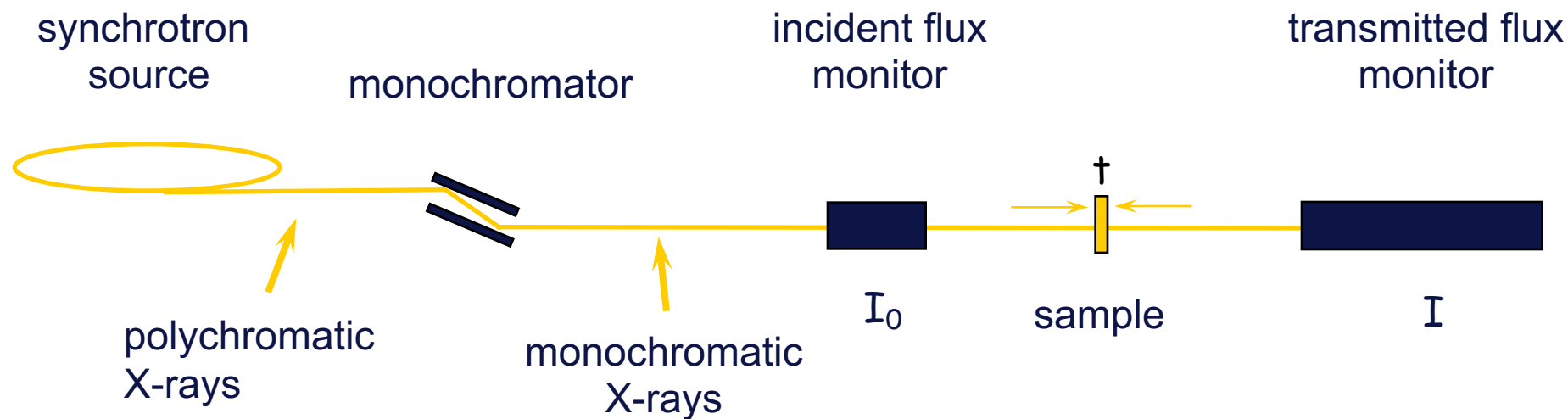
$$I = I_0 \exp[-\mu t]$$



linear absorption coefficient

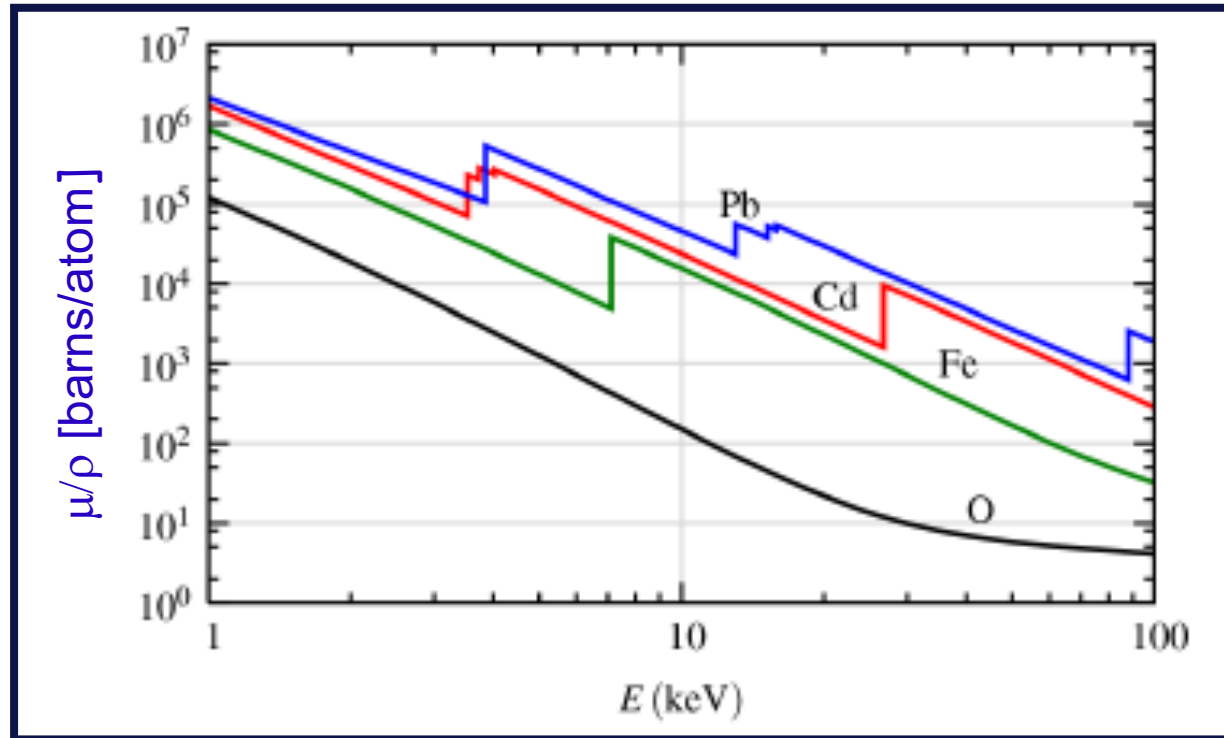
$$\mu t = \ln [I_0 / I]$$

The Absorption Coefficient μ



1. Measure I_0 and I as a function of E_x
2. Calculate: $\mu t = \ln [I_0/I]$

The Absorption Coefficient μ

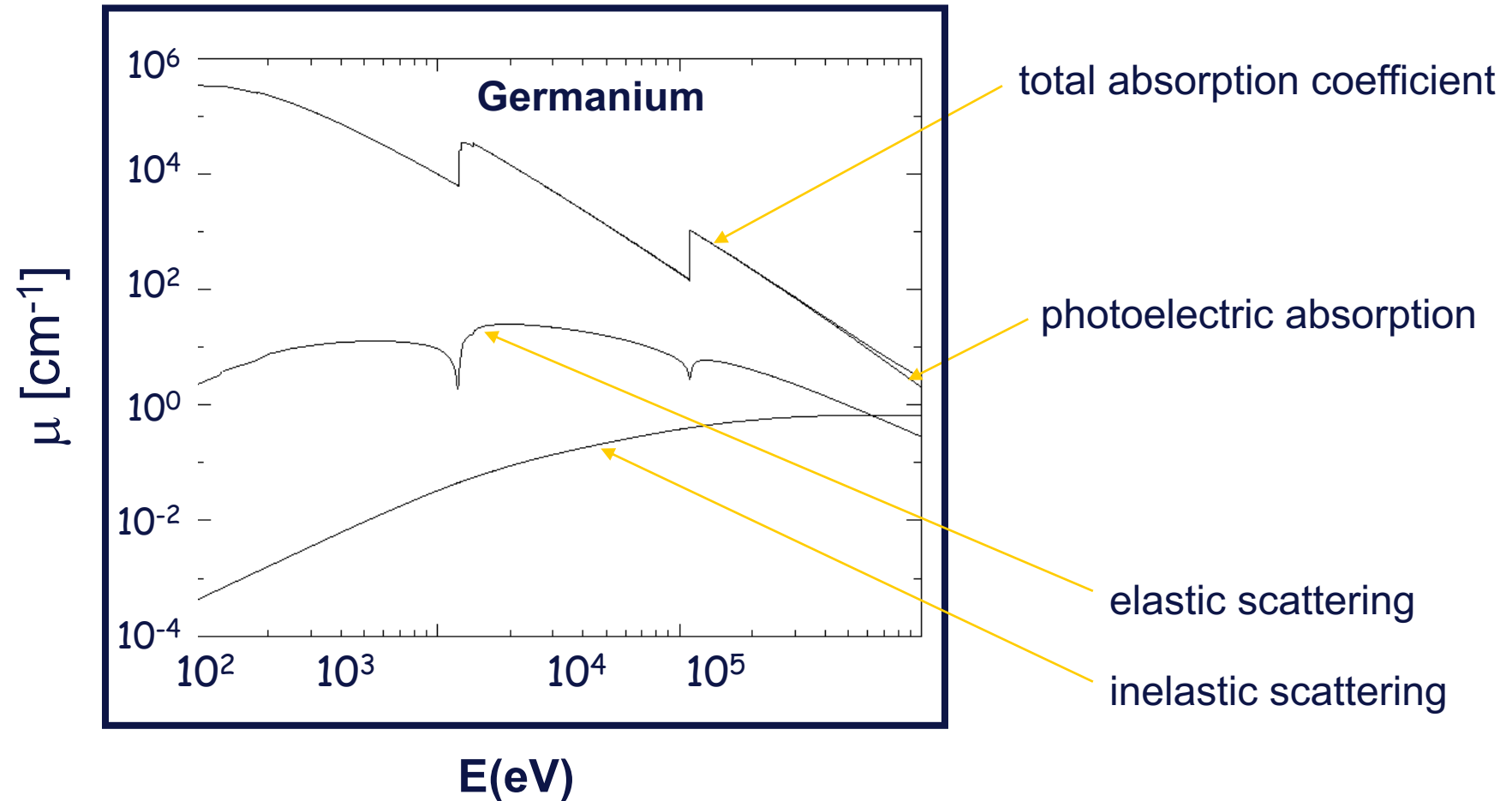


μ depends strongly on X-ray energy **E** and atomic number **Z**, the density ρ and atomic mass **A**

$$\mu \approx \frac{\rho Z^4}{A E^3}$$

μ has sudden jumps (**absorption edges**) which occur at energies characteristic of the element.

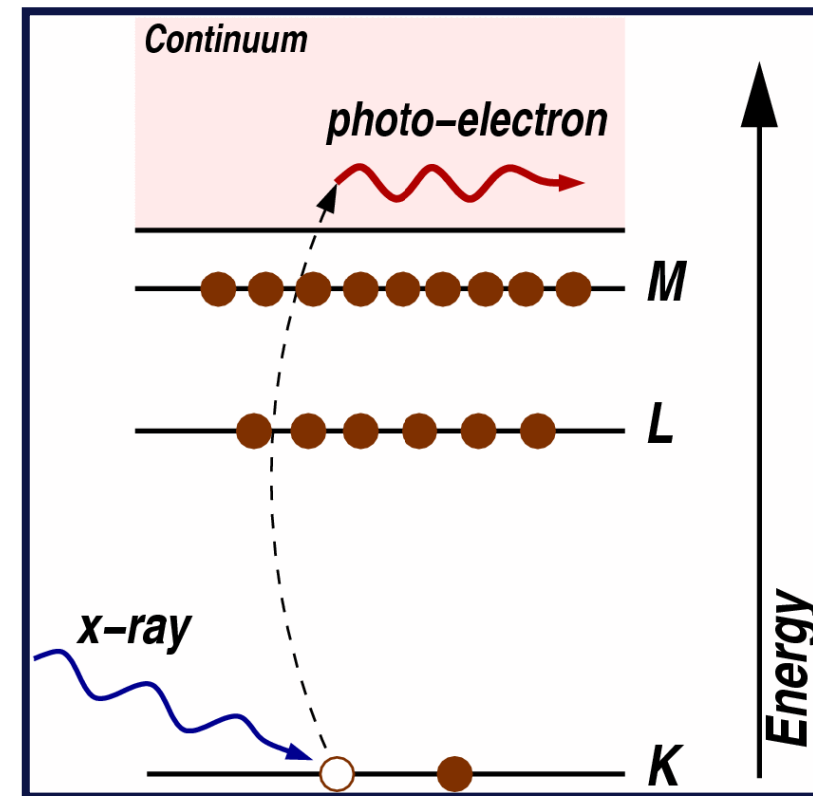
The Absorption Coefficient μ



Photoelectric Absorption: basics

- X-rays are absorbed by all matter through the **photoelectric effect**
- An X-ray is absorbed by an atom when the energy of the X-ray is transferred to a core-level electron (K , L , or M shell) which is ejected from the atom.
- The atom is left in an **excited state** with an empty electronic level (a **core hole**). Any excess energy from the X-ray is given to the ejected **photoelectron**.
- Depending on its energy, and its position in the sample, the photoelectron can:

- Wander around the absorber, scatter from other electrons ie. neighbor atoms (XAFS)
- Leave the surface of the sample (core – level photoemission spectroscopy)

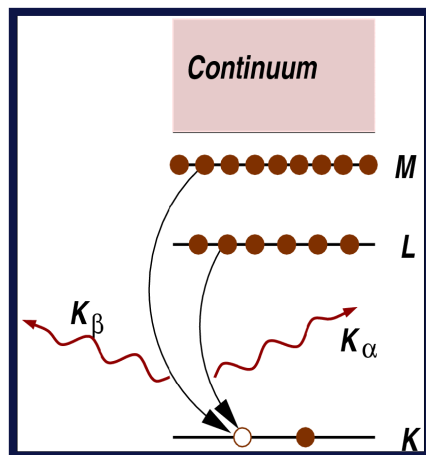


De-excitation: Fluorescence and Auger Effect

- When X-rays are absorbed by the photoelectric effect, the excited core-hole will relax back to a “ground state” of the atom.
- A higher-level core electron drops into the core hole, and a **fluorescent X-ray** or **Auger electron** is emitted.
- X-ray fluorescence and Auger emission occur at discrete energies characteristic of the absorbing atom, and can be used to identify the absorbing atom.

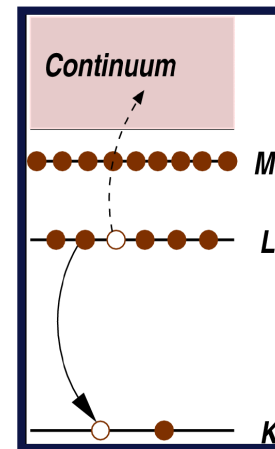
X-ray Fluorescence

An X-ray with energy = the difference of the core-levels is emitted.



Auger Effect

An electron is promoted to the continuum from another core-level.



The “zoo” of core - hole spectroscopies

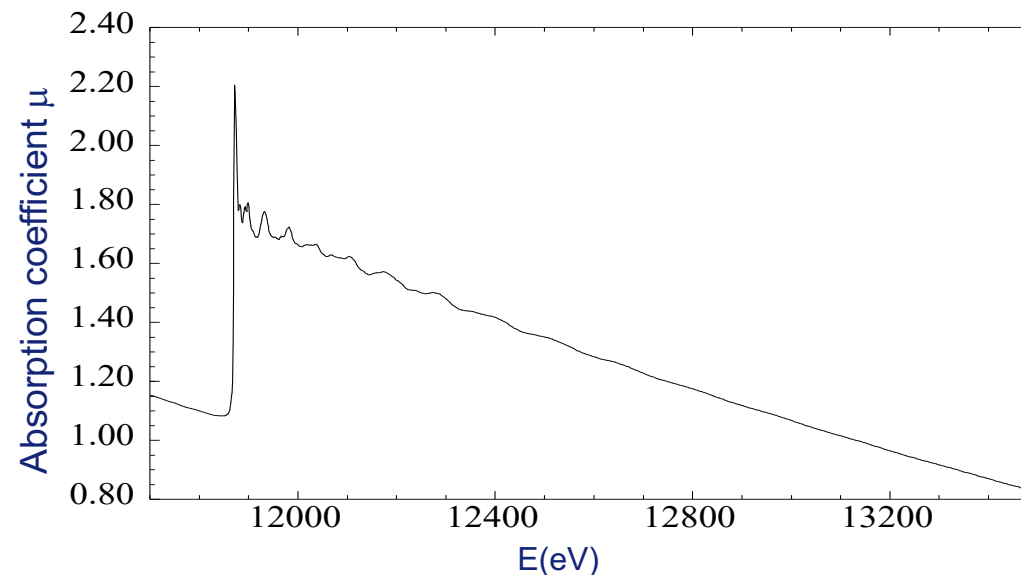
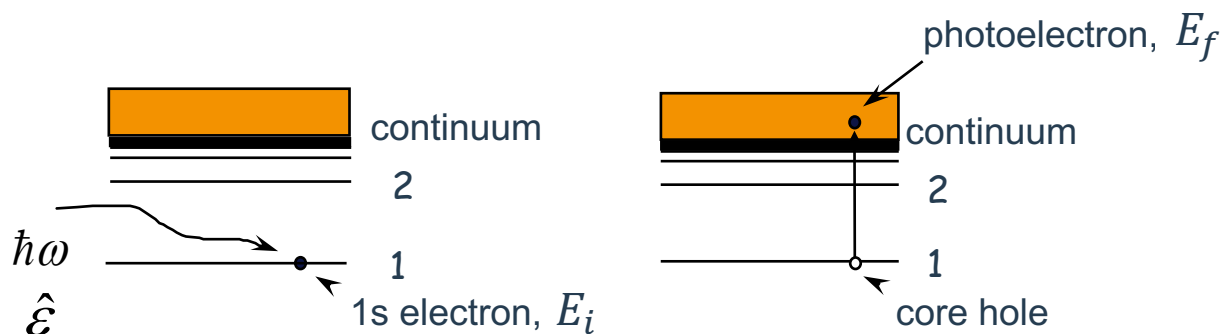
	X-ray Absorption Spectroscopy (XAS)	X-ray Emission Spectroscopy (XES)	Inelastic X-ray Scattering (IXS)
Incident photon energy far from core-electron binding energy	Extended X-ray Absorption Fine Structure (EXAFS)	Non resonant XES K_β spectroscopy	Non resonant IXS of core-electrons X-ray Raman Scattering (XRS)
Incident photon energy around core-electron binding energy	X-ray Absorption Near Edge Structure (XANES)	Resonant XES (RXES)	Resonant IXS (RIXS) High Energy Resolved Fluorescence Detected XAS (HERFD)
Dichroism	X-ray Linear Dichroism (XLD) X-ray Magnetic Linear Dichroism (XMLD) X-ray Magnetic Circular Dichroism (XMCD)	RXES – MCD	RIXS – MCD

Transition Probability

1. The absorption coefficient $\mu(E)$ reflects the probability of absorption of a photon

$|i\rangle$ photon ($\hbar\omega, \hat{\varepsilon}$)
+ core electron

$\langle f|$ photoelectron +
core hole



2. From a quantum mechanics perspective, the absorption coefficient $\mu(E)$ reflects the **transition probability** from **initial state i** to **final state f**

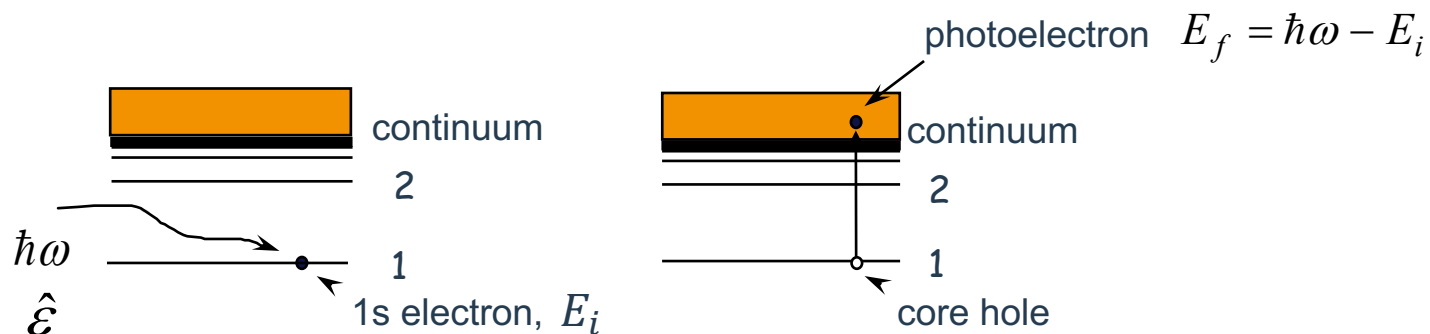
Transition Probability

$$\mu(E) = \mu(\omega) \approx \sum_f W_{if} \quad W_{if} \approx \left| \langle f | \hat{H}_I | i \rangle \right|^2 \rho(E_f) \quad \text{transition probability from } |i\rangle \text{ to } |f\rangle$$

$\rho(E_f)$ density of final states, compatible with energy conservation: $\delta(E_f - E_i - \hbar\omega)$

$\left| \langle f | \hat{H}_I | i \rangle \right|$ matrix element of H_I between initial and final state

$\hat{H}_I$ "electromagnetic field $\leftrightarrow$ atom" interaction hamiltonian operator



Wavefunctions and probability density

■ In quantum mechanics, we use the concept of wavefunction to describe the probability of finding a particle at position $\vec{r}$ and time t : $\psi(\vec{r}, t)$

■ This function represents a “probability amplitude” and is a complex number:

Initial state: $|i\rangle$ or $\psi_i(\vec{r})$ $\langle i|$ or $\psi_i^*(\vec{r})$ final state: $|f\rangle$ or $\psi_f(\vec{r})$ $\langle f|$ or $\psi_f^*(\vec{r})$

■ The modulus squared of this quantity represents a probability or probability density: $|\psi_i(\vec{r})\psi_i^*(\vec{r})| = |\psi_i(\vec{r})|^2$ where : $|\int d\vec{r} \psi_i(\vec{r})\psi_i^*(\vec{r})|^2 = 1$

■ The notation $|\langle f|\hat{H}_I|i\rangle|^2$ stands for : $|\int d\vec{r} \psi_f^*(\vec{r})\hat{H}_I \psi_i(\vec{r})|^2$

■ So, if $\hat{H}_I|i\rangle = |f\rangle$ then the transition probability from $|i\rangle$ to $|f\rangle$ is equal to 1.

Transition Probability

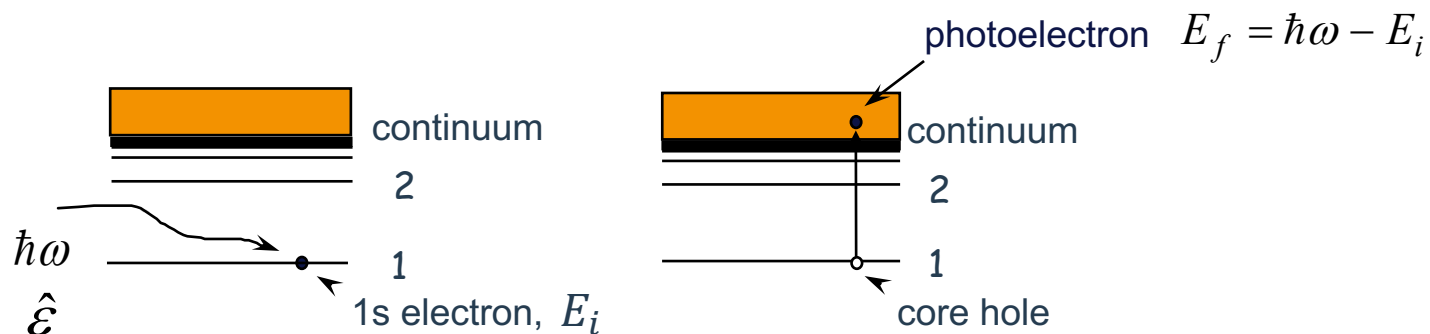
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Consequence 1: “Core-hole broadening”

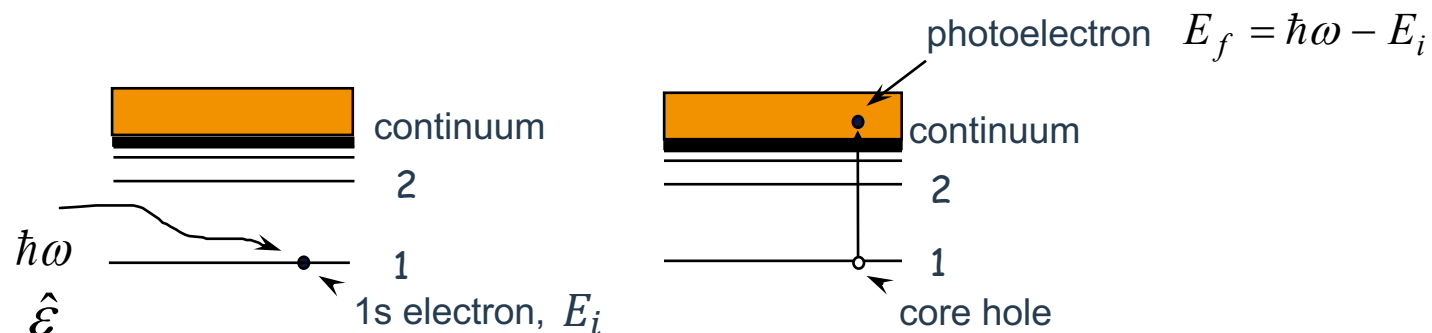
$\rho(E_f)$ density of final states, compatible with energy conservation: $\delta(E_f - E_i - \hbar\omega)$

Uncertainty in final state energy due to finite corehole lifetime δE_{ch} : $\delta E_f = \delta E_{ch} \sim \frac{\hbar}{\delta t_{ch}}$

Cu K ~ 1.4 eV; Cu L ~ 0.5 eV

Ag K ~ 7 eV

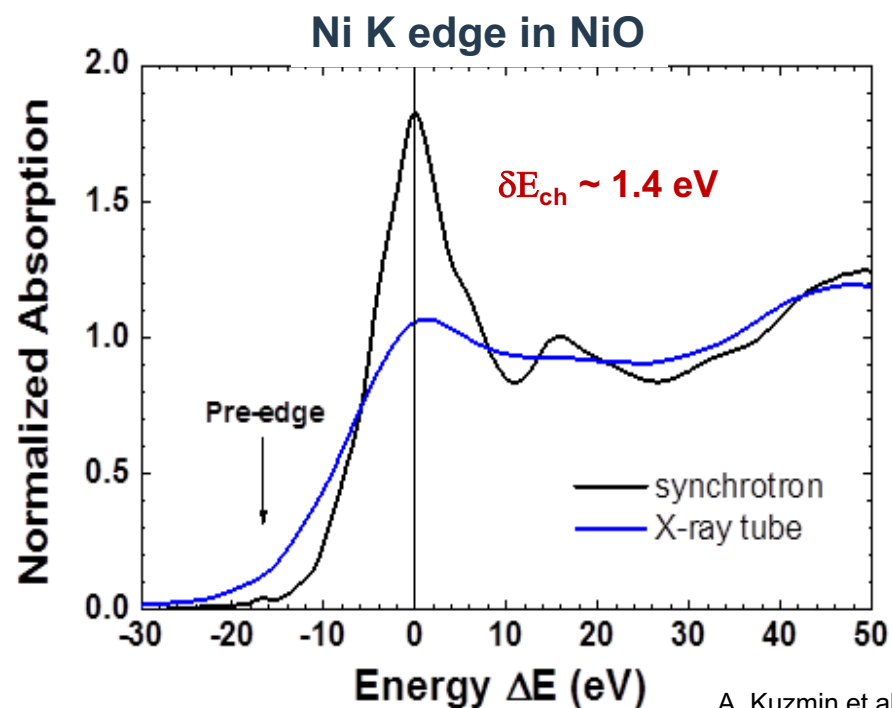
Au K ~ 50 eV



Spectral broadening: two contributions

■ Intrinsic (core-hole): $\frac{\delta E_{\text{ch}}}{E} \sim$ inversely proportional to core-hole lifetime

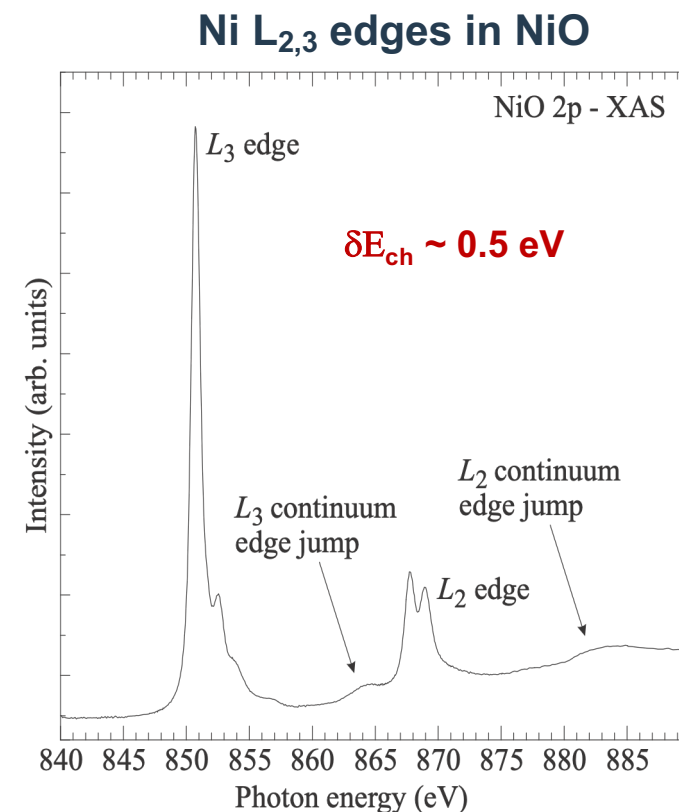
■ Instrumental: $\frac{\delta E_{\text{instr}}}{E} \sim$ divergence of incoming beam & intrinsic resolution of monochromator



$$\frac{\delta E_{\text{instr}}}{E} \sim 10^{-4}$$

$$\frac{\delta E_{\text{instr}}}{E} \sim 10^{-3}$$

A. Kuzmin et al. Phys. Stat. Sol. (a) 135 (1993) 133-141



L. H. Tjeng et al.

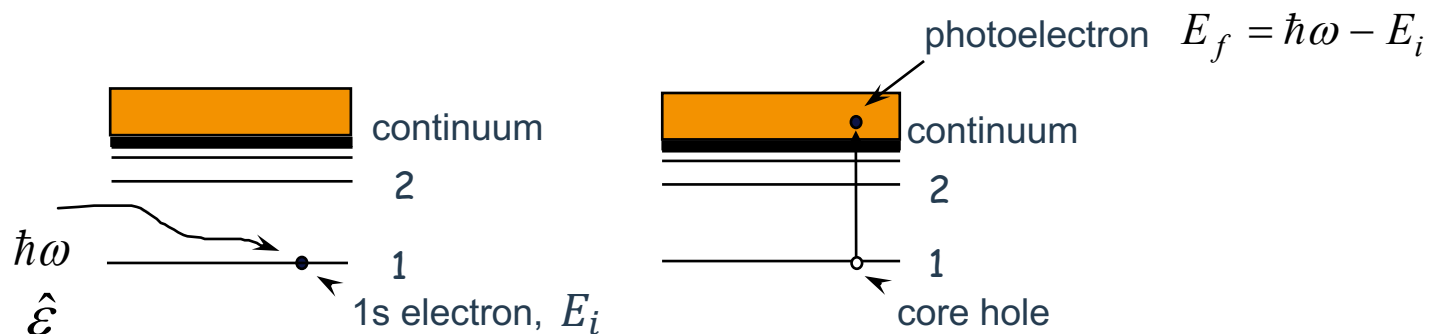
Transition Probability

$$\mu(E) = \mu(\omega) \approx \sum_f W_{if} \quad W_{if} \approx \left| \langle f | \hat{H}_I | i \rangle \right|^2 \rho(E_f) \quad \text{transition probability from } |i\rangle \text{ to } |f\rangle$$

$\rho(E_f)$ density of final states, compatible with energy conservation: $\delta(E_f - E_i - \hbar\omega)$

$\left| \langle f | \hat{H}_I | i \rangle \right|$ matrix element of H_I between initial and final state

$\hat{H}_I$ "electromagnetic field $\leftrightarrow$ atom" interaction hamiltonian operator



Consequences 2 & 3: “Dipole Selection Rules & Dichroism”

$\hat{H}_I$ “electromagnetic field $\leftrightarrow$ atom” interaction hamiltonian operator

$$\hat{H}_I \approx \hat{\epsilon} \cdot \vec{r} \quad (\text{“The Dipole Approximation”})$$

polarization vector of photon

position vector of core electron

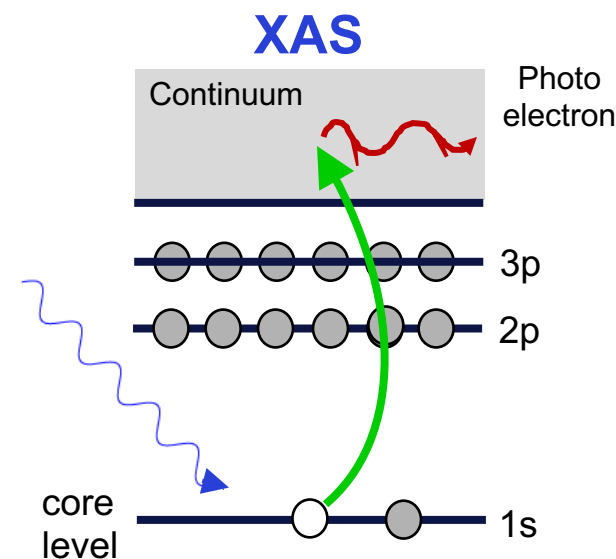
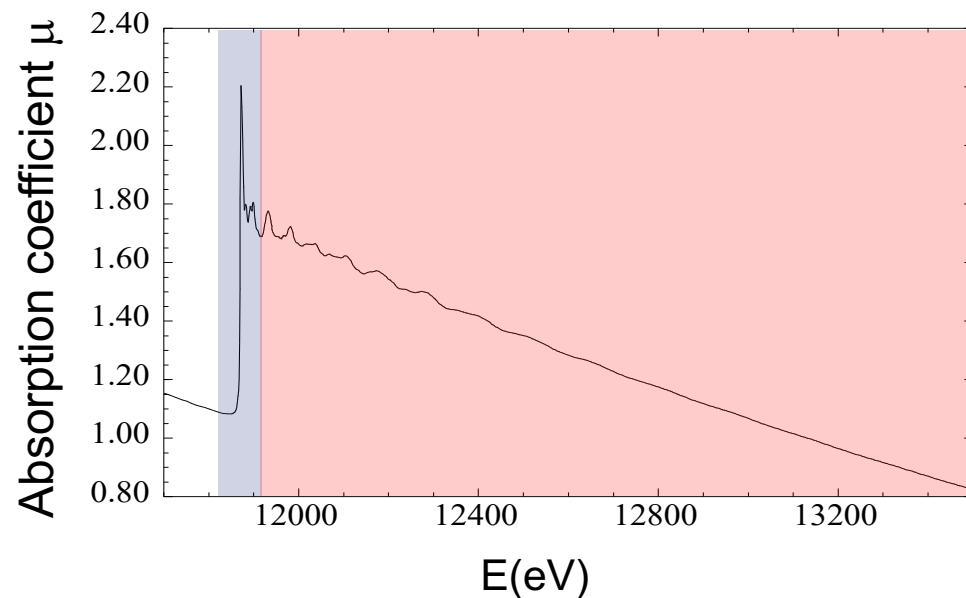
$$\mu(\omega) \approx \sum_f |\langle f | \hat{\epsilon} \cdot \vec{r} | i \rangle|^2$$

Many matrix elements cancel out
→ not all transitions are possible (Dipole Selection Rules)

Absorption coefficient depends on $\hat{\epsilon}$
→ polarization effects (i.e. X-ray dichroism)

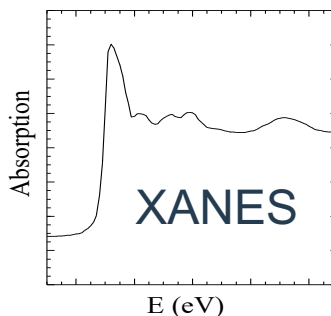
i.e. $\Delta\ell = \pm 1$ K-edge: $1s (\ell = 0) \rightarrow p (\ell = 1)$
L-edge: $2p (\ell = 1) \rightarrow s (\ell = 0), d (\ell = 2)$

X-ray Absorption (Fine Structure) Spectroscopy



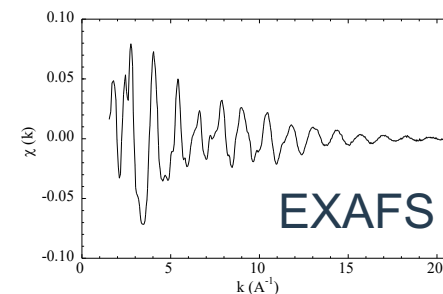
Electronic structure

- valence, oxidation state
- site symmetry, coordination chemistry
- orbital hybridization, occupancy



Local structure

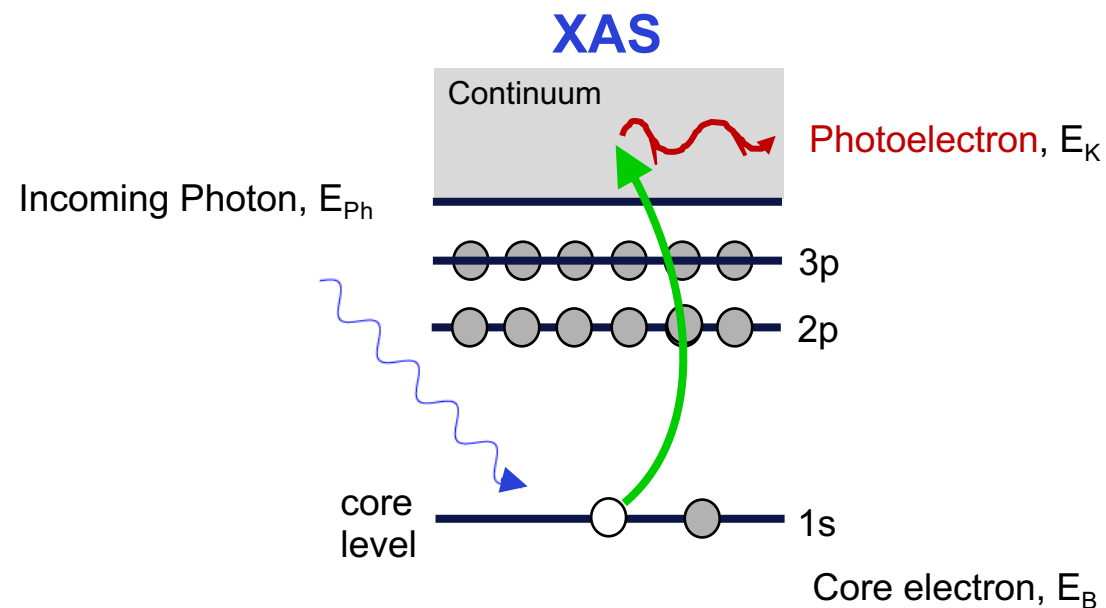
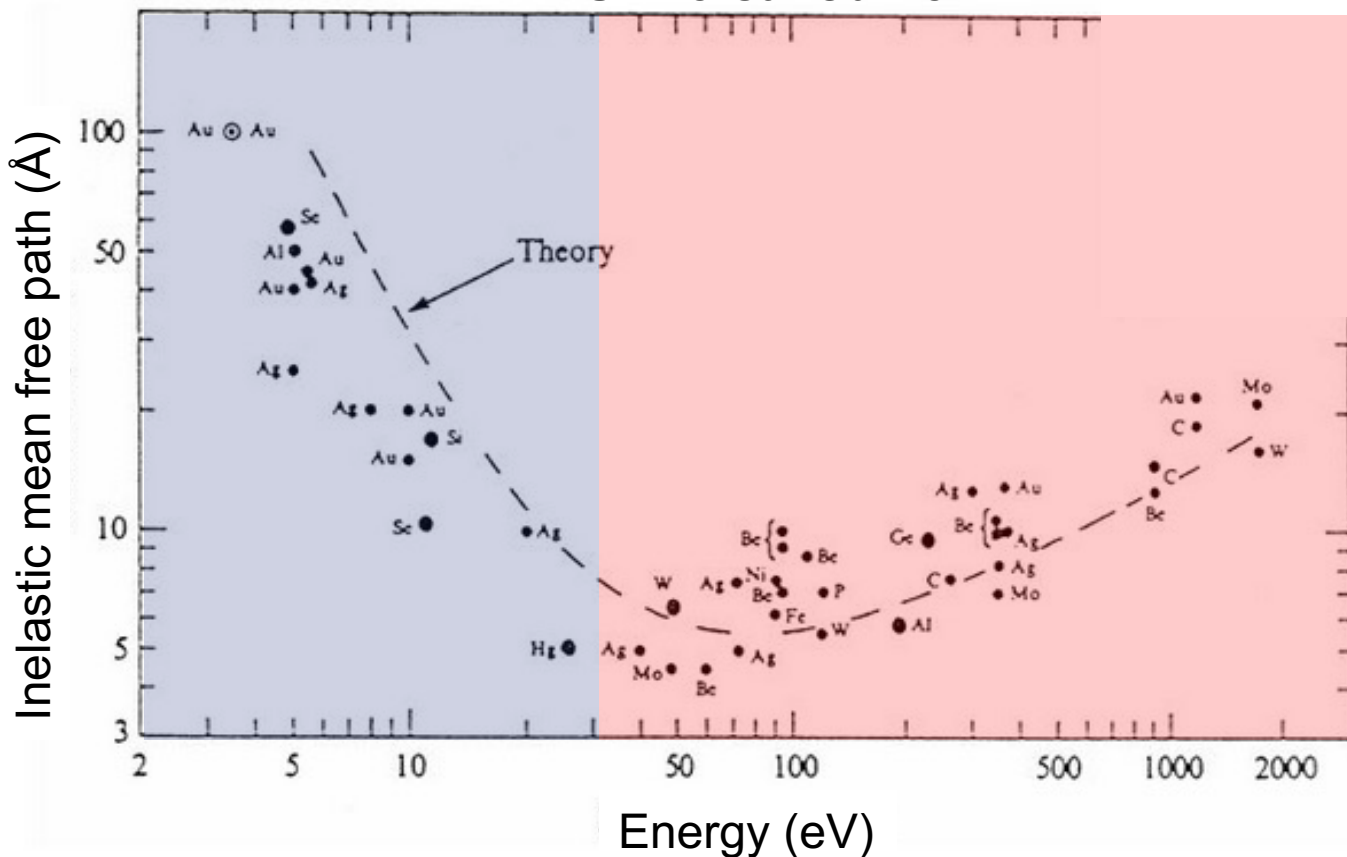
- near neighbor distances
- number and type of neighbors
- static and thermal relative displacements



The kinetic energy of the photoelectron is important....

“how far do electrons travel before losing their energy”

Universal Curve



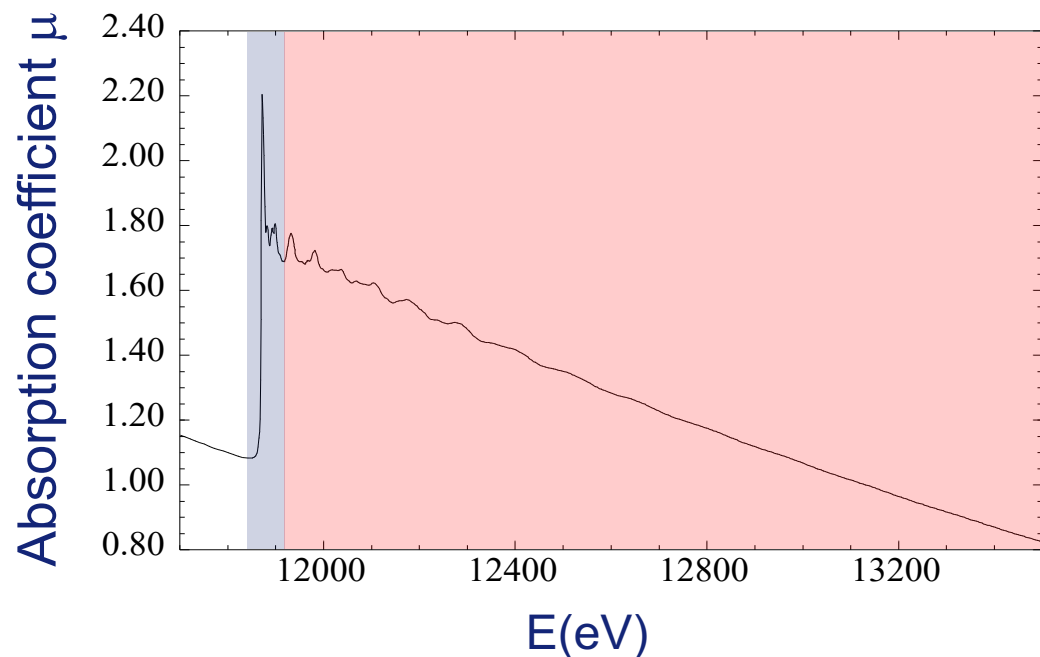
Any excess energy from the X-ray is given to the ejected **photoelectron**.

The kinetic energy of the photoelectron is:

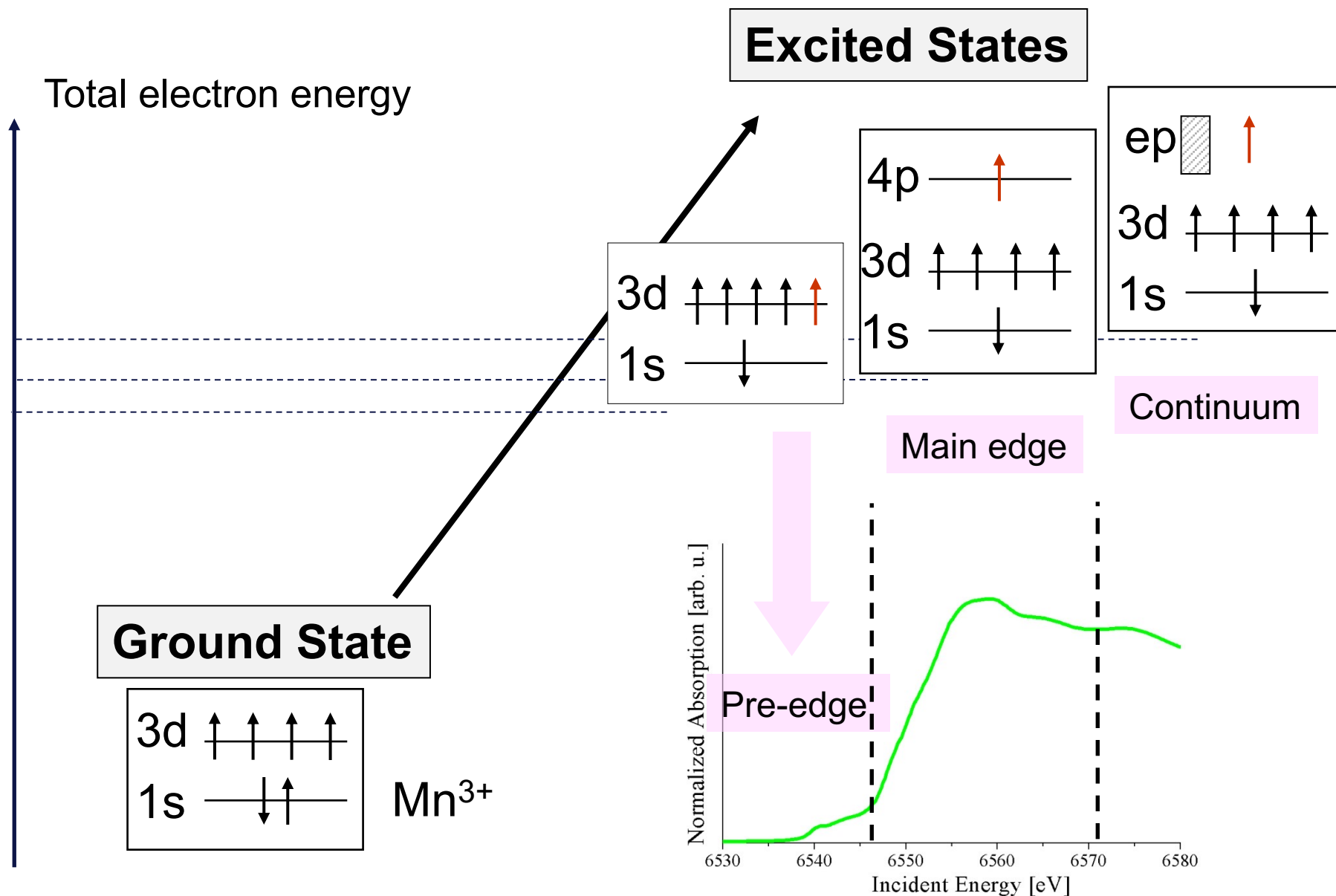
$$E_{\text{Kinetic}} = E_{\text{Ph}} - E_B$$

Explains why XANES and EXAFS cannot be analyzed in the same way

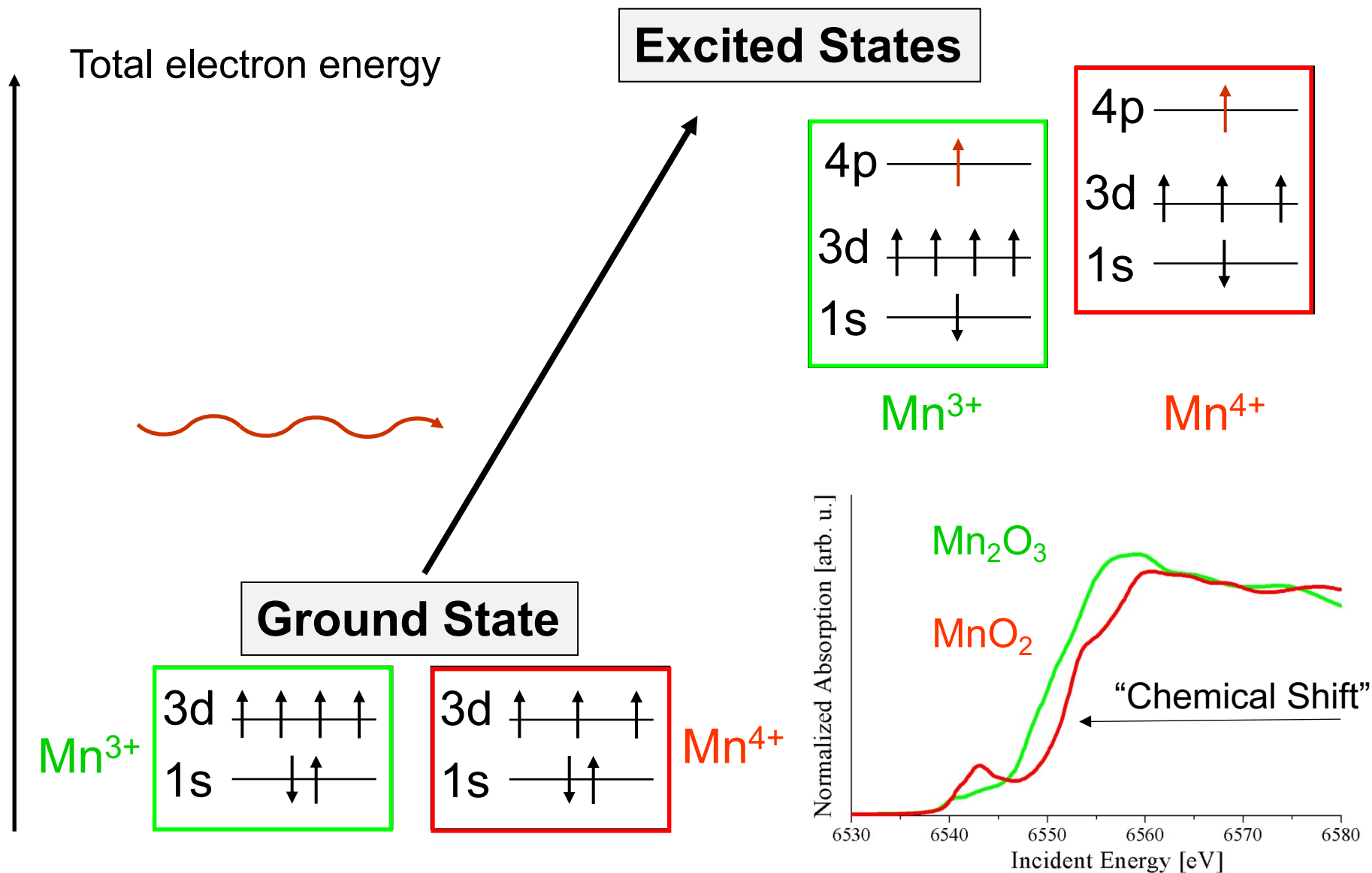
- Approximations can be used to interpret EXAFS, that are not valid for XANES
- Cannot be analyzed in the same way
- Different information content.



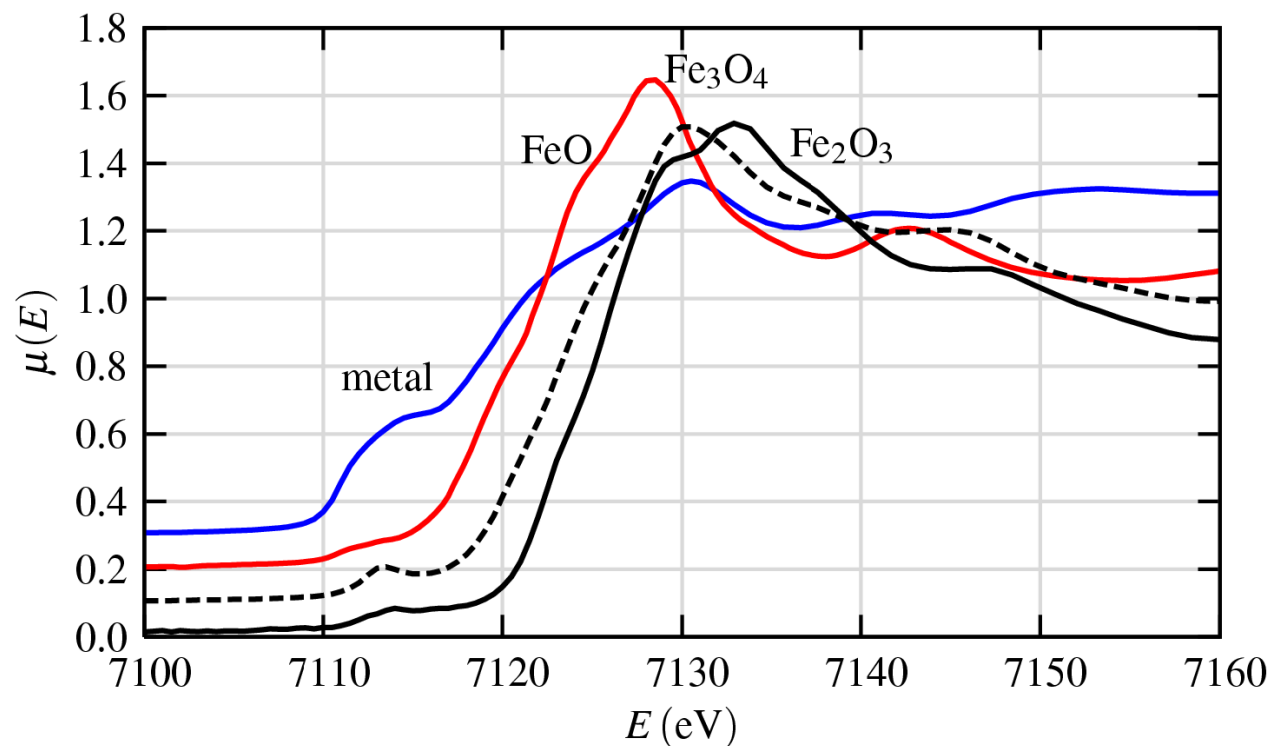
X-ray absorption near edge spectroscopy (XANES)



XANES: onset of absorption sensitive to oxidation state of absorber



Edge shifts and pre-edge peaks in Fe oxides by Fe_K XANES

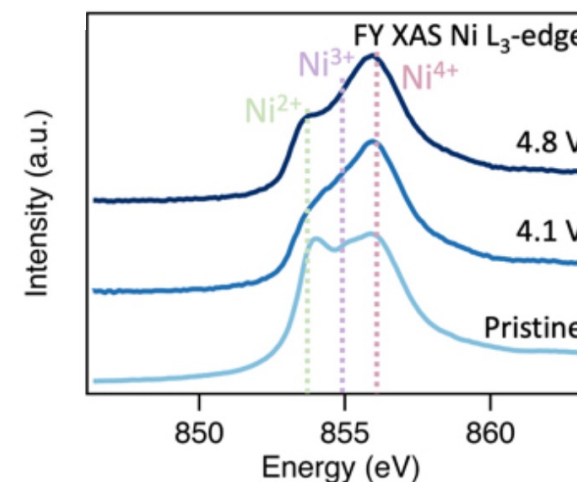
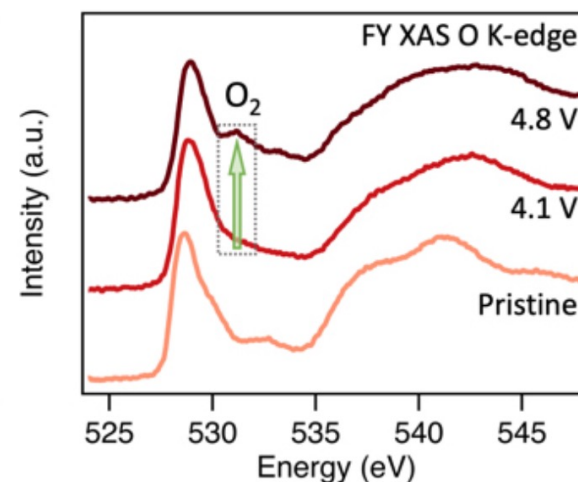
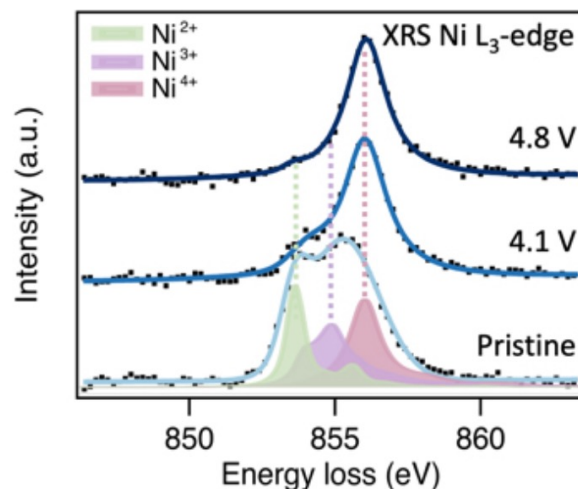
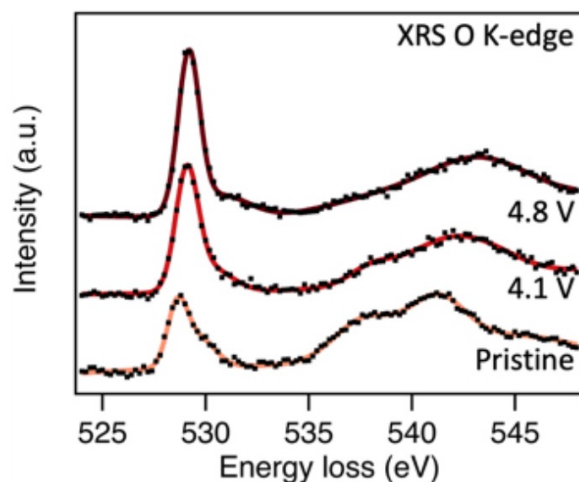


- ❖ The shift of the edge position can be used to determine the valence state
- ❖ The heights and positions of pre-edge peaks can also be reliably used to determine $\text{Fe}^{3+}/\text{Fe}^{2+}$ ratios (and similar ratios for many cations).

Redox dynamics in lithium-ion battery cathodes

Towards lower-cost, longer-lasting lithium-ion batteries for green transportation

- Top candidate for enhanced charge storage and stability over thousands of cycles → LiNiO_2 cathodes
- Batteries degrade and lose capacity rapidly → need to understand exactly how O and Ni exchange electrons deep within the bulk material during operation to stop this degradation
- Need depth-resolved analysis of redox process, to distinguish reversible bulk redox activities from near-surface degradation



- X-ray Raman spectroscopy/scattering @ID20 ESRF
- Hard X-rays** (10keV): **penetrates** ($\sim 10 \mu\text{m}$) inside battery cell to map electronic changes in real time (*operando*).
- Redox process involves Ni-O rehybridization throughout the bulk of LiNiO_2

- FY-XANES spectroscopy @B07-B DLS
- Soft X-rays** → information depth $\sim 200 \text{ nm}$
- Surface degradation leads to Ni reduction and trapping of molecular O_2 → development of a denser NiO-like reduced surface layer that hinders Li transport.

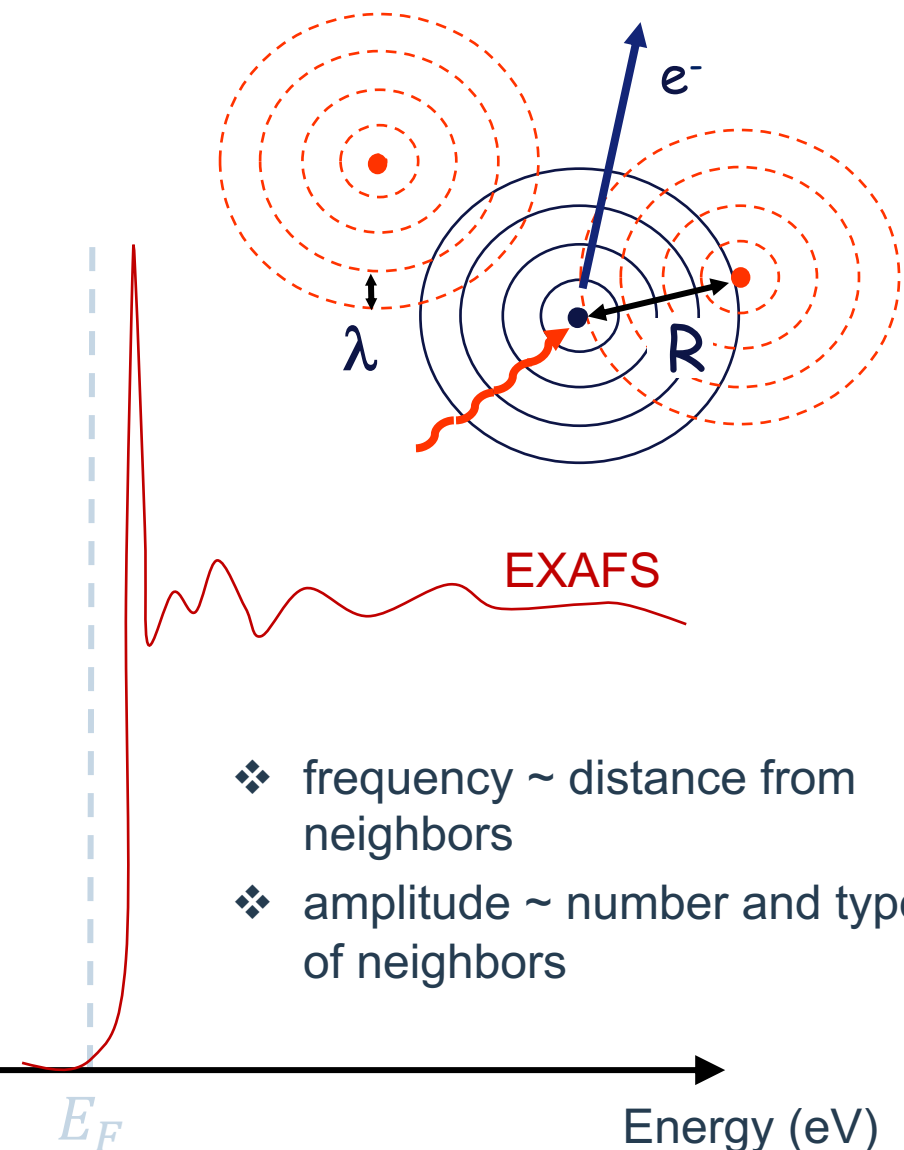
Extended X-ray absorption fine structure (EXAFS): measuring local structure

Due to a **quantum effect**, the autointerference of photoelectron wave modifies the absorption coefficient value:

1. As E is scanned above E_0 , E_{kin} is varied, and consequently k and λ .

$$E_{\text{kin}} = E - E_0 = \frac{p^2}{2m} = \frac{\hbar^2 k^2}{2m}$$

2. The outgoing and backscattered parts of the wave interfere either constructively or destructively, depending on the ratio between λ and R .
3. It is the **interference** between outgoing and incoming waves that gives rise to the sinusoidal variation of $\mu(E)$



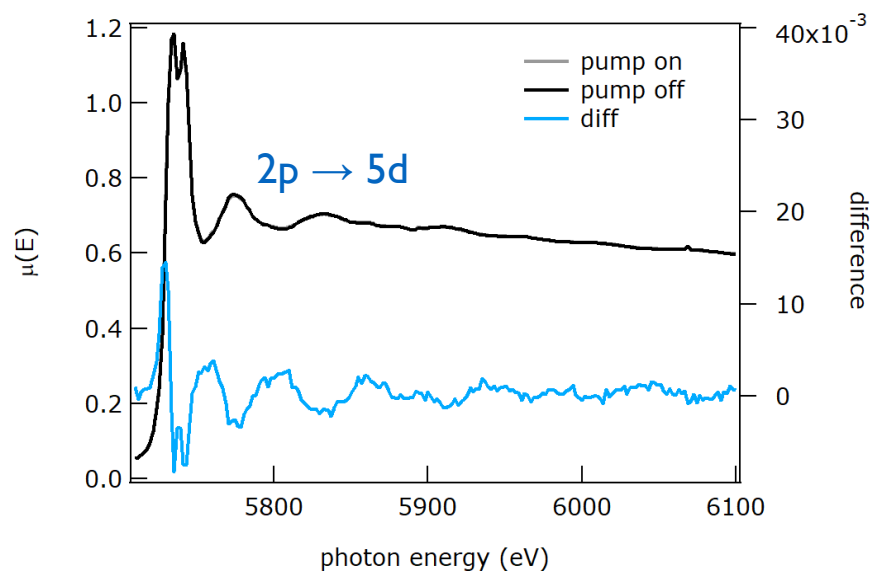
Structure and electronic properties of excited states in CeO₂ by fs-resolved Ce L₃ XAS

Improving water splitting processes for hydrogen production

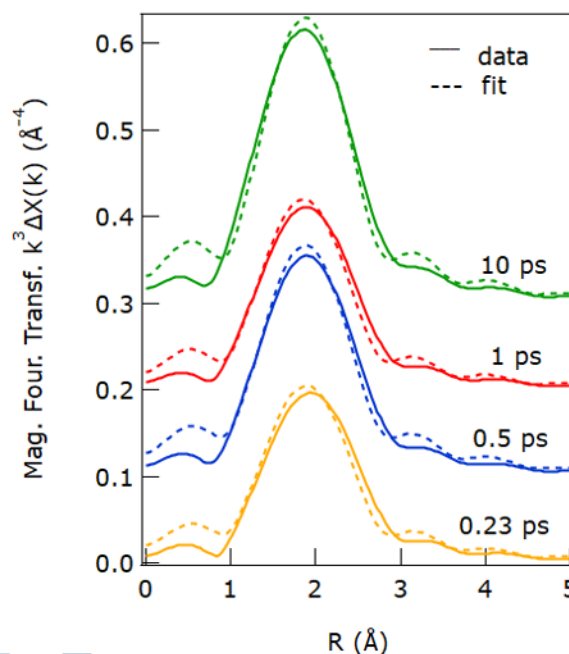
Coupling between excited states and LO phonons → polarons

How does this influence functionality of photocatalytic materials

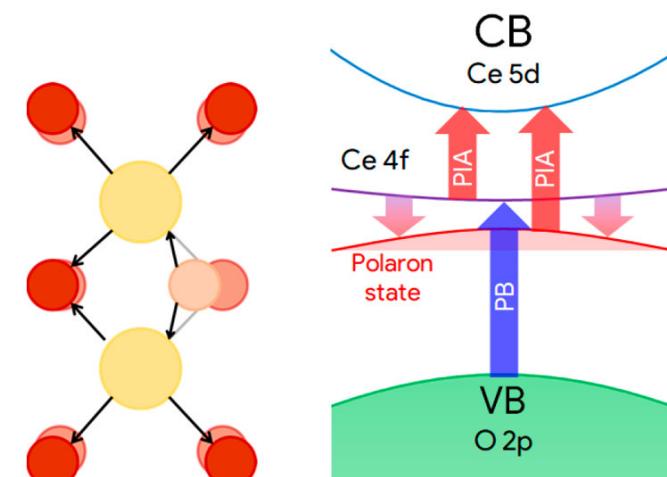
pump-probe Ce L₃ edge EXAFS at $\Delta t = 2$ ps



$\Delta R_{\text{Ce-O}} \rightarrow +0.13 \text{ \AA}$

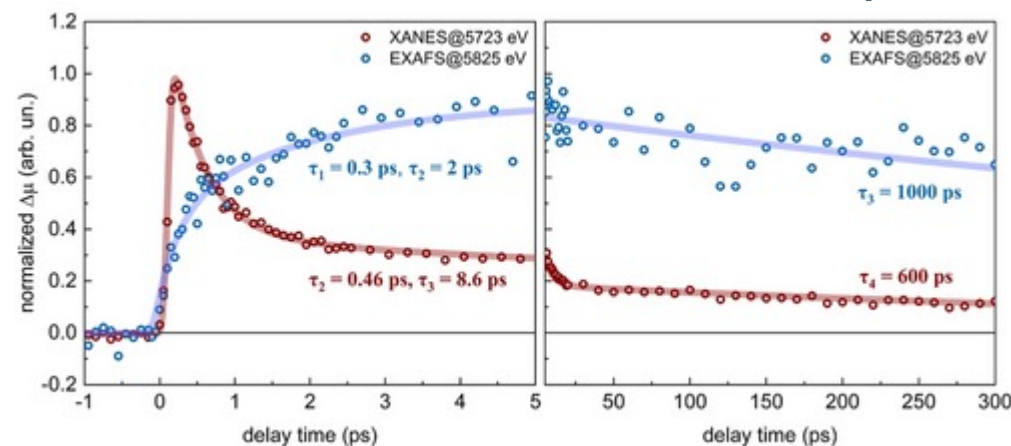


photoinduced structural distortion consistent with the formation of a small polaron state



Cresi et al. *JPCL* 11, 5686 (2020)

Electronic → 300 - 500 fs Structural → 600 ps - 1 ns



FXE@EuXFEL fs-resolved XAS

Arsenic sequestration by organic sulphur in peat by As_K EXAFS

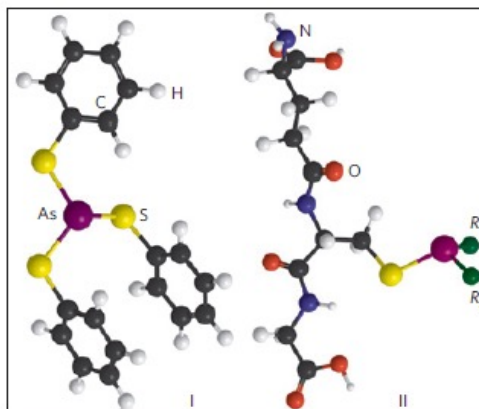
Does natural organic matter serve as geochemical trap for As ?

- Wetlands cover more than 6% of the global ice-free land area
- Play major role in storage, transformation, mobilization of trace elements

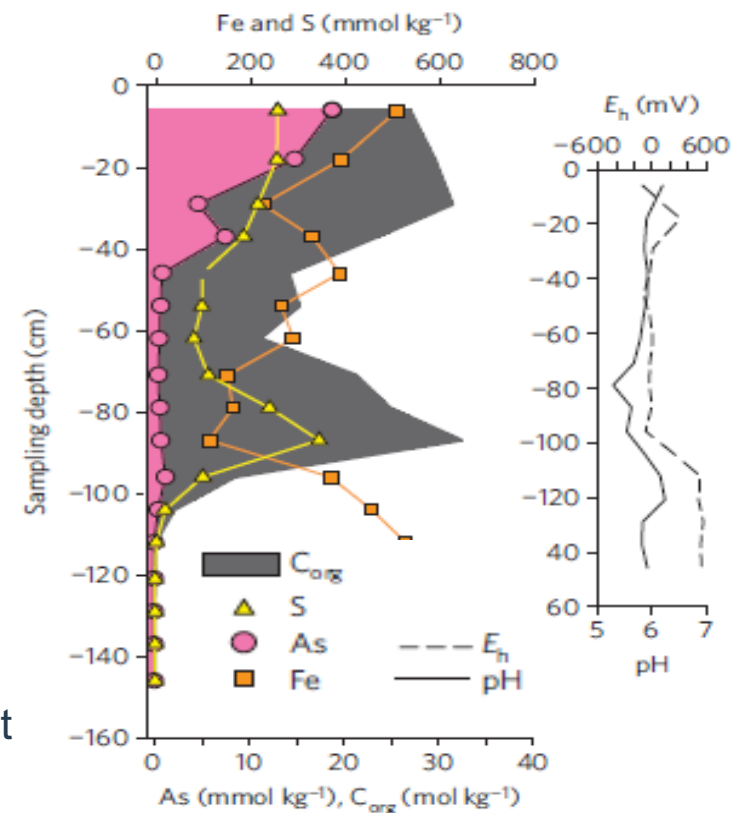
Peat samples collected from a naturally arsenic-enriched wetland in Switzerland.

As_K EXAFS (3-1000 ppm)

Trap: covalent bonds between As⁺³ and organic S groups



- Spectroscopic evidence for As binding to peat
- Identified dominating binding mechanism
- Quantified extent of As binding to peat



Outline

■ Radiation – matter interaction

■ Experimental Techniques

■ Based on Scattering

- ▶ Elastic Scattering
- ▶ Inelastic Scattering

■ Based on Photoelectric Absorption

▶ Core-hole Spectroscopies

- X-ray Absorption Spectroscopy
- X-ray Emission Spectroscopy, RIXS, X-ray Raman
- X-ray Polarization Dependent Spectroscopies
- X-ray Photoemission Spectroscopy

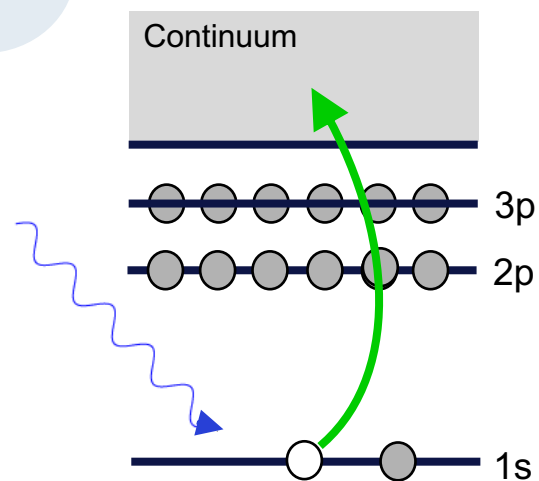
■ Imaging Methods

■ Pause : 11h40-11h50

X-Ray Emission Spectroscopy: XES

1

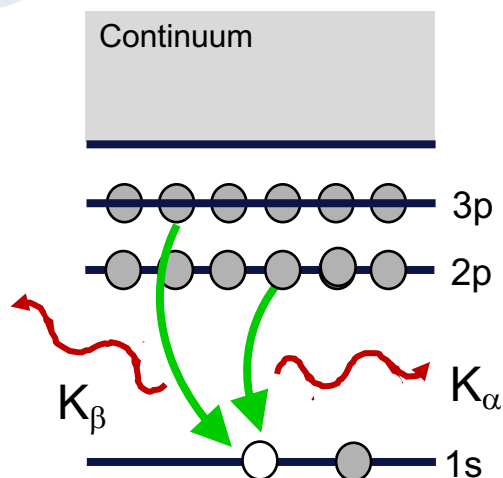
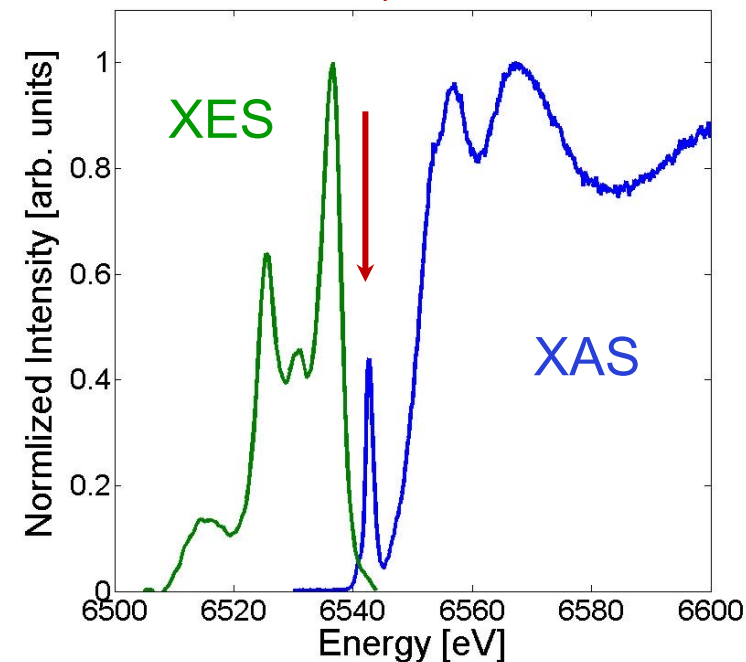
XAS



- K_α and K_β : non-resonant XES
- Incident photon energy is far from absorption edge
- Photoelectron well into continuum

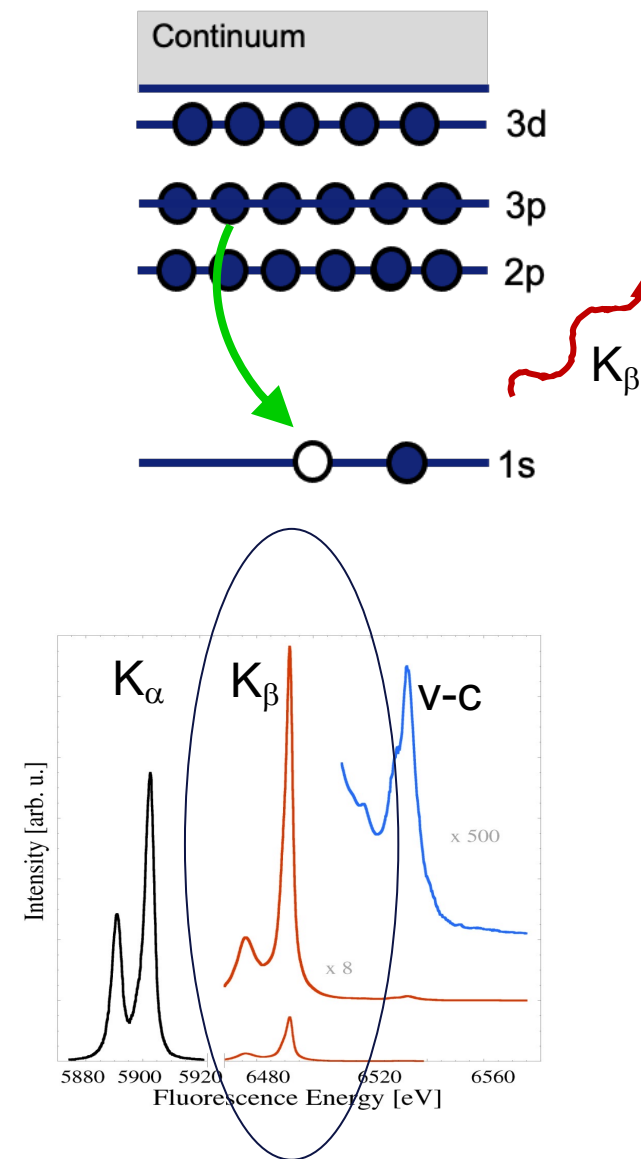
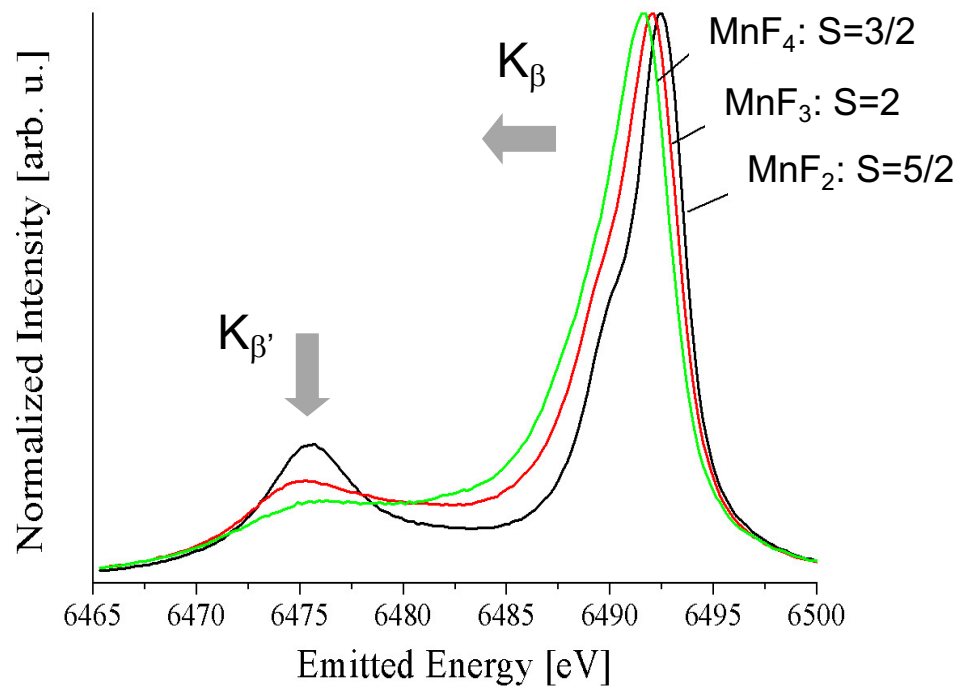
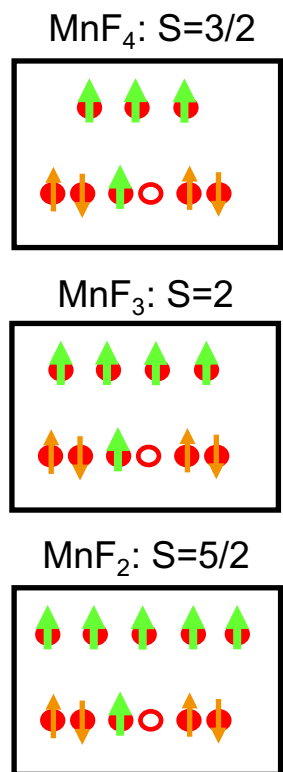
2

XES

Occupied states E_F Unoccupied states

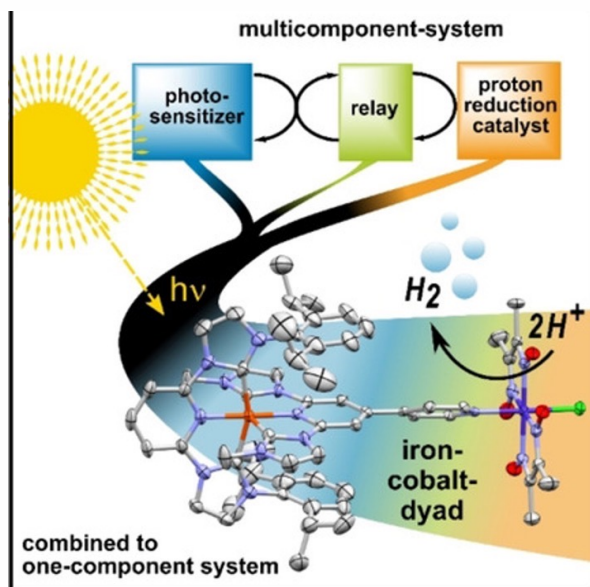
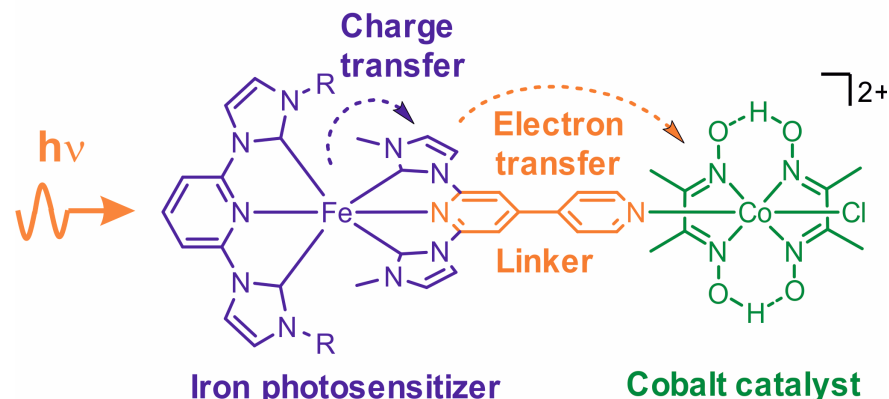
Valence band spin state by K_β XES

- Local magnetic moment
- Spin transitions

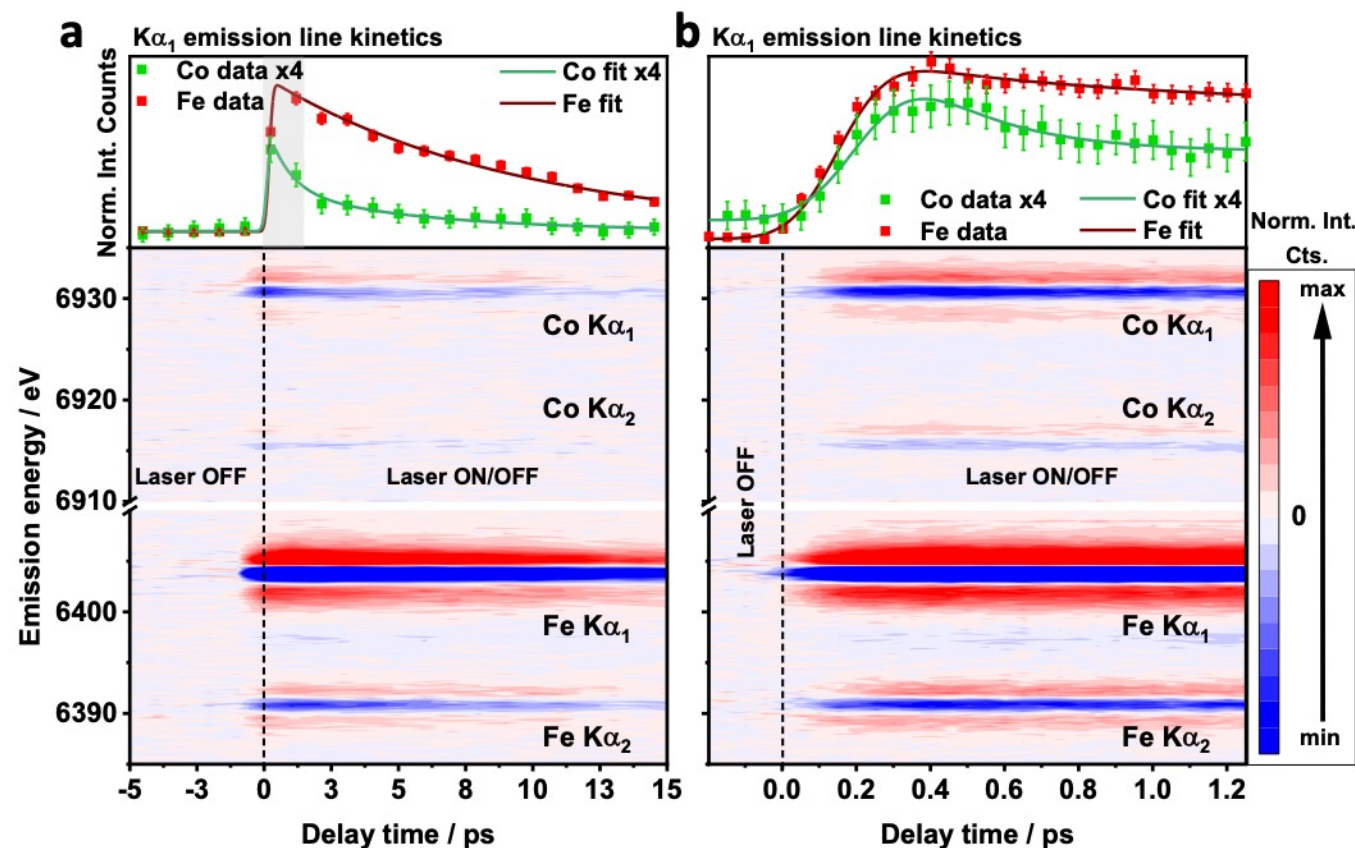


Excited State Landscape in a Base Metal Dyad by fs-resolved K_{α} XES

Using light for a sustainable production of molecular hydrogen



FXE@EuXFEL Ultrafast Two-Color XES



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■ Based on Photoelectric Absorption

▶ Core-hole Spectroscopies

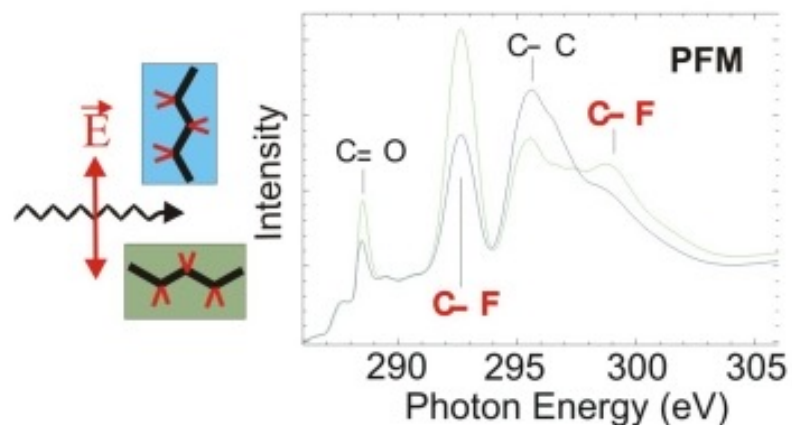
- X-ray Absorption Spectroscopy
- X-ray Emission Spectroscopy, RIXS, X-ray Raman
- X-ray Polarization Dependent Spectroscopies
- X-ray Photoemission Spectroscopy

■ Imaging Methods

Polarisation Dependent X-ray Spectroscopies

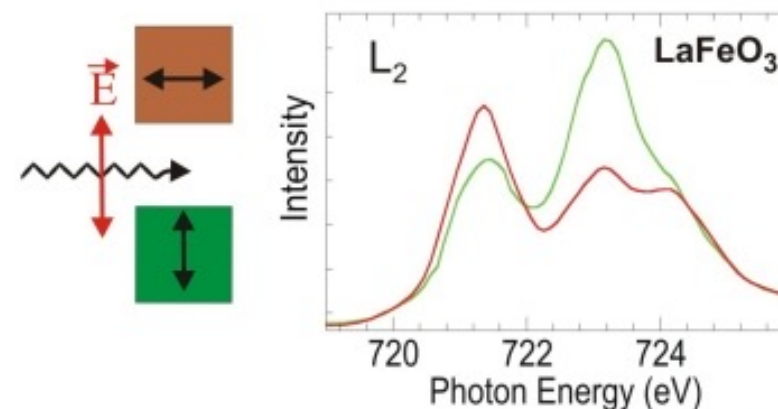
X-ray Linear Dichroism

Stöhr *et al.*, Phys. Rev. Lett. **47**, 381 (1981)



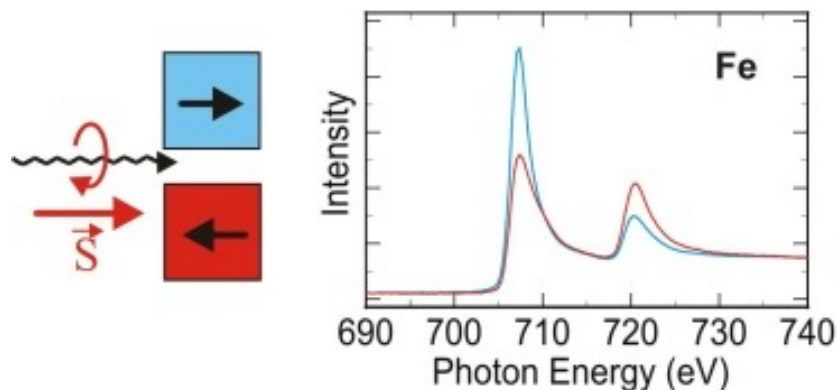
X-ray Magnetic Linear Dichroism

Van der Laan *et al.*, Phys. Rev. B **34**, 6529 (1986)



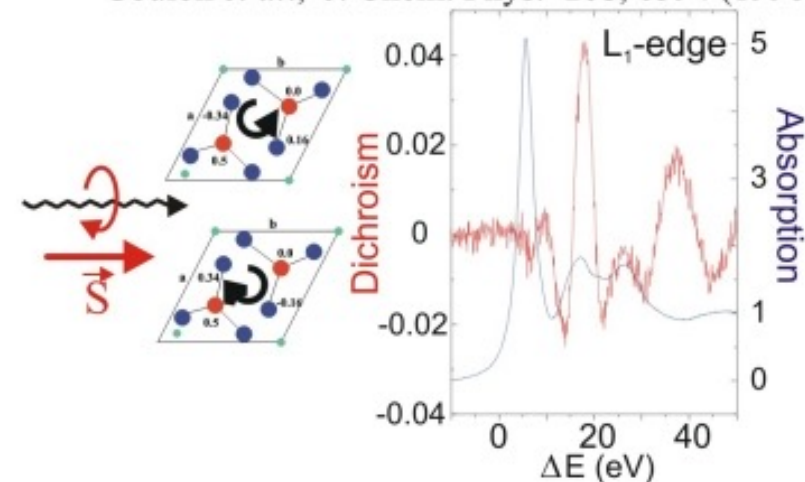
X-ray Magnetic Circular Dichroism

Schütz *et al.*, Phys. Rev. Lett. **58**, 737 (1987)



X-ray Natural Circular Dichroism

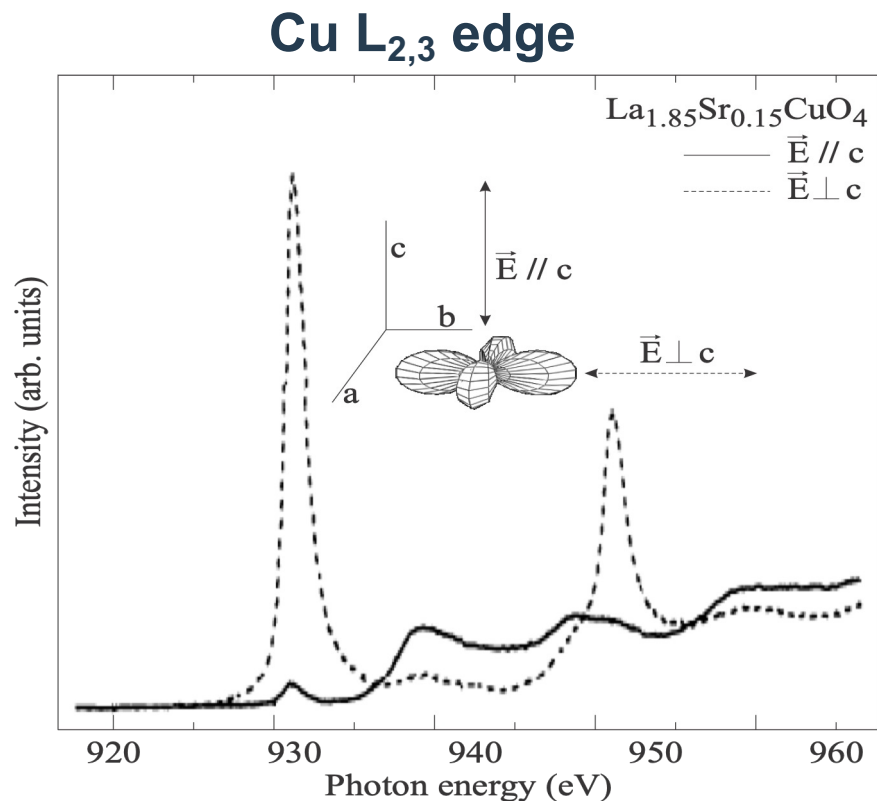
Goulon *et al.*, J. Chem. Phys. **108**, 6394 (1998)



X-ray Linear Dichroism

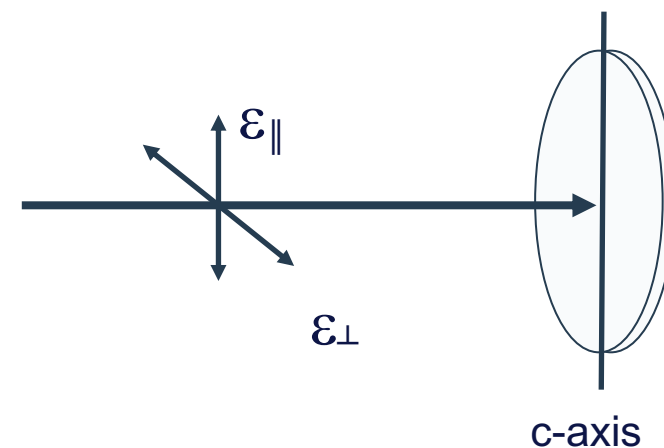
- Dependence of absorption on relative orientation between x-ray polarization and crystallographic axis
- $\vec{\epsilon}$: search-light for no. of valence holes in different directions

$$XLD = \frac{\mu^{\parallel} - \mu^{\perp}}{\mu^{\parallel} + \mu^{\perp}} \quad \begin{array}{l} \mu^{\parallel} : \vec{\epsilon} \parallel \vec{c} \\ \mu^{\perp} : \vec{\epsilon} \perp \vec{c} \end{array}$$



C. T. Chen et al PRL 68, 2543 (1992)

linearly polarized
X-rays



- The polarisation dependence → symmetry of empty orbitals
- Sensitive to 3d orbital occupation along polarization direction

X-ray Magnetic Circular Dichroism

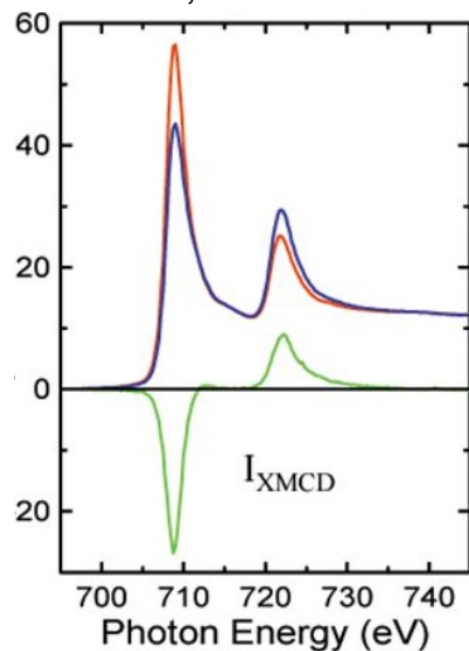
- To make the absorption process spin dependent
- circularly polarized photons
- transfer their angular momentum to photoelectron

$$XMCD = \frac{\mu^L - \mu^R}{\mu^L + \mu^R}$$

μ^L : absorption coefficient using CPL

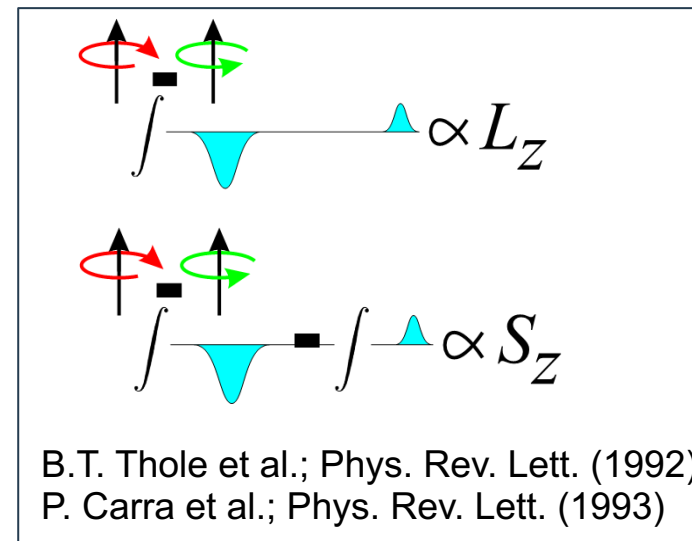
μ^R : absorption coefficient using CPR

Fe $L_{2,3}$ -edge XMCD

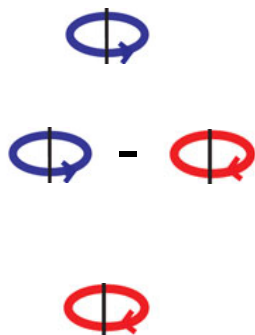
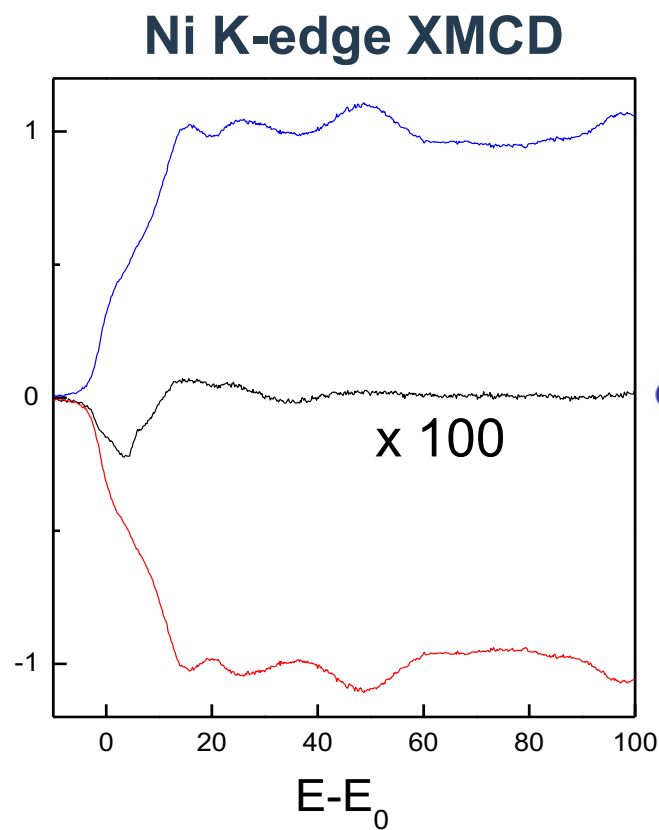


L-edge XMCD ($2p \rightarrow 3d$)
Direct probe of valence band magnetization

- A unique probe for element-specific spin- and orbital magnetization
- L_z and S_z can be evaluated directly from Sum Rules



X-ray Magnetic Circular Dichroism



$$XMCD = \frac{\mu^L - \mu^R}{\mu^L + \mu^R}$$

μ^L : absorption coefficient using CPL

μ^R : absorption coefficient using CPR

K-edge XMCD ($1s \rightarrow 4p$)

Indirect information on valence band magnetization

→ Much smaller signal

→ More difficult interpretation

- net magnetic moment on absorber
- sublattice magnetization
- changes in ferromagnetic order
- hybridization effects, induced moments

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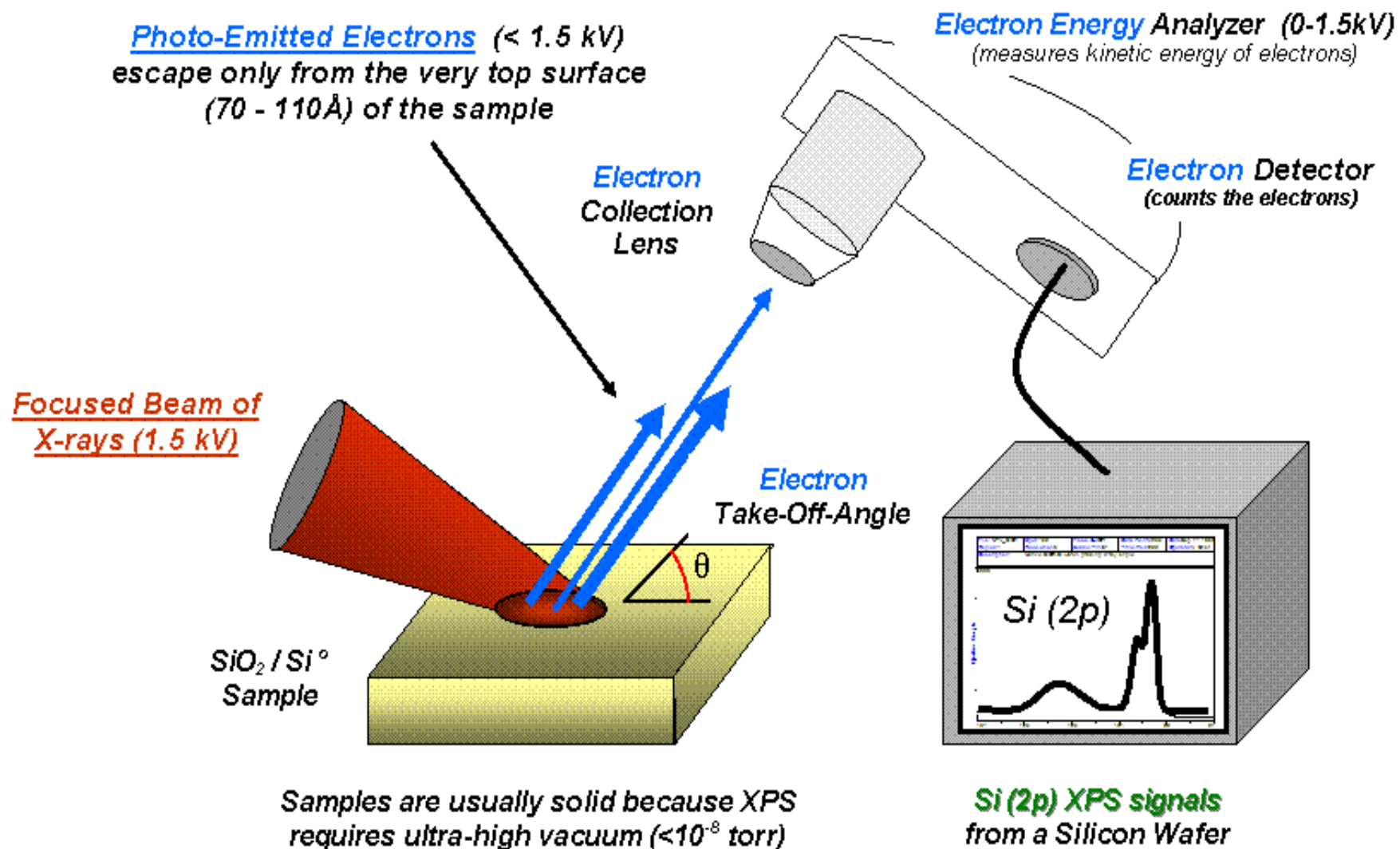
■ Based on Photoelectric Absorption

▶ **Core-hole Spectroscopies**

- X-ray Absorption Spectroscopy
- X-ray Emission Spectroscopy, RIXS, X-ray Raman
- X-ray Polarization Dependent Spectroscopies
- **X-ray Photoemission Spectroscopy**

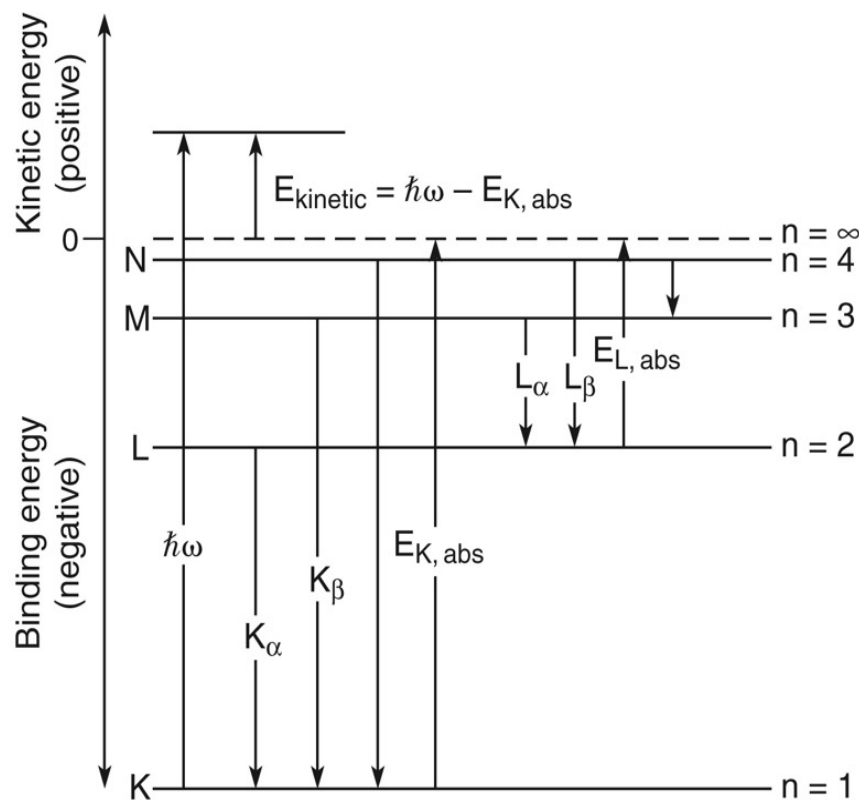
■ Imaging Methods

X-ray Photoelectron Spectroscopy

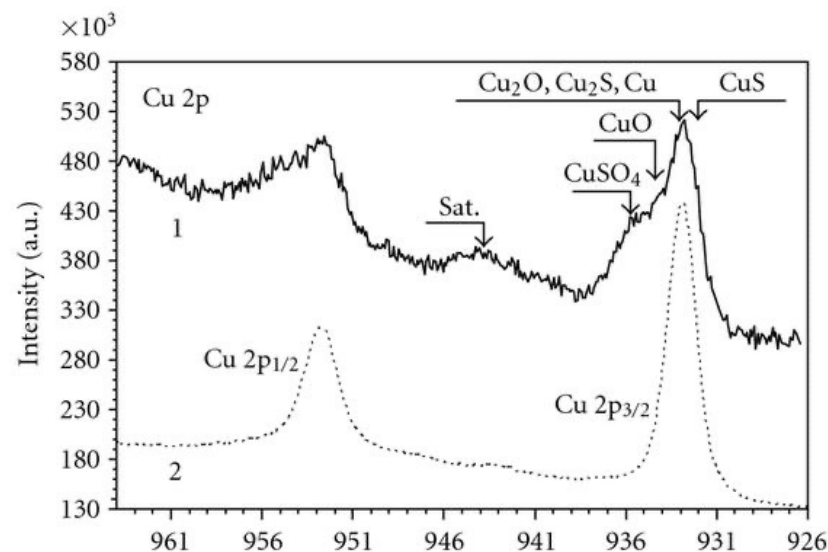


X-ray Photoelectron Spectroscopy for Chemical Analysis

- Binding energies of the measured electrons are characteristic of the chemical structure and molecular bonding of the material



Cu L₂ (2p_{1/2}) and L₃ (2p_{3/2}) photoemission



$$E_i = \hbar\omega - E_f - \phi$$

binding energy
of electron

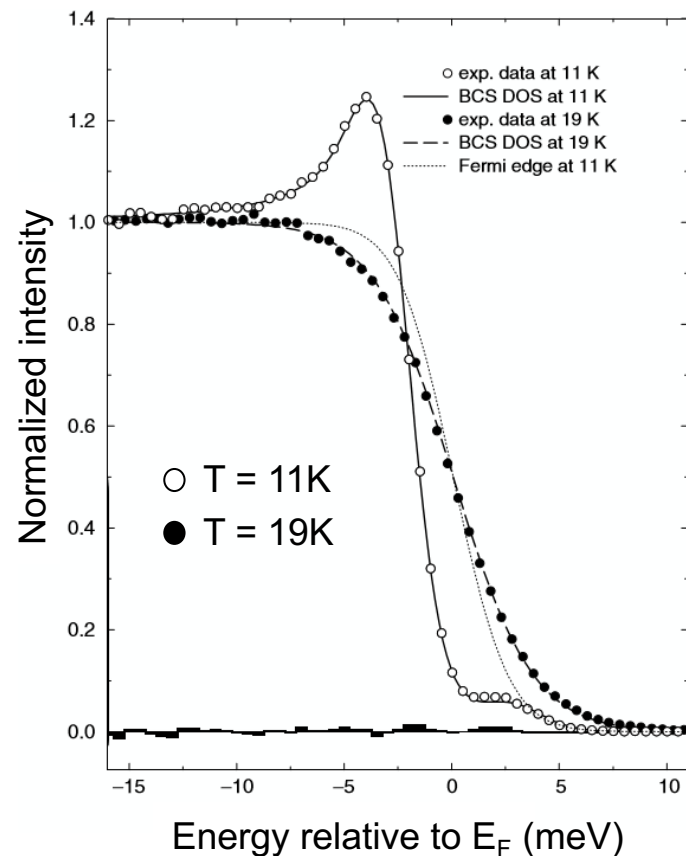
electron work function

incoming photon
energy → measured

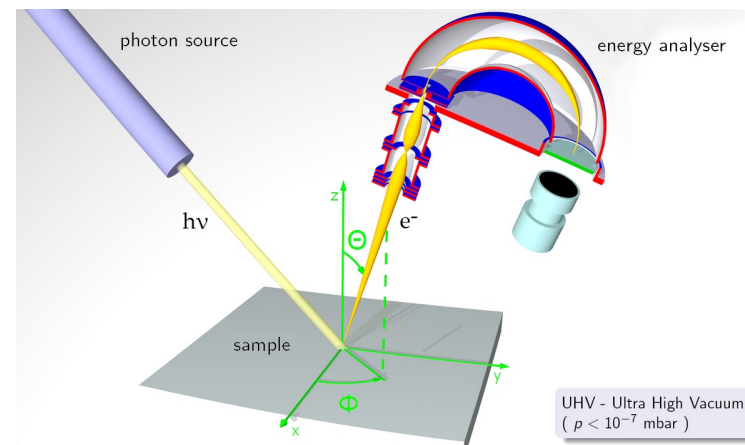
kinetic energy of outgoing
photoelectron → measured

Valence band photoemission & band structure determination

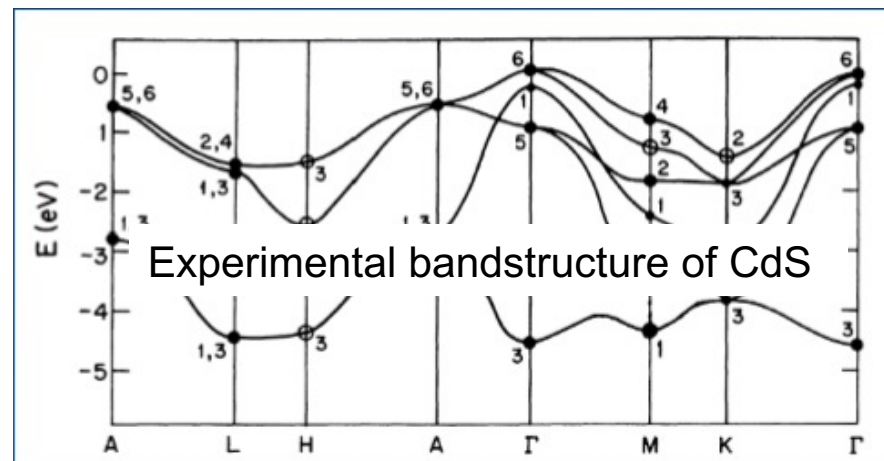
- Opening of the superconducting gap at $T < T_c$
- Less electron states at the Fermi level



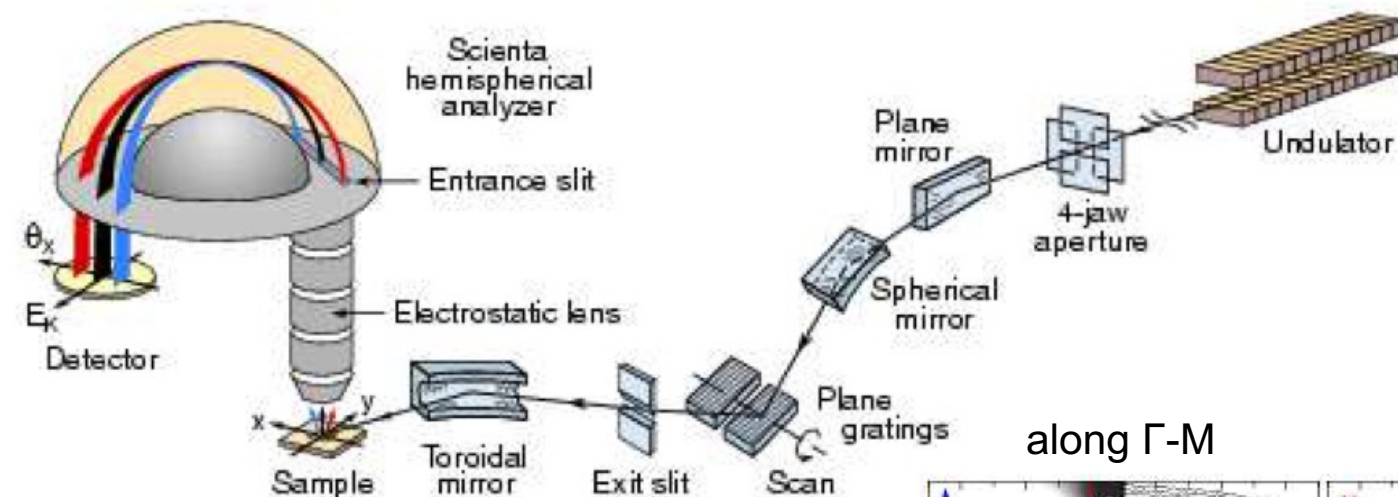
Angle-Resolved Photo-Electron Spectroscopy



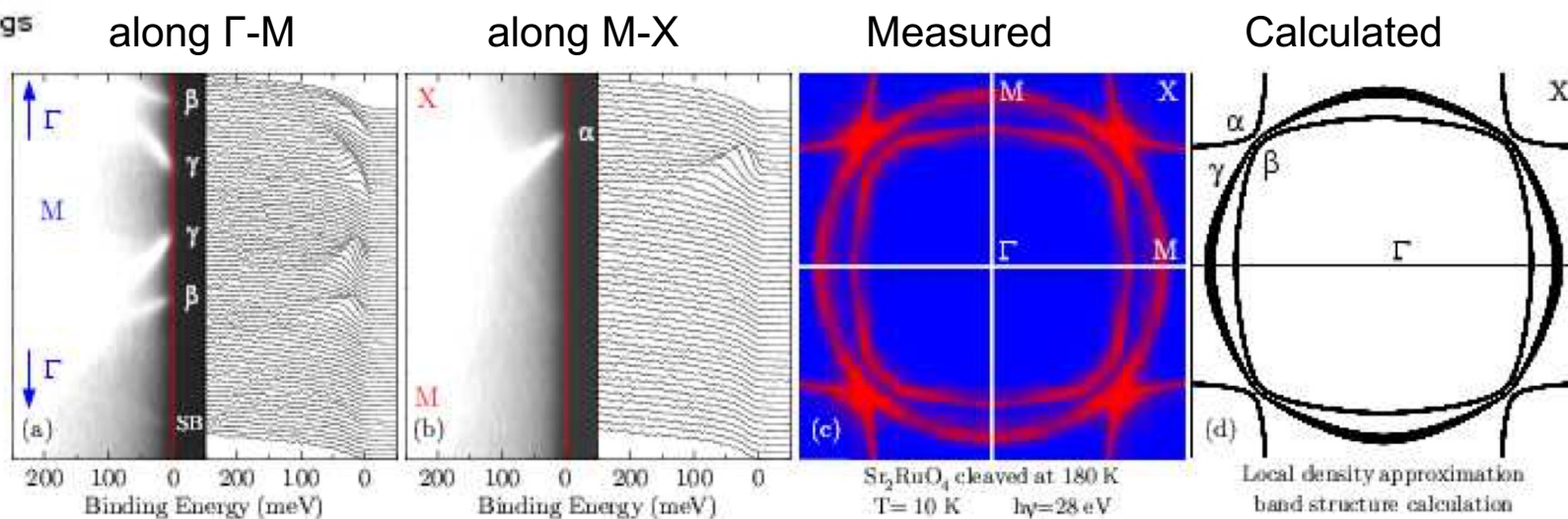
Distribution of electrons in reciprocal space



High resolution ARPES reveals Fermi surface in novel superconductors



- Novel superconductor Sr_2RuO_4
- Isostructural to the archetypal cuprate parent compound La_2CuO_4 (but RuO_2 planes replace the CuO_2 planes)
- Its low-energy electronic structure is characterized by three bands crossing the chemical potential



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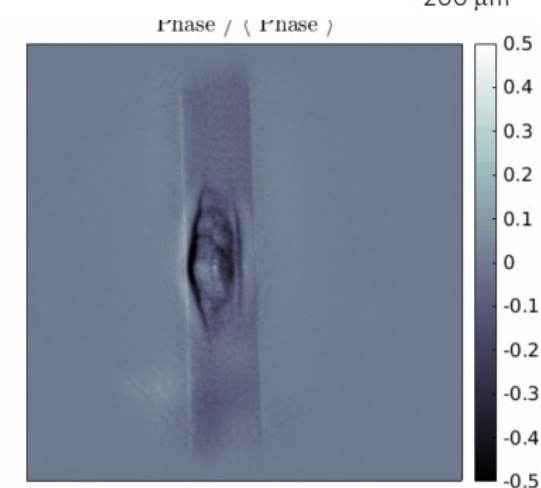
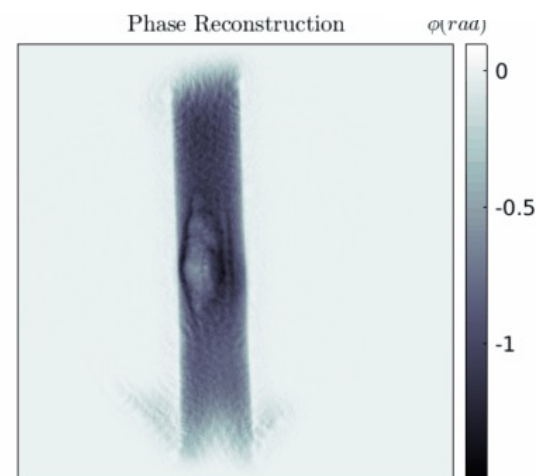
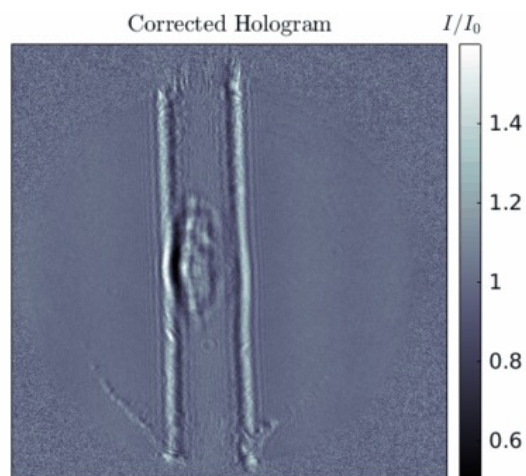
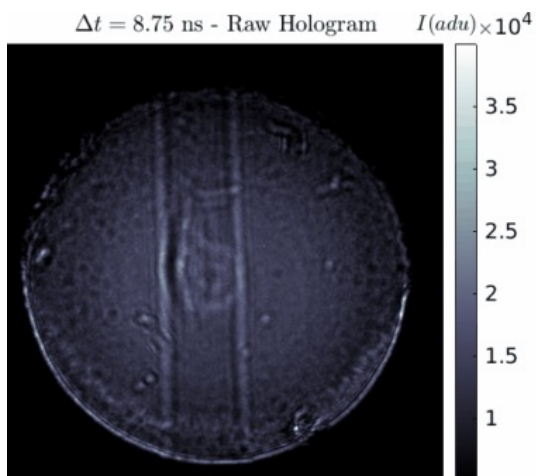
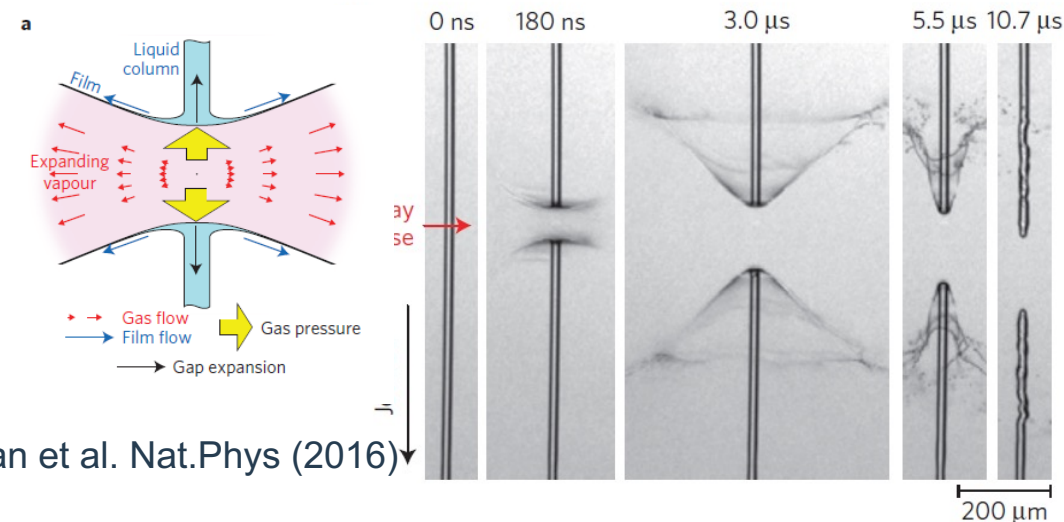
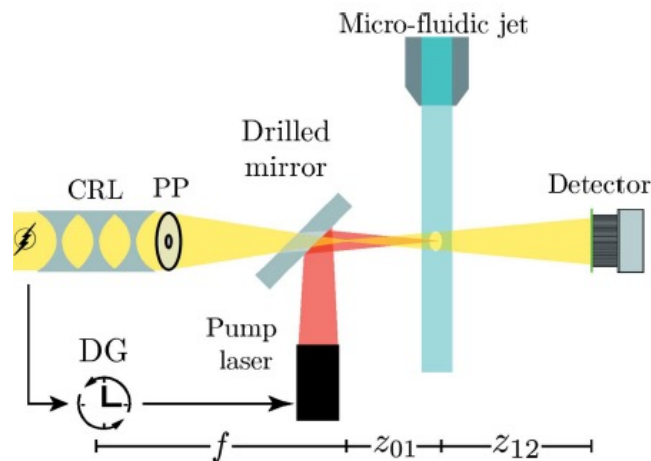
■ Based on Photoelectric Absorption

- ▶ Core-hole Spectroscopies
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 - X-ray Emission Spectroscopy, RIXS, X-ray Raman
 - X-ray Polarization Dependent Spectroscopies
 - X-ray Photoemission Spectroscopy

■ Imaging Methods

Density Imaging: Shock propagation by ultrafast holography

- Phase contrast imaging
- Very sensitive to tiny changes in refractive index
- Provides density profile





Strain Imaging: core-shell nanowires by nano-XRD

Towards high-efficiency solar cells, light-emitting diodes (LEDs), and nanoscale photodetectors

■ GaAs-GaInP core shell nanowire

■ Lab source XRD

■ few mm² spot

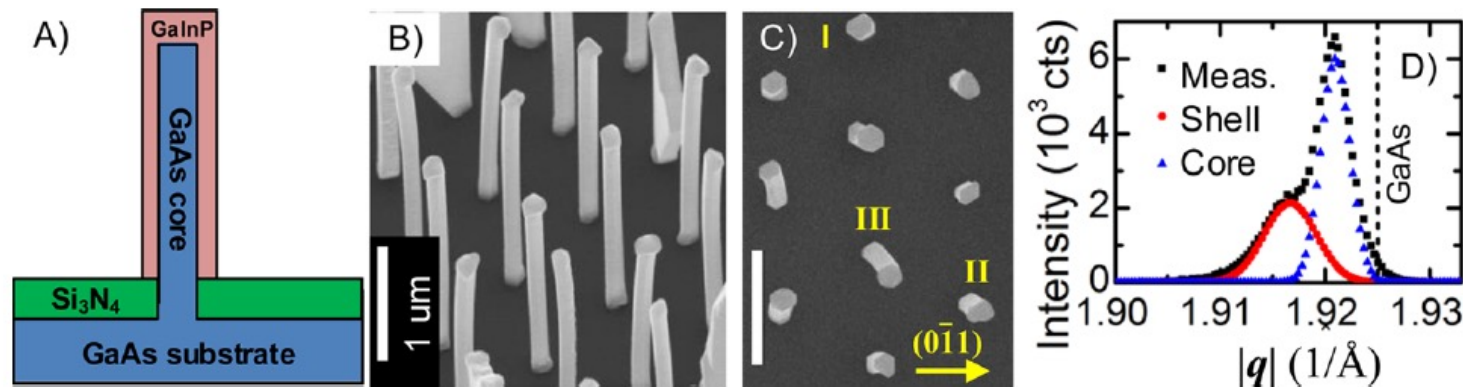
■ Measure (111) reflection – average over many wires

■ «Nano-XRD» @ ID11 ESRF

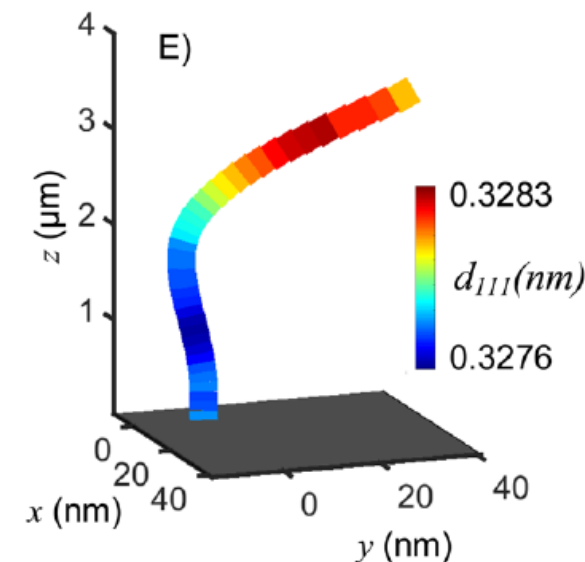
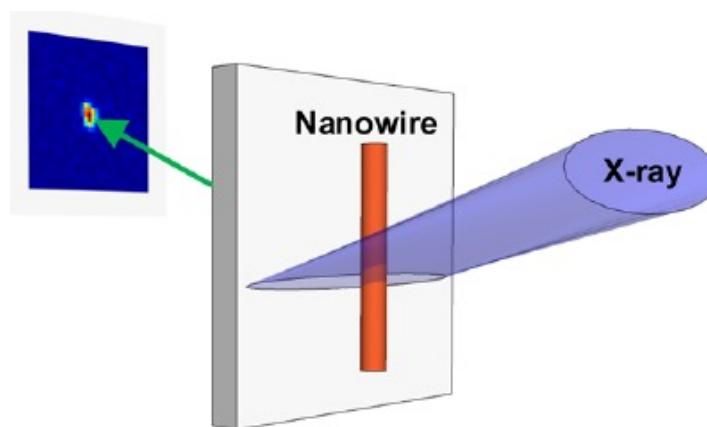
■ Spot size 0.1 × 6 mm

■ Measure individual wires

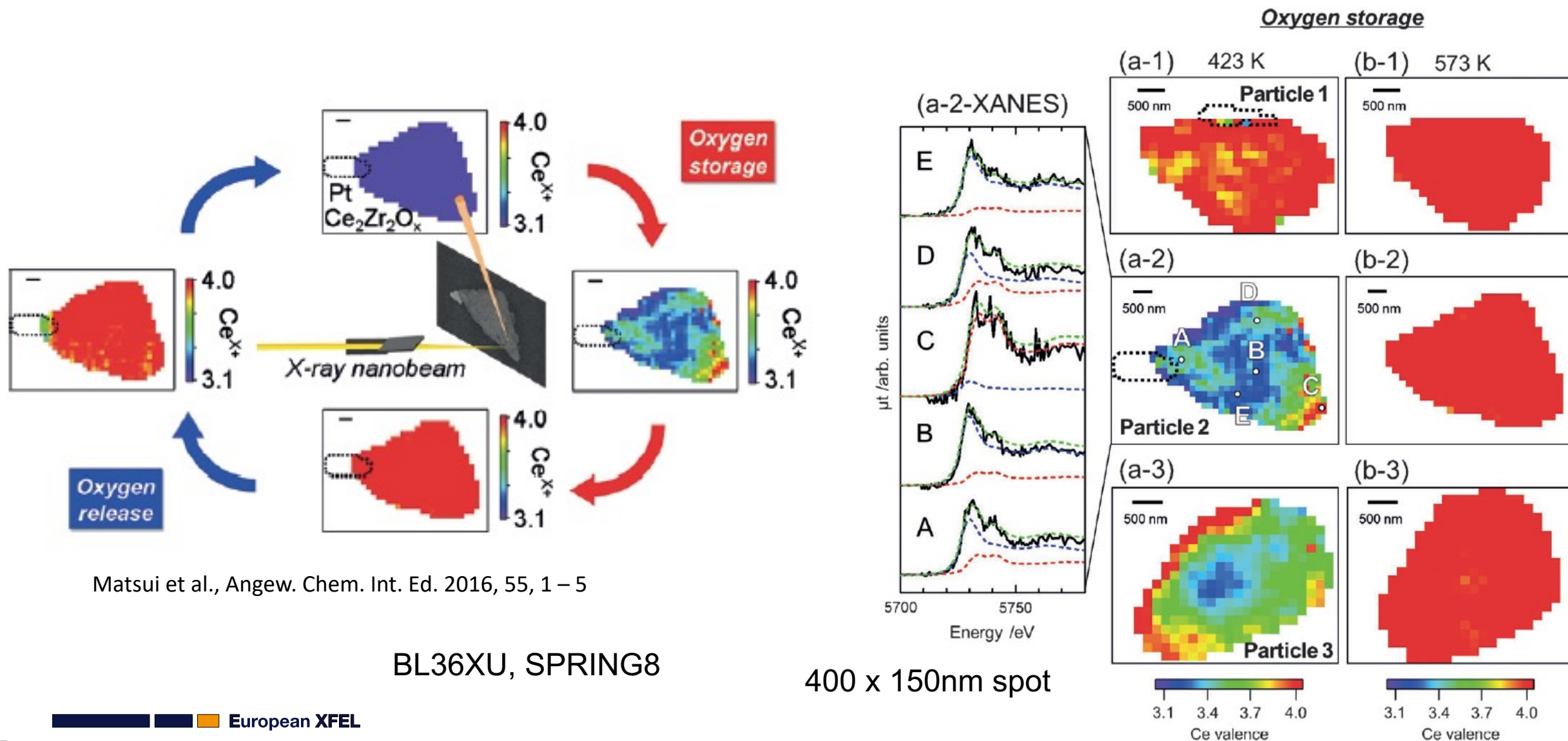
■ Spatial variation of (111) lattice spacing in a bent nanowire



SEM



Chemical Imaging: Oxygen diffusion in individual catalyst particle

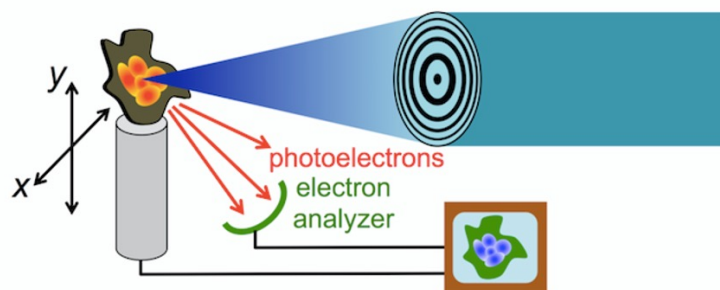


Matsui et al., Angew. Chem. Int. Ed. 2016, 55, 1 – 5

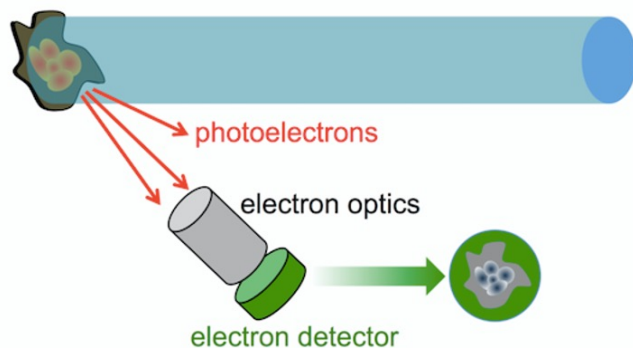
BL36XU, SPRING8

400 x 150 nm spot

Photoelectron spectro-microscopy for chemical contrast



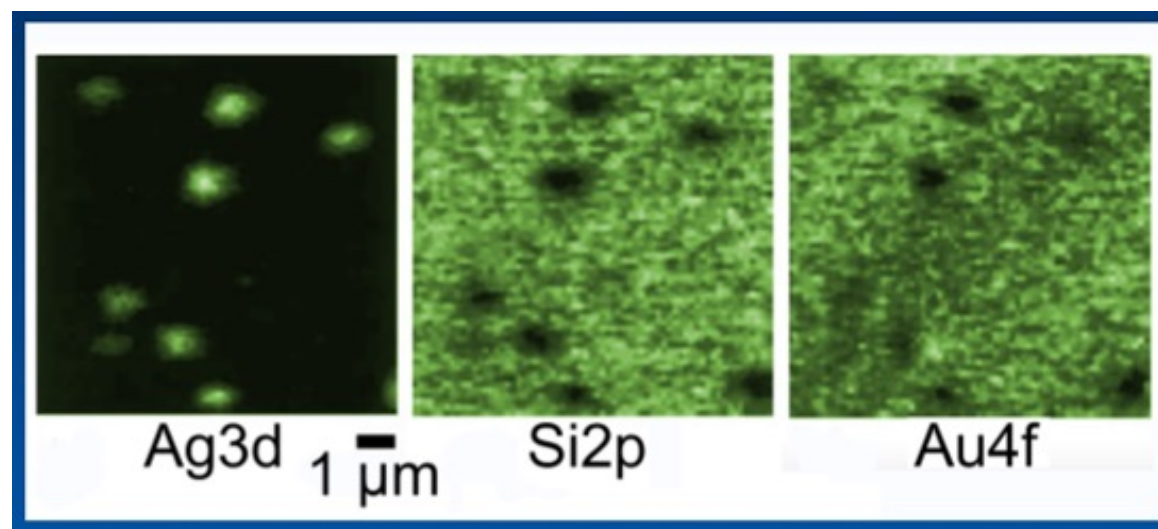
Scanning mode



Full field mode

Photoelectron images at different photoelectron energies

→ chemical contrast inversion



M. Marsi et al.; J. Elec. Spectros. 84, 73 (1997)

Thank you for your attention !

